# A stereochemically flexible approach to pyrrolidines based on 5-endo-trig iodocyclisations of homoallylic sulfonamides 

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5-endo-trig Iodocyclisations of the $(E)$-homoallylic sulfonamides 24 in the presence of potassium carbonate give excellent yields of trans-2,5-disubstituted-3-iodopyrrolidines 36. In the absence of base, these initial kinetic products undergo rapid isomerization to the corresponding cis-2,5-disubstituted pyrrolidines 37 by a ring opening-reclosure mechanism to these thermodynamic isomers. Octahydroindole derivatives 40 and 41 can be similarly obtained from the alk-2-enylcyclohexylsulfonamides $\mathbf{3 1}$ and 32. Attempted 5-endo-trig cyclisations of homoallylic carbamates were generally unsuccessful and resulted instead in relatively inefficient 6-exo cyclisations involving the carbamate carbonyl group. Deiodination provides both trans- and cis-2,5-disubstituted pyrrolidines [49 and 50]; the free amine derived from trans-2-butyl-5-pentylpyrrolidine 49d is a component of a fire ant defence pheromone. X-Ray crystallography was carried out for $\mathbf{3 7 f}, \mathbf{3 6}$, and $\mathbf{3 7 k}$.

One of the fundamental tenets of Baldwin's rules is that 5-exotrig cyclisations $\mathbf{1}$ are amongst the most favoured while 5-endotrig processes 2 are the least favoured (Fig. 1). ${ }^{1}$ However, it is now well established from results obtained by a number of groups, ${ }^{2}$ including ours, ${ }^{3}$ that 5 -endo-trig cyclisations of ( $E$ )-homoallylic alcohols $\mathbf{3}$ can be induced using both iodonium and phenylselenenium cations leading to excellent yields of tetrahydrofurans [e.g. 4] (Scheme 1). Such cyclisations are


Scheme 1
usually highly stereoselective, in contrast to similar 5-exo-trig cyclisations; ${ }^{4}(Z)$-homoallylic alcohols can similarly be converted into the all-cis-diastereoisomers 5, although in poorer

yields. ${ }^{3}$ These observations can be explained by the transition state conformation 6 wherein the selective generation of the 2,5-trans-iodotetrahydrofurans 4 is controlled by the pseudoequatorial positioning of the substituent $\mathrm{R}^{1}$. The necessary 'axial' position which has to be adopted by the alkene substituent $(Z)$ accounts for the poorer yields obtained from simple ( $Z$ )-homoallylic alcohols, although a suitably positioned hydroxy group results in much higher yields and stereo-


1


2

Fig. 1
selection. ${ }^{5}$ As electrophile-driven reactions, these cyclisations are not exceptions to Baldwin's rules; indeed, a nucleophilic centre, when positioned such that it can undergo a competing 5-exo cyclisation will react preferentially, except in exceptional circumstances. ${ }^{2,3,6}$ Further, the additional demands present in such 5-endo cyclisations seem to emphasise the requirement for the well-ordered and probably late transition state conformation $6,{ }^{7}$ of the lowest possible energy, in contrast to the less stereoselective 5-exo cyclisations. In the light of the synthetic utility of these reactions, we asked the perhaps obvious question: is it possible to carry out similar cyclisations with homoallylic amine derivatives 7 and thus define a new and equally useful approach to pyrrolidines $\mathbf{8}$ (Scheme 2)? Herein, we report




Scheme 2
in full that this is indeed possible and describe some of the surprises associated with this chemistry. ${ }^{8}$

## Results and discussion

Our first consideration was the nature of the nitrogen protecting group $\mathrm{R}^{3}$ (Scheme 2). That the amine group required
protection seemed plain from studies of the reactions between amines and molecular iodine which result in N -iodination, the formation of molecular complexes and, in some cases, unusual oxidation processes. ${ }^{9}$ Further, the stability of an iodopyrrolidine with no $N$-protection was an additional point of concern. It seemed unlikely that carbamates or $N$-acyl derivatives would be an optimum choice. The disfavoured nature of 5-endo-trig cyclisations suggested that such substrates as $\mathbf{9}$ would

preferentially undergo a 6-exo cyclisation (10) to give, at least initially, the heterocycles 11. Alternatively, loss of the carbamate alkyl group, certainly plausible if an $N$-Boc or Cbz group were to be used (i.e. $\mathrm{R}=\mathrm{Bu}^{\mathrm{t}}$ or $\mathrm{PhCH}_{2}$ ), would lead to the cyclic carbamates 12, also by a 6-exo process. In these considerations, we were guided by previous reports on the 5-exo-trig cyclisation of pent-4-enamine derivatives 13 , leading to pyrrolidines 14 , and related cyclisations of pent-4-enamides leading, when successful, to the corresponding 5-membered lactams. ${ }^{4}$ For example, conversion of pent-4-enamides into the corresponding $\mathrm{N}, \mathrm{O}$-bis(trimethylsilyl) derivatives ${ }^{10}$ or thioimidates ${ }^{11}$ is necessary for the nitrogen to act as the nucleophile, leading to lactams; similarly, direct iodocyclisation of unsaturated thioamides results in sulfur acting as the nucleophilic centre, leading to tetrahydrothiophene derivatives. ${ }^{12}$ However, both $N$-tosyl-3-hydroxypent-4-enamines ${ }^{13}$ and related unsaturated $N$-alkoxyamines ${ }^{14}$ undergo smooth iodocyclisations while various $N$-acyl derivatives are unreactive, ${ }^{13}$ perhaps because of an insufficiently low $\mathrm{p} K_{\mathrm{a}}$ value of the $\mathrm{N}-\mathrm{H}$ bond. ${ }^{15}$ Direct 5exo iodocyclisation of an amide is possible in the case of but-3-enyl- $\beta$-lactams to give carbapenems, not surprisingly, as the alternative cyclisation via oxygen would lead to a rather strained bridged species. ${ }^{16}$ Apparent examples of 5-endo amide cyclisations in a similar system ${ }^{17}$ or ones which lead to $\beta$-alkoxypyrrolidines ${ }^{18}$ may, in reality, proceed by a 5-exo process ${ }^{6}$ or by a more complex mechanism, respectively. Similarly, selenocyclisations of unmodified pent-4-enamides tend to result in reaction via oxygen rather than nitrogen to give tetrahydrofuran derivatives. ${ }^{19}$ However, in contrast, such 5-exo cyclisations of carbamates $\mathbf{1 3}$, derived from pent-4-enamines, generally work well to give N -alkoxycarbonylpyrrolidines. ${ }^{20}$ Although free alkenylamines fail to cyclise smoothly, orthoallylanilines do undergo selenocyclisation efficiently, as exemplified in an elegant approach to the mitomycin skeleton. ${ }^{21}$ Similarly, $N$-acetylpent-4-enamines $\left(13 ; \mathrm{R}^{3}=\mathrm{Me}\right)^{22}$ and the corresponding N -pentenylimidates ${ }^{23}$ cyclise in a 5-exo manner to give selenomethylpyrrolidines. An attempted 5-endo selenocyclisation of $N$-acetylbut-3-enamine resulted instead in reaction at oxygen by a 6-exo pathway (cf. 11). Mercury(II)-induced

5-exo cyclisations seem to be more flexible, as these are possible with both $N$-alkylpent-4-enamines ${ }^{24,25}$ and the corresponding carbamates. ${ }^{26}$ By careful choice of conditions and reagents, these can be highly stereoselective, giving the 2,5-trans diastereoisomers as the kinetic products and the corresponding $2,5-$ cis isomers under thermodynamic conditions. Interestingly, an isolated example of a 5-endo cyclisation has been reported: treatment of $N$-propylbut-3-enamine with mercury(II) chloride in tetrahydrofuran gives 3-chloromercurio-1-propylpyrrolidine. ${ }^{24}$ A similar mechanism probably operates in the generation of bis-mercuriopyrrolidines by the addition of tosylamine and mercury(II) nitrate to 1,4 -dienes leading to $\beta$-mercuriopyrrolidines. ${ }^{27}$ Direct 5-exo sulfenocyclisation of $N$-benzoyl-pent-4-enamines $\left(\mathbf{1 3} ; \mathrm{R}^{3}=\mathrm{Ph}\right)^{28}$ and of N -tosyl-2-allylanilines ${ }^{29}$ is also viable, given that an activated sulfenium species is employed (Scheme 3). Alternatively, the necessary episulfonium

species can be generated by a prior mercury(iI)-induced cyclisation ${ }^{30}$ or by the addition of benzenesulfenyl chloride to an N -alkylbut-3-enamine followed by intramolecular displacement of the halogen by sulfur. ${ }^{31}$ In this latter example, the overall reaction is a 5-endo process. There are also many examples of palladium-induced 5-exo cyclisations of these types of amine derivatives $(\mathbf{9}, \mathbf{1 3})$; an advantage of this methodology is the range of additional substituents which can be incorporated by further reactions of the intermediate palladium species. ${ }^{32}$

In view of the chemistry outlined above, we chose to use the sulfonamide group for protection of the nitrogen function in our initial studies of the prospects for effecting 5-endo-trig iodocyclisations of homoallylic amine derivatives, since we expected that it would not participate directly in such reactions. More positively, this functionality had previously been found to be particularly reactive in electrophile-induced cyclisations, presumably because of its ability to lower the $\mathrm{p} K_{\mathrm{a}}$ of the $\mathrm{N}-\mathrm{H}$ bond. ${ }^{15,33}$ Thus, we required a series of sulfonamides 15 . As we

had available proven methodology for the formation of the corresponding homoallylic alcohols $\mathbf{1 6},{ }^{3,7}$ an efficient approach to such substrates appeared to be by Mitsunobu reaction ${ }^{34}$ between these and the sulfonamide derivatives 17, as developed by Weinreb and colleagues. ${ }^{35}$ As we reported previously, ${ }^{3,7}$ the alcohols 16 were readily prepared by sequential YamaguchiHirao condensation ${ }^{36}$ between monosubstituted epoxides $\mathbf{1 8}$ and acetylenes 19 followed by reduction of the resulting alcohols 20 using various methods to give either the $(Z)$ - or (E)-homoallylic alcohols 21 or 22 (Scheme 4). In our hands, the Weinreb reagents $\mathbf{1 7}$ were easily prepared as previously reported and underwent smooth Mitsunobu reactions with the foregoing


18 - 19 i) $\mathrm{BuLi} \mid$ ii) $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$


20


21


22
Scheme 4


23
a) $\mathrm{R}^{1}=\mathrm{Me} ; \mathrm{R}^{2}=\mathrm{H} ; \mathrm{R}^{3}=\mathrm{Ph}$;
b) $\mathrm{R}^{1}=\mathrm{H} ; \mathrm{R}^{2}=\mathrm{Et} ; \mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$;
c) $\mathrm{R}^{1}=\mathrm{H} ; \mathrm{R}^{2}=\mathrm{Et} ; \mathrm{R}^{3}=\mathrm{Ph}$;
d) $\mathrm{R}^{1}=\mathrm{Me} ; \mathrm{R}^{2}=\mathrm{Bu} ; \mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$;
e) as d) but ( $R$ )-enantiomer;
f) $\mathrm{R}^{1}=\mathrm{Et} ; \mathrm{R}^{2}=\mathrm{Bu} ; \mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$;
g) $R^{1}=E t ; R^{2}=B u ; R^{3}=M e ;$
h) $\mathrm{R}^{1}=\mathrm{Et} ; \mathrm{R}^{2}=$ Pent; $\mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$;
i) $\mathrm{R}^{1}=\mathrm{Bu} ; \mathrm{R}^{2}=$ Pent; $\mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$;
j) $\mathrm{R}^{1}=\mathrm{Et} ; \mathrm{R}^{2}=\mathrm{Ph} ; \mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$;
k) $R^{1}=E t ; R^{2}=P h ; R^{3}=P h$.



26
a) $\mathrm{R}^{1}=\mathrm{H} ; \mathrm{R}^{2}=\mathrm{Et} ; \mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$;
b) $\mathrm{R}^{1}=\mathrm{H} ; \mathrm{R}^{2}=\mathrm{Et} ; \mathrm{R}^{3}=\mathrm{Me}$;
c) $\mathrm{R}^{1}=\mathrm{Et} ; \mathrm{R}^{2}=\mathrm{Bu} ; \mathrm{R}^{3}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$.



28
a) $\mathrm{R}=\mathrm{Boc}$;
b) $R=H$.
$\left[\mathrm{Ts}=p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{SO}_{2}\right]$
alcohols to give the expected amine derivatives $\mathbf{2 3}$ and $\mathbf{2 5}$ in good isolated yields. These were then deprotected using trifluoroacetic acid to give a representative series of precursors 24
and 26. The more highly substituted analogues 27b and 28b were prepared similarly starting from the corresponding epoxybutanes. Condensation between epoxycyclohexane and hex-1-yne provided the cyclohexanol $29 ;{ }^{7}$ subsequent Mitsunobu reaction ${ }^{34}$ but using phthalimide as the nucleophile gave the expected product 30a in poor but sufficient yield (Scheme 5). When the $N$-Boc-sulfonamides 17 were used in place of phthalimide, only elimination products were isolated; various azide displacements were similarly unproductive. ${ }^{37}$ Protecting group exchange and reductions then gave the required $(E)$ - and $(Z)$-isomers 31 and 32. The preparation of a range of homoallylic amines protected by carbonyl-based groups was achieved by a different approach (Scheme 6): enolate Claisen rearrangement ${ }^{38}$ of the ester $\mathbf{3 3}$ led smoothly to the unsaturated acid 34, a modified Curtius rearrangement of which using diphenyl phosphorazidate ${ }^{39}$ gave the Boc-protected unsaturated amine 35a. Deprotection gave the corresponding amine $\mathbf{3 5 b}$; standard acylation methods then gave a series of derivatives $\mathbf{3 5 c} \mathbf{c}$.

In our initial studies, we naturally chose to use the conditions which had been so successful in effecting iodocyclisations of homoallylic alcohols, i.e. treatment of the appropriate substrate with three equivalents each of iodine and sodium hydrogen carbonate in dry acetonitrile at ambient temperature. Our first experiment proved the viability of the idea, using the sulfonamide $\mathbf{2 4 a}$ having no distal substituent. We were pleased to find that cyclisation did occur, although the reaction gave a number of other products and almost no stereoselection (Table 1). Encouraged by this, we next found that the sulfonamide $\mathbf{2 4 b}$ underwent smooth and rapid cyclisation to give only the trans-2,3-iodopyrrolidine 36b. Similarly the phenylsulfonyl derivative $\mathbf{2 4 c}$ gave only the iodopyrrolidine 36c in good yield, a stereoselection which reflects the expected anti-addition across the double bond. However, there was distinctly less stereoselection when we attempted to form trisubstituted pyrrolidines under these conditions. Thus, the precursors $\mathbf{2 4 f}$ and $\mathbf{2 4 j}$, while undergoing encouragingly efficient cyclisations, gave synthetically uninteresting mixtures of the products $\mathbf{3 6 f}-\mathbf{3 7 f}$ and $\mathbf{3 6 j}-\mathbf{3 7} \mathbf{j}$, in marked contrast to similar cyclisations leading to tetrahydrofurans. ${ }^{3}$ Various changes to the conditions including lowering the temperature to $0{ }^{\circ} \mathrm{C}$ and even $-78^{\circ} \mathrm{C}$ and changing the solvent to ether or dichloromethane did not result in any improvement. Fortunately, we observed that the isomer ratio in the products $\mathbf{3 6 f}-\mathbf{3 7 f}$ derived from sulfonamide $\mathbf{2 4 f}$ varied significantly, depending upon the exact method of work up. We wondered if this could be due to the presence of variable amounts of acid arising from the cyclisation itself causing an isomerisation and reasoned that a stronger base might improve matters. To our satisfaction, changing the base from sodium hydrogen carbonate to the slightly stronger potassium carbonate resulted in the almost exclusive formation of the iodopyrrolidine $\mathbf{3 6}$ from the sulfonamide $\mathbf{2 4 f}$ under conditions otherwise identical to those used initially. Subsequently, this was used as the base throughout and, as summarized in Table 1, all such reactions were highly selective in favour of the 2,5-trans-isomers 36. A surprise associated with this particular recipe was the effect of water. In our previous work on the tetrahydrofuran synthesis, we had deduced that the presence of water was especially deleterious to cyclisation, due to its competing by attacking the iodonium ion intermediate to give the corresponding iodohydrins. However, we discovered that, in marked contrast to this, in the present sulfonamide cyclisations, when potassium carbonate was used as the base, the addition of small amounts of water not only improved the yields but also the stereoselectivity, in favour of isomers 36. We assume that this is because, under these conditions, the hydrogen iodide produced is more quickly neutralised by the base and hence cannot cause isomerization of these initial products (see below) and that water did not compete as a nucleophile to give iodohydrins.



Scheme 6
These results suggested that, in the absence of base, it might be possible to completely isomerize the presumed initial products $\mathbf{3 6}$ to the isomers $\mathbf{3 7}$. We were delighted to find that when sulfonamide $\mathbf{2 4 f}$ was treated with three equivalents of iodine in acetonitrile, an extremely rapid cyclisation occurred from which was isolated only isomer $\mathbf{3 7 f}$ in excellent yield. The same result was obtained with all of the other arylsulfonamides, as detailed in Table 1. For reasons which are unclear, it was fortunate that we chose this particular protecting group. As shown in Table 1, the corresponding methanesulfonamide $\mathbf{2 4 g}$ failed to undergo isomerization but instead gave the same
product $\mathbf{3 6 g}$, no matter whether base was present or not. The more highly substituted isomers $\mathbf{2 7 b}$ and $\mathbf{2 8 b}$ also underwent smooth cyclisation to provide good yields of the iodopyrrolidines 38 and 39 , essentially as single isomers, whether or not any base was present. Finally, both the $(E)$ - and $(Z)$-alkenyl substituted cyclohexane derivatives 31 and 32 underwent cyclisation simply by the addition of iodine (i.e. no base was used) to give the perhydroindole derivatives $\mathbf{4 0}$ and 41 , in much the same way as the corresponding octahydrobenzofurans can be formed from alk-2-enylcycloalkanols. ${ }^{7}$ However, these last two cyclisations are no more than proof of concept as the yields attained during preparation of the precursors were so poor.

We then turned to an examination of similar iodocyclisations of the related $(Z)$-sulfonamides $\mathbf{2 6}$; the results are summarized in Table 2. As is evident from this Table, such cyclisations were not as successful under any of the conditions tried, both in terms of yields and stereoselection. The products were identified from NMR spectra of the mixtures obtained, by comparisons with pure isomers obtained as indicated in Table 1. These results are very much in line with those obtained from cyclisations of the related $(Z)$-homoallylic alcohols (see below). ${ }^{3}$ The results from cyclisations of the trisubstituted pyrrolidine precursor 26c were unexpected. Conformation 6 suggests that the initial product should be the all-cis isomer 44; however, less of this was isolated under basic conditions than under acidic

Table 1 Iodocyclisations of ( $E$ )-homoallylic sulfonamides 24


| Substrate | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ | Time/min | Additive | Yield (\%) | Ratio 36 : 37 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 24a | Me | H | Ph | 60 | $\mathrm{NaHCO}_{3}$ | 42 | 60: 40 |
| 24b | H | Et | p-Tol | 10 | $\mathrm{NaHCO}_{3}$ | 84 | 100:0 |
|  |  |  |  | 20 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | 73 | 100: 0 |
|  |  |  |  | 2 | No base | 88 | 100:0 |
| 24c |  | Et |  | 10 | $\mathrm{NaHCO}_{3}$ | 84 | 100: 0 |
| 24d | Me | Bu | p-Tol | 25 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | 85 | 93:7 |
|  |  |  |  | 3 | No base | 86 | 0: 100 |
| 24e [( $R$ )-24d] |  |  |  | 3 | No base | 85 | 0: 100 |
| 24f | Et | Bu | p-Tol | 25 | $\mathrm{NaHCO}_{3}$ | 76 | $74: 26$ |
|  |  |  |  | 30 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | 74 | 94:6 |
|  |  |  |  | 3 | No base | 78 | 0: 100 |
| 24g | Et | Bu | Me | 25 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | 72 | 97:3 |
|  |  |  |  | 3 | No base | 78 | 94:6 |
| 24h | Et | Pent | p-Tol | 25 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | 79 | 97:3 |
|  |  |  |  | 3 | No base | 74 | 0: 100 |
| 24i | Bu | Pent | p-Tol | 35 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | 78 | $94: 6$ |
|  |  |  |  | 15 | No base | 76 | 0:100 |
| 24j | Et | Ph | p-Tol | 40 | $\mathrm{NaHCO}_{3}$ | 76 | $65: 35$ |
|  |  |  |  | 45 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | 83 | 94: 6 |
|  |  |  |  | 10 | No base | 76 | 0: 100 |
| 24k | Et | Ph | Ph | 10 | No base | 81 | $0: 100$ |

conditions, when isomerization was expected, suggesting that perhaps an alternative conformation is involved.

The final group of substrates examined were the series of ( $E$ )- $N$-acyl derivatives 35 , which we expected to undergo competing 6 -exo cyclisation to at least some degree, if not completely (see above; structures 9 to 12). In the event (Table 3), only small amounts ( $10 \%$ ) of the desired pyrrolidines 47 and 48 were formed in all cases. These were identified by comparisons with foregoing compounds (Table 1). The major products were tentatively identified as those from 6 -exo cyclisations. In the case of the $N$-Boc derivative 35a, loss of the tert-butyl group was evident; other NMR data on the $2: 1$ mixture of isomers obtained were consistent with the cyclic carbamate structures 45, epimeric at the methyl substituent, given the expected antiaddition across the double bond. Although this cyclisation is reasonably efficient, the relative lack of stereoselection rather detracts from any synthetic utility. Not surprisingly, under acidic conditions, some of the starting amine $\mathbf{3 5 b}$ was also obtained. Similar cyclisations of the acetyl and trifluoroacetyl derivatives $\mathbf{3 5}$ c and $35 d$ gave the alternative 6 -exo products 46c,d, as unstable mixtures of isomers which again were not fully characterized. This latter mode of cyclisation also appeared to occur with the methoxycarbonyl derivative 35e, presumably because loss of a methyl carbocation is much less favourable than loss of tert-butyl. Overall, most of these


cyclisations gave mixtures of a number of compounds and are certainly not synthetically useful, at least under the present sets of conditions.

Finally, we briefly examined some reactions of the initial products 36 and 37 . Firstly, deiodination using tributyltin hydride led smoothly to the expected 2,5 -disubstituted pyrrolidines 49 and 50 in good yields. In no cases was any isomerization observed. A number of such 2,5-dialkylpyrrolidines, with combinations of $\mathrm{C}_{2}, \mathrm{C}_{4}, \mathrm{C}_{6}$ and $\mathrm{C}_{7}$ side chains, have been identified as components in the defence secretions of various fire ants. ${ }^{40}$ A specific example is the free amine derived from the trans-5-butyl-2-pentylpyrrolidine 49d which has previously been synthesized by low yielding and non-stereoselective approaches, in which a final step is removal of the $N$-tolylsulfonyl group using lithium aluminium hydride. ${ }^{41}$ This present chemistry therefore could be used to selectively prepare both possible diastereoisomers of such structures and to obtain optically pure samples. It also proved possible to effect baseinduced $\mathrm{E}_{2}$ eliminations of the initial products to provide access to the corresponding 3-pyrrolines 51 and 52; again, no isomerization occurred in the two examples examined.

Throughout these studies, the structures of the various isomeric pyrrolidines were determined in much the same way as in our work on tetrahydrofuran synthesis. ${ }^{3}$ Firstly, CHI functions were identified in ${ }^{13} \mathrm{C}$ spectra from their unique position around $20-30 \mathrm{ppm}$, due to the heavy atom effect. ${ }^{42}$ Subsequent ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ and ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ correlation experiments then allowed full assignment of each resonance and crucially confirmed that the products were indeed pyrrolidines and not azetidines which could arise from 4-exo-trig cyclisations. As is typical of fivemembered rings, coupling constant data were not helpful in determining stereochemistry. In some examples (36f, 36j), the problem of overlapping resonances was solved by running the ${ }^{1} \mathrm{H}$ NMR spectra in solvents other than deuteriochloroform. However, NOE measurements gave reasonable indications of the assigned stereochemistries, but were certainly not definitive, as was the case with the related tetrahydrofurans. ${ }^{3}$ Definite structural proof was finally achieved by pertinent literature comparisons, especially of ${ }^{1} \mathrm{H}$ NMR data ${ }^{43}$ and, in particular, by single crystal X-ray determinations of the cis-2,5-dialkyl-3-iodopyrrolidine $\mathbf{3 7 f}$ together with the isomeric 2-phenylpyrrolidines $\mathbf{3 6 j}$ and $\mathbf{3 7 k} .^{44}$ Although many of the iodopyrrolidines were obtained as solids, these did not provide crystals suitable for X-ray analysis from the initial recrystallisations. Suitable samples were eventually obtained by a vapour


Table 2 Iodocyclisations of ( $Z$ )-homoallylic sulfonamides 26

|  |  |  | R | Time/min | Base | Yield (\%) | Product ratio |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 26a,b |  <br> 42 |  <br> 43 | $\begin{aligned} & \mathrm{R}=\mathrm{Ts} \\ & \mathrm{R}=\mathrm{Me} \end{aligned}$ | $\begin{aligned} & 60 \\ & 60 \\ & 60 \\ & 60 \\ & 24 \mathrm{~h} \\ & 60 \end{aligned}$ | $\mathrm{NaHCO}_{3}$ <br> $\mathrm{K}_{2} \mathrm{CO}_{3}$ <br> No base <br> $\mathrm{NaHCO}_{3}$ <br> $\mathrm{K}_{2} \mathrm{CO}_{3}$ <br> No base | $\begin{gathered} 40 \\ 52 \\ 35 \\ 45 \\ \sim 0 \\ 40 \end{gathered}$ | $\begin{aligned} & \text { 42a }: \mathbf{4 3 a}[\mathbf{3 6 b}]=35: 65 \\ & \text { 42a }: \mathbf{4 3 a}[\mathbf{3 6 b}]=38: 62 \\ & \text { Gross mixture } \\ & \text { 42b }: \mathbf{4 3 b}=50: 50 \end{aligned}$ <br> Gross mixture |
| 26c |  | + $36 \mathrm{f}+37 \mathrm{f}$ |  | 60 60 60 | $\mathrm{NaHCO}_{3}$ $\mathrm{K}_{2} \mathrm{CO}_{3}$ <br> No base | 65 63 57 | 44 : 36f : 37f = <br> 43: 30: 27 <br> 44: 36f : 37f = <br> 27: 83 :~0 <br> 44: 36f : 37f = <br> 66:19:15 |

Table 3 Iodocyclisations of homoallylic carbamates and amides 35


diffusion method in which a concentrated ethereal solution of the particular iodopyrrolidine was stored for up to one month in an atmosphere of hexane at ambient temperature. The crystal structure determinations of the iodopyrrolidines 37f and $\mathbf{3 6 j}$ proceeded uneventfully and confirmed the respective 2,5-cis and 2,5-trans stereochemical assignments made on the basis of NMR data; these are shown in Figs. 2 and 3. However, the vapour diffusion crystallisation of iodopyrrolidine $\mathbf{3 7 k}$ provided an unexpected complication: on close inspection of the sample, two crystalline forms were apparent, one as orange clusters (mp 103-104 ${ }^{\circ} \mathrm{C}$ ) while the other, minor, component (mp $95-97^{\circ} \mathrm{C}$ ) appeared as colourless rectangles. Fearing either decomposition of the sample or, worse, an unexpected isomerisation, individual samples were separated by hand and both were subjected to X-ray analysis. We were surprised and delighted to find that both confirmed the assigned 2,5-cis stereochemistry of compound $\mathbf{3 7 k}$ and that the two crystal forms were in fact differing conformations of the same diastereoisomer. The major and minor structures observed, $\mathbf{3 7 k}(\mathbf{i})$ and $\mathbf{3 7 k}$ (ii), are presented in Figs. 4 and 5 respectively. In the major conformation, the phenyl and iodine substituents appear in pseudoequatorial positions with the ethyl group positioned


Fig. 2 (2RS,3SR,5SR)-2-Butyl-5-ethyl-3-iodo-1-p-tolylsulfonylpyrrolidine 37 f .
pseudoaxially and the arylsulfonyl group trans to both $\alpha$-substituents. This more stable conformation contrasts with the minor diastereoisomer (Fig. 5) in which the iodine appears to be pseudoaxial. Subsequently, we observed, as expected, that samples of the minor conformation $\mathbf{3 7 k}(i i)$ exhibited spectroscopic data (infra-red, ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR) in solution which were identical to those displayed by the bulk sample of compound 37k. Despite this amusing complication, comparisons of the spectral data displayed by these three compounds with all the other products then completed the structural assignments.
It therefore seemed most likely that the trans-2,5disubstituted pyrrolidines $\mathbf{3 6}$ obtained from cyclisations under basic conditions were the kinetic products and that these were isomerized under acidic conditions to the thermodynamically more stable cis-2,5-disubstituted pyrrolidines $37 .{ }^{45}$ There was, however, a final structural problem associated with this idea: the trans-disposition of the 2 -substituent and the 3-iodo group in the 2,5 -cis-stereoisomers 37 strongly suggested that isomerization of the initial trans-isomers $\mathbf{3 6}$ was occurring by ring opening and reclosure. However, it was possible that, in some way, the 5 -centre in isomers $\mathbf{3 6}$ was undergoing epimerization to give the more stable $c i s$-isomers 37 . As the two possible structures are enantiomeric [i.e. 37 or $\mathbf{3 6}$ with $\mathrm{R}^{1}$ 'down' or $\alpha$-], we were unable to distinguish these two pathways using the foregoing


Fig. 3 (2RS,3SR,5RS)-5-Ethyl-3-iodo-2-phenyl-1-p-tolylsulfonylpyrrolidine $\mathbf{3 6 j}$.


Fig. $4(2 R S, 3 S R, 5 S R)$-5-Ethyl-3-iodo-2-phenyl-1-phenylsulfonylpyrrolidine $\mathbf{3 7 k}(\mathbf{i})$.


Fig. 5 (2RS,3SR,5SR)-5-Ethyl-3-iodo-2-phenyl-1-phenylsulfonylpyrrolidine $\mathbf{3 7 k}(\mathrm{ii})$.
racemates. We therefore prepared the enantiopure homoallylic sulfonamide 24e $[(R)$-24d], starting from $(S)$-epoxypropane, and subjected this to cyclisation under acidic conditions and isolated the expected iodopyrrolidine 37e. Cycloreversion by deiodination using zinc metal in acetic acid then provided the starting sulfonamide $\mathbf{2 4 e}$ with essentially the same optical rotation (Scheme 7). As confirmation of this, we found that selected examples of the trans-stereoisomers $\mathbf{3 6 d}, \mathbf{f}, \mathbf{j}$ underwent isomerization to the corresponding cis-isomers $\mathbf{3 7 d}, \mathbf{f}, \mathbf{j}$ when exposed to hydrogen iodide; other acids such as sulfuric acid and trifluoroacetic acid did not cause isomerization although exposure to 6 M hydrochloric acid resulted in partial isomeriz-


Scheme 8
ation (Scheme 8). Hence, we conclude that isomerization does indeed occur by cycloreversion and recyclisation and not by epimerization at the 5-position.

The formation of the trans-pyrrolidines 36 as kinetic products can be understood by the chair-like conformation 53, in which the substituent R is positioned equatorially, along much the same lines as that proposed (6) for the related tetrahydrofuran synthesis; the pseudoaxial position of the substituent ' $Z$ ' in such a conformation presumably accounts for the poor returns from cyclisations of the $(Z)$-homoallylic sulfonamides 26. The tetrasubstituted iodopyrrolidines 38 and 39


Scheme 9
were resistant to isomerisation. This was not surprising, as a similar conformation 54 would lead to the cis-isomer $\mathbf{3 8}$ from the anti-sulfonamide 27b which would not gain from such an isomerisation to a less thermodynamically stable 2,5-transpyrrolidine. The syn-isomer 28b presumably undergoes cyclisation via conformation 55 in which all substituents are effectively equatorial to give a product 39 in which each substituent has a trans relationship to the adjacent groups, thus
presumably overcoming any advantage associated with a 2,5 -cis stereochemistry. These considerations should be useful in synthetic design in this area. A limitation of the present method is that this type of cyclisation, when applied to homoallylic sulfonamides having a terminal alkene function (24a), gives a poor yield of a mixture of isomers ( $\mathbf{3 6 a}$ and $\mathbf{3 7 b}$ ). After completion of these studies, a possible solution to this was reported in similar iodocyclisations, but of $N$-9-phenylfluorenyl homoallylic amines 56, which give good yields of the pyrrolidines $57 .{ }^{46}$ However, the presence of an allylic acetate group may also assist these cyclisations. ${ }^{5}$

In summary, this application of the 5 -endo-trig method has provided a flexible approach to both trans- and cis-2,5disubstituted pyrrolidines. Optically pure pyrrolidines will be obtained by starting with single enantiomers of the homoallylic alcohols 22, as demonstrated in the preparation of the iodopyrrolidine 37 e , at least in examples such as this where there is no opportunity for racemization. One limitation of this present chemistry is the lack of functionality present in the substituents. In preliminary studies, we have partly addressed this deficiency by demonstrating that sulfonamides derived from $\alpha$-allylated glycine esters undergo similarly efficient and flexible iodocyclisations to provide examples of substituted prolines in a highly stereoselective manner. ${ }^{47}$ Similarly, selenocyclisations of homoallylic sulfonamides $\mathbf{2 4}$ are also highly stereoselective, although a variety of conditions have to be employed to achieve this. ${ }^{48}$ No doubt many other side-chain substituents will be compatible with the foregoing chemistry. Exceptions may well be when heteroatoms are present in positions in which these can undergo competing 5-exo cyclisations as, despite the facile nature of the cyclisations described in this paper, Baldwin's rules are likely to apply in such competitive situations.

## Experimental

## General details

Melting points were determined on a Köfler hot stage apparatus. Optical rotations were measured using an Optical Activity AA-10 polarimeter and are given in $10^{-1}$ deg $\mathrm{cm}^{2} \mathrm{~g}^{-1}$. Infrared spectra were recorded using a Perkin-Elmer 1600 series Fourier transform spectrometer using thin films between sodium chloride plates, unless otherwise stated. ${ }^{1} \mathrm{H}$ NMR spectra were determined using a Bruker WM-250, a JEOL EX-270 or a Bruker AM-400 spectrometer, operating at the frequencies indicated [i.e. (250) refers to 250 MHz etc.]. ${ }^{13} \mathrm{C}$ NMR spectra were determined using any of the latter three instruments, operating at $62.5,67.8$ and 100.1 MHz respectively, as indicated after $\delta_{\mathrm{C}}$. Unless otherwise stated, all spectra were determined using dilute solutions in deuteriochloroform and tetramethylsilane as internal standard. $J$ Values are expressed in hertz. Mass spectra were measured using either an AEI MS902 or a VG 7070E instrument, both operating in the electron impact mode, unless otherwise stated; FAB, electrospray (ES) and atmospheric pressure chemical ionization (APcI) spectra were obtained using the latter instrument or were obtained from the EPSRC Mass Spectrometry Service at UC Swansea.

Unless otherwise stated, all reactions were carried out in anhydrous solvents which were obtained by the usual methods. ${ }^{49}$ All organic solutions from work-ups were dried by brief exposure to anhydrous magnesium sulfate followed by filtration. Solvents were removed by rotary evaporation. CC refers to column chromatography over Sorbsil silica gel using the eluants specified.

The Mitsunobu reagents $\mathbf{1 7}$ were prepared in similar yields to those previously reported and showed identical spectroscopic and analytical data: $N$-tert-butoxycarbonylbenzenesulfonamide 17a, mp $94-95^{\circ} \mathrm{C}$ (lit. ${ }^{50} \mathrm{mp} 93-95^{\circ} \mathrm{C}$ ); $N$-tert-butoxy-carbonyltoluene- $p$-sulfonamide $\mathbf{1 7 b}, \mathrm{mp} 116-117^{\circ} \mathrm{C}$ (lit. ${ }^{35} \mathrm{mp}$
$115-117^{\circ} \mathrm{C}$ ); $N$-tert-butoxycarbonylmethanesulfonamide $\mathbf{1 7 c}$, mp 106-107 ${ }^{\circ} \mathrm{C}$ (lit. ${ }^{51} \mathrm{mp} 108-109^{\circ} \mathrm{C}$ ).

## Mitsunobu reactions: general procedure

To a stirred solution of an alkenol 21 or 22 ( $10 \mathrm{mmol}, 1$ eq.), a Mitsunobu sulfonamide reagent 17 ( $10 \mathrm{mmol}, 1$ eq.) and triphenylphosphine ( $11 \mathrm{mmol}, 1.1 \mathrm{eq}$.) in dry tetrahydrofuran $(60 \mathrm{ml})$ cooled in an ice-bath was added dropwise diethyl azodicarboxylate ( $11 \mathrm{mmol}, 1.1 \mathrm{eq}$.). The reaction mixture was stirred at ambient temperature for 6 h then treated with water $(50 \mathrm{ml})$ and extracted with dichloromethane $(3 \times 50 \mathrm{ml})$. The combined organic extracts were dried and evaporated to leave the crude product 23, 25, 27a or 28a which was either purified by column chromatography (cc) or deprotected without further purification (see later).

## 4-[(N-tert-Butoxycarbonylphenyl)sulfonylamino]pent-1-ene

23a. A solution of pent-4-en-2-ol ( $0.50 \mathrm{~g}, 5.8 \mathrm{mmol})$ was reacted with Mitsunobu reagent $17 \mathrm{a}(1.50 \mathrm{~g}, 5.8 \mathrm{mmol})$ as described in the general procedure to yield a yellow semi-solid which was purified by cc (hexane-ethyl acetate $50: 50$ ) to afford the title compound $23 \mathrm{a}\left(1.40 \mathrm{~g}, 74 \%\right.$ ) as a viscous, pale yellow oil: $R_{\mathrm{f}} 0.76$ (hexane-ethyl acetate $50: 50$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 2956$ (s), 2940 (s), 2880 (s), $1732(\mathrm{C}=\mathrm{O}), 1456$ (s) and $1363(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.93-7.42(5 \mathrm{H}$, $\mathrm{m}, \mathrm{Ph}), 5.72(1 \mathrm{H}$, dddd, $J 15.8,12.2,6.3$ and $6.3,2-\mathrm{H}), 5.10-$ $4.96\left(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{CH}_{2}\right), 4.64-4.58(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 2.79-2.40(2 \mathrm{H}, \mathrm{m}$, $\left.3-\mathrm{CH}_{2}\right), 1.47(3 \mathrm{H}, \mathrm{d}, J 8.3,5-\mathrm{Me})$ and $1.32\left(9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}}{ }^{\mathrm{Bu}}\right) ; \delta_{\mathrm{C}}(67.8)$ 150.4 (C=O), 140.8 (Ar-C), 135.1 (Ar-CH), 132.6 (2-CH), 128.6 ( $\mathrm{Ar}-\mathrm{CH}$ ), $127.7(\mathrm{Ar}-\mathrm{CH}), 117.6\left(1-\mathrm{CH}_{2}\right), 84.0$ ( ${ }^{\mathrm{t} B u-\mathrm{C}), ~} 54.9$ (4-CH), $39.3\left(3-\mathrm{CH}_{2}\right), 27.8$ ( ${ }^{\mathrm{t}} \mathrm{Bu}-\mathrm{Me}$ ) and $19.4(5-\mathrm{Me}) ; \mathrm{m} / \mathrm{z}(\mathrm{EI})$ $284\left(\mathrm{M}^{+}-\mathrm{C}_{3} \mathrm{H}_{5}, 30 \%\right), 198$ (4), 184 (100), 141 (61), 77 (65) and 57 (98).
( E)-3-[( $N$-tert-Butoxycarbonyl)-p-tolylsulfonylamino]dec-5ene 23f. ( $E$ )-Dec-5-en-3-ol ( $0.66 \mathrm{~g}, 4.20 \mathrm{mmol}$ ) was reacted with Mitsunobu reagent $\mathbf{1 7 b}(1.13 \mathrm{~g}, 4.20 \mathrm{mmol})$ as described in the general procedure to give a yellow solid which was purified by cc (dichloromethane) to afford the title compound $\mathbf{2 3 f}(1.20 \mathrm{~g}$, $74 \%$ ) as a pale yellow, viscous oil: $R_{\mathrm{f}} 0.68$ (dichloromethane); $\nu_{\text {max }} / \mathrm{cm}^{-1} 2964$ (s), 2930 (s), 2874 (w), 1726 (C=O), 1589 (w), 1458 (s) and 1357 (s); $\delta_{\mathrm{H}}(400) 7.83$ ( $2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{Ar}-\mathrm{H}$ ), 7.26 $(2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{Ar}-\mathrm{H}), 5.50(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $6.2,6-\mathrm{H}), 5.37$ $(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $6.3,5-\mathrm{H}), 4.34(1 \mathrm{H}, \mathrm{tt}, J 7.3$ and $6.3,3-\mathrm{H})$, 2.61-2.37 ( $2 \mathrm{H}, \mathrm{m}, 4-\mathrm{CH}_{2}$ ), 2.42 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 1.98-1.72 ( 4 H , $\left.\mathrm{m}, 2-\mathrm{and} 7-\mathrm{CH}_{2}\right), 1.40-1.11\left(4 \mathrm{H}, \mathrm{m}, 8-\mathrm{and} 9-\mathrm{CH}_{2}\right), 1.38(9 \mathrm{H}, \mathrm{s}$, $\left.{ }^{\mathrm{t}} \mathrm{Bu}\right), 0.96(3 \mathrm{H}, \mathrm{t}, J 7.3,1-\mathrm{Me})$ and $0.87(3 \mathrm{H}, \mathrm{t}, J 6.8,10-\mathrm{Me})$; $\delta_{\mathrm{C}}(100) 150.7$ (C=O), 143.6, 137.7 (both Ar-C), 133.5 (=CH), 128.8, 128.3 (both Ar-CH), 126.5 ( $=\mathrm{CH}$ ), 83.5 ('Bu-C), 65.3 (3-CH), 36.6, 32.2, 31.4 (all $\mathrm{CH}_{2}$ ), 27.9 ('Bu-Me), 26.3, 22.2 (both $\mathrm{CH}_{2}$ ), 21.4 (Ar-Me), 13.9 ( $10-\mathrm{Me}$ ) and 11.5 ( $13-\mathrm{Me}$ ); $m / z$ (EI) $336\left(\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}, 3 \%\right), 312$ (2), 254 (18), 212 (100), 156 (4), 138 (13) and 91 (38) [Found: $\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}, 336.1635$. $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{NO}_{3} \mathrm{~S}$ requires $\left.M, 336.1633\right]$.

## ( E )-3-[( $N$-tert-Butoxycarbonyl)methylsulfonylamino]dec-5-

ene 23g. ( $E$ )-Dec-5-en-3-ol ( $0.10 \mathrm{~g}, 0.65 \mathrm{mmol}$ ) was reacted with Mitsunobu reagent $17 \mathrm{c}(0.13 \mathrm{~g}, 0.65 \mathrm{mmol})$ as described in the general procedure to give a crude product as a yellow, semisolid which was purified by cc (dichloromethane) to afford the title compound $\mathbf{2 3 g}(0.16 \mathrm{~g}, 76 \%)$ as a pale yellow, viscous oil: $R_{\mathrm{f}}$ 0.43 (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 2965$ (s), 2932 (s), 2875 (w), 1724 (C=O), 1459 (w), 1395 (w) and $1353(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 5.46(1 \mathrm{H}$, $\mathrm{dt}, J 15.3$ and $6.7,6-\mathrm{H}), 5.32(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and $6.6,5-\mathrm{H}), 4.18$ $(1 \mathrm{H}, \mathrm{tt}, J 7.2$ and $6.3,3-\mathrm{H}), 3.21\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{Me}\right), 2.61-2.58(1 \mathrm{H}$, $\left.\mathrm{m}, 4-\mathrm{H}_{\alpha}\right), 2.32-2.21\left(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{\beta}\right), 1.98-1.89(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{and}$ $\left.7-\mathrm{CH}_{2}\right), 1.65-1.54\left(2 \mathrm{H}, \mathrm{m}, 8-\mathrm{CH}_{2}\right), 1.53\left(9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}} \mathrm{Bu}\right), 1.30(2 \mathrm{H}$, $\left.\mathrm{m}, 9-\mathrm{CH}_{2}\right), 0.92(3 \mathrm{H}, \mathrm{t}, J 7.4,1-\mathrm{Me})$ and $0.86(3 \mathrm{H}, \mathrm{t}, J 7.1$, $10-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 133.8,126.6$ (both $=\mathrm{CH}$ ), 83.9 ('Bu-C), 61.2 (3-CH), $42.1\left(\mathrm{SO}_{2} \mathrm{Me}\right), 36.4,32.2,31.4\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 28.0\left({ }^{(t} \mathrm{Bu}-\mathrm{Me}\right)$,
26.3, 22.1 (both $\mathrm{CH}_{2}$ ), 13.8 ( $10-\mathrm{Me}$ ) and 11.3 ( $1-\mathrm{Me}$ ); $m / z$ ( FAB ) $334\left(\mathrm{M}^{+}+1,12 \%\right), 298$ (54), 278 (100), 234 (18), 154 (20), 136 (47), 126 (10), 83 (35), 57 (90) and 55 (20) [Found: $\mathrm{M}^{+}+1$, 334.2056. $\mathrm{C}_{16} \mathrm{H}_{32} \mathrm{NO}_{4} \mathrm{~S}$ requires $M, 334.2052$ ].

## ( $E$ )-5-[( $N$-tert-Butoxycarbonyl)-p-tolylsulfonylamino]tridec-

7-ene 23i. ( $E$ )-Tridec-7-en-5-ol ( $1.90 \mathrm{~g}, 9.60 \mathrm{mmol}$ ) was reacted with Mitsunobu reagent $\mathbf{1 7 b}(2.60 \mathrm{~g}, 9.60 \mathrm{mmol})$ under the general conditions to give a crude product as a yellow semi-solid which was purified by cc (hexane-ethyl acetate $80: 20$ ) to yield the title compound $\mathbf{2 3 i}$ ( $3.20 \mathrm{~g}, 71 \%$ ) as a colourless, viscous oil: $R_{\mathrm{f}} 0.66$ (hexane-ethyl acetate $80: 20$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 2957$ (s), 2929 (s), $2860(\mathrm{w}), 1726$ (C=O), $1598(\mathrm{w}), 1356(\mathrm{~s})$ and $1278(\mathrm{~s}) ; \delta_{\mathrm{H}}$ (400) 7.83 ( $2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}$ ), 7.28 ( $2 \mathrm{H}, \mathrm{d}, J 8.2$, Ar-H), 5.43 $(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and $6.8,8-\mathrm{H}), 5.35(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and $6.6,7-\mathrm{H})$, 4.48-4.30 ( $1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}$ ), 2.64-2.37 ( $2 \mathrm{H}, \mathrm{m}, 6-\mathrm{CH}_{2}$ ), $2.43(3 \mathrm{H}, \mathrm{s}$, Ar-Me), 1.98-1.65 (4H, m, 4- and 9-CH2), 1.38 ( $9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{H}} \mathrm{Bu}$ ), $1.34-1.21\left(10 \mathrm{H}, \mathrm{m}, 2-, 3-, 10-, 11-\mathrm{and} 12-\mathrm{CH}_{2}\right), 0.82(3 \mathrm{H}, \mathrm{t}$, $J 6.8, \mathrm{Me})$ and $0.79(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me}) ; \delta_{\mathrm{C}}(100) 151.3(\mathrm{C}=\mathrm{O})$, 143.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 137.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 133.7 ( $=\mathrm{CH}$ ), 128.9 ( $\mathrm{Ar}-\mathrm{CH}$ ), 128.4 (Ar-CH), 126.7 (=CH), 83.7 ( $\left.{ }^{( } \mathrm{Bu}-\mathrm{C}\right)$, 60.1 (5-CH), 36.9, 33.7, 33.1, 32.6, 29.1 (all CH2 ), 28.0 ( ${ }^{(\mathrm{Bu}} \mathrm{Me}$ ), 26.8, 22.5 (both $\mathrm{CH}_{2}$ ), $21.53(\mathrm{Ar}-\mathrm{Me})$ and 13.9 ( $1-$ and $13-\mathrm{Me}$ ); $m / z(\mathrm{ES}) 452\left(\mathrm{M}^{+}+1\right.$, $23 \%$ ), 396 (91), 356 (100) and 245 (34).

## ( E)-4-[( $N$-tert-Butoxycarbonyl)-p-tolylsulfonylamino]-1-

phenylhex-1-ene 23j. ( $E$ )-1-Phenylhex-1-en-4-ol ( $3.00 \mathrm{~g}, 17.00$ mmol ) was reacted with Mitsunobu reagent $\mathbf{1 7 b}(4.62 \mathrm{~g}, 17.00$ mmol ) under the conditions described in the general procedure to give the crude product as a yellow semi-solid which was purified by cc (hexane-ethyl acetate $80: 20$ ) to afford the title compound $\mathbf{2 3 j}(5.60 \mathrm{~g}, 81 \%)$ as a colourless solid, mp 106 $107^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.51$ (hexane-ethyl acetate $80: 20$ ); $v_{\text {max }} / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 2962(\mathrm{~s}), 2930(\mathrm{~s}), 2862(\mathrm{~s}), 1727(\mathrm{C}=\mathrm{O})$ and $1462(\mathrm{~s}) ;$ $\delta_{\mathrm{H}}(250) 7.74$ ( $2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{Ar}-\mathrm{H}$ ), $7.33-7.19$ ( $5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}$ ), 6.99 $(2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{Ar}-\mathrm{H}), 6.39(1 \mathrm{H}, \mathrm{d}, J 15.7,1-\mathrm{H}), 6.10(1 \mathrm{H}$, ddd, $J 15.7,7.5$ and $6.8,2-\mathrm{H}), 4.50(1 \mathrm{H}$, tdd, $J 6.2,6.2$ and $6.1,4-\mathrm{H}$ ), $2.89\left(1 \mathrm{H}\right.$, ddd, $J 14.0,7.5$ and $\left.6.2,3-\mathrm{H}_{\omega}\right), 2.62(1 \mathrm{H}$, ddd, $J 14.0$, 6.8 and $\left.6.1,3-\mathrm{H}_{\beta}\right), 2.31(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.10-1.55(2 \mathrm{H}, \mathrm{m}$, $\left.5-\mathrm{CH}_{2}\right), 1.38\left(9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}} \mathrm{Bu}\right)$ and $1.01(3 \mathrm{H}, \mathrm{t}, J 7.7,6-\mathrm{Me}) ; \delta_{\mathrm{C}}(100)$ 150.9 (C=O), 143.6, 137.5, 136.9 (all Ar-C), 132.7 (Ar-CH), 128.9 (=CH), 128.4, 128.3, 127.3, 127.14 (all Ar-CH), 126.2 $(=\mathrm{CH}), 83.9$ ( ${ }^{\mathrm{B} B u-\mathrm{C}), ~} 61.3(4-\mathrm{CH}), 36.9\left(3-\mathrm{CH}_{2}\right), 28.0\left({ }^{\mathrm{t}} \mathrm{Bu}-\mathrm{Me}\right)$, $26.9\left(5-\mathrm{CH}_{2}\right), 21.5$ ( $\mathrm{Ar}-\mathrm{Me}$ ) and $11.5(6-\mathrm{Me}) ; \mathrm{m} / \mathrm{z}(\mathrm{ES}) 430$ $\left(\mathrm{M}^{+}+1,100 \%\right), 374(92), 330(84), 83$ (28) and 60 (52) [Found: C, 66.9; H, 7.4; N, 3.2. $\mathrm{C}_{24} \mathrm{H}_{31} \mathrm{NO}_{4}$ S requires C, 67.1; H, 7.3; N, 3.2\%].

## (Z)-1-[( $N$-tert-Butoxycarbonyl)-p-tolylsulfonylamino]hex-3-

 ene 25a. ( $Z$ )-Hex-3-en-1-ol $(0.50 \mathrm{~g}, 5.0 \mathrm{mmol})$ was reacted with Mitsunobu reagent $\mathbf{1 7 b}(1.35 \mathrm{~g}, 5.0 \mathrm{mmol})$ as described above to give an orange semi-solid which was purified by cc (dichloromethane) to yield the title compound $\mathbf{2 5 a}(1.30 \mathrm{~g}, 74 \%)$ as a viscous, pale yellow oil: $R_{\mathrm{f}} 0.62$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}$ 3307 (w), 2969 (s), 2933 (s), 1728 (C=O), 1356 (s) and 1289 (s); $\delta_{\mathrm{H}}(400) 7.79(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 7.29(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H})$, 5.51-5.29 (2H, m, 3- and 4-H), 3.86-3.79 ( $2 \mathrm{H}, \mathrm{m}, 1-\mathrm{CH}_{2}$ ), 2.62$2.45\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{CH}_{2}\right), 2.44(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.11(2 \mathrm{H}, \mathrm{qd}, J 7.5$ and $\left.7.4,5-\mathrm{CH}_{2}\right) 1.35\left(9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}} \mathrm{Bu}\right)$ and $0.98(3 \mathrm{H}, \mathrm{t}, J 7.5,6-\mathrm{Me})$; $\delta_{\mathrm{C}}(100) 151.0(\mathrm{C}=\mathrm{O}), 144.0$, 137.7 (both Ar-C), 134.8 (=CH), 129.2, 127.8 (both Ar-CH), 124.1 (=CH), 84.0 ('Bu-C), 46.6 (1-CH2), $28.2\left(\mathrm{CH}_{2}\right), 27.9\left({ }^{( } \mathrm{Bu}-\mathrm{Me}\right), 21.5(\mathrm{Ar}-\mathrm{Me}), 20.6\left(\mathrm{CH}_{2}\right)$ and $14.2(6-\mathrm{Me}) ; m / z(\mathrm{FAB}) 354\left(\mathrm{M}^{+}+1,36 \%\right), 298$ (100), 280 (7), 254 (17), 184 (13), 155 (33), 139 (10), 126 (30), 91 (17), 82 (13) and 57 (32) [Found: $\mathrm{M}^{+}+1,354.1749 . \mathrm{C}_{18} \mathrm{H}_{28} \mathrm{NO}_{4} \mathrm{~S}$ requires $M, 354.1739$ ].(Z)-1-[(N-tert-Butoxycarbonyl)methylsulfonylamino]hex-3ene 25b. ( $Z$ )-Hex-3-en-1-ol ( $0.10 \mathrm{~g}, 1.0 \mathrm{mmol}$ ) was reacted with Mitsunobu reagent $17 \mathrm{c}(0.19 \mathrm{~g}, 1.0 \mathrm{mmol})$ as described in the
general procedure to give a dark yellow semi-solid which was purified by cc (dichloromethane) to yield the title compound 25b $(0.20 \mathrm{~g}, 80 \%)$ as a thick, pale yellow oil: $R_{\mathrm{f}} 0.52$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 3008$ (w), 2973 (s), 2935 (s), 2875 (w), 1726 $(\mathrm{C}=\mathrm{O}), 1487(\mathrm{w})$ and $1355(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 5.48-5.29$ ( $2 \mathrm{H}, \mathrm{m}, 3$ - and $4-\mathrm{H}), 3.69\left(2 \mathrm{H}, \mathrm{t}, J 7.3,1-\mathrm{CH}_{2}\right), 3.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{Me}\right), 2.37(2 \mathrm{H}$, $\mathrm{td}, J 7.3$ and $\left.7.2,2-\mathrm{CH}_{2}\right), 2.05\left(2 \mathrm{H}, \mathrm{qd}, J 7.5\right.$ and $\left.7.3,5-\mathrm{CH}_{2}\right)$, $1.54\left(9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}} \mathrm{Bu}\right)$ and $0.95(3 \mathrm{H}, \mathrm{t}, J 7.5,6-\mathrm{Me})$; $\delta_{\mathrm{C}}(100) 151.6$ $(\mathrm{C}=\mathrm{O}), 134.6,124.1($ both $=\mathrm{CH}), 84.1\left({ }^{( } \mathrm{Bu}-\mathrm{C}\right), 45.6\left(1-\mathrm{CH}_{2}\right)$, $42.0\left(\mathrm{SO}_{2} \mathrm{Me}\right), 27.8\left({ }^{( } \mathrm{Bu}-\mathrm{Me}\right), 27.4,20.3$ (both $\left.\mathrm{CH}_{2}\right)$ and 14.0 (6-Me); $m / z(\mathrm{FAB}) 278\left(\mathrm{M}^{+}+1,36 \%\right), 222(100), 204$ (9), 178 (33), 154 (44), 137 (41), 83 (23) and 57 (73) [Found: $\mathrm{M}^{+}+1$, 278.1429. $\mathrm{C}_{12} \mathrm{H}_{24} \mathrm{NO}_{4} \mathrm{~S}$ requires $M, 278.1426$ ].
(1RS,2RS)-2-(Hex-1-ynyl)-1-phthalimidocyclohexane 30a. A solution of phthalimide ( $5.20 \mathrm{~g}, 36.0 \mathrm{mmol}$ ) and triphenylphosphine ( $9.40 \mathrm{~g}, 36.0 \mathrm{mmol}$ ) in dry THF ( 100 ml ) was stirred under nitrogen at ambient temperature. To this was added alkynol $29(2.16 \mathrm{~g}, 12 \mathrm{mmol})^{7}$ and diethyl azodicarboxylate $(5 \mathrm{ml}, 36.0 \mathrm{mmol})$. Stirring was continued for 48 h , after which time the solvent was evaporated and the crude product purified by cc (hexane-ether $80: 20$ ) to furnish the phthalimide 30a ( 1.50 $\mathrm{g}, 46 \%$ ) as a viscous oil: $v_{\max } / \mathrm{cm}^{-1} 1772(\mathrm{C}=\mathrm{O}), 1714(\mathrm{C}=\mathrm{O})$ and 1330; $\delta_{\mathrm{H}}$ (400) 7.85-7.78 ( $2 \mathrm{H}, \mathrm{m}$, Ar-H), 7.73-7.67 ( $2 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar}-\mathrm{H}), 4.11(1 \mathrm{H}, \mathrm{dt}, J 13.0$ and $3.6,1-\mathrm{H}), 3.53-3.02(1 \mathrm{H}, \mathrm{m})$, $2.09\left(2 \mathrm{H}, \mathrm{td}, J 7.0\right.$ and $\left.2.3,3^{\prime}-\mathrm{CH}_{2}\right), 1.97-1.22(10 \mathrm{H}, \mathrm{m})$ and $0.90(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 168.5$ (Ar-C), 133.6, 122.8 (both Ar-CH), 83.5, 79.2 (both C), 55.4, 33.0 (both CH), 31.1, 26.3, 24.9, 22.1, 20.9, $20.7\left(\mathrm{all} \mathrm{CH}_{2}\right)$ and $13.40(\mathrm{Me}) ; m / z(\mathrm{ES})$ $310\left(\mathrm{M}^{+}+1,100 \%\right), 151(24)$ and 134 (51) [Found: $\mathrm{M}^{+}+1$, 310.1807. $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{NO}_{2}$ requires $M, 310.1806$ ].

## Boc deprotections: general procedure

To a stirred solution of an N -Boc-arene (or alkane) sulfonamide ( $10.0 \mathrm{mmol}, 1 \mathrm{eq}$.) in dichloromethane ( 50 ml ) was added dropwise trifluoroacetic acid ( $11.4 \mathrm{ml}, 100 \mathrm{mmol}, 10$ eq.). The resulting mixture was stirred at ambient temperature, typically for $10-20 \mathrm{~h}$. Once the reaction was complete, the mixture was neutralised using 2 M aqueous sodium carbonate and the resulting two layers separated. The aqueous layer was extracted with dichloromethane $(3 \times 50 \mathrm{ml})$ and the combined organic solutions dried and evaporated. The resulting crude product was purified by column chromatography. In examples of crude compounds from the Mitsunobu reaction, the yields for the deprotection step and the quantity of trifluoroacetic acid used were based on the starting homoallylic alcohol.

4-(Phenylsulfonylamino)pent-1-ene 24a. ${ }^{\mathbf{5 2}}$ Boc-Sulfonamide 23a ( $0.50 \mathrm{~g}, 1.54 \mathrm{mmol}$ ) was deprotected as described in the general procedure to give the crude product as a yellow oil, which was purified by cc (hexane-ethyl acetate $60: 40$ ) to afford the title compound $\mathbf{2 4 a}(0.28 \mathrm{~g}, 82 \%)$ as a viscous, colourless oil: $R_{\mathrm{f}} 0.67$ (hexane-ethyl acetate $60: 40$ ); $\delta_{\mathrm{H}}(250) 7.92-7.49$ ( 5 H , $\mathrm{m}, \mathrm{Ph}), 5.57(1 \mathrm{H}, \mathrm{ddt}, J 15.6,12.4$ and $6.4,2-\mathrm{H}), 5.02-4.87(2 \mathrm{H}$, $\mathrm{m}, 1-\mathrm{CH}_{2}$ ), 3.49 ( $1 \mathrm{H}, \mathrm{qdd}, J$ 8.4, 6.3 and $6.2,4-\mathrm{H}$ ), 2.20-2.03 $\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{CH}_{2}\right)$ and $1.07(3 \mathrm{H}, \mathrm{d}, J 8.4,5-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 140.9$ (Ar-C), 133.3 (Ar-CH), 132.4 (2-CH), 128.9, 126.9 (both $\mathrm{Ar}-\mathrm{CH}), 118.5\left(1-\mathrm{CH}_{2}\right), 49.3(4-\mathrm{CH}), 41.3\left(3-\mathrm{CH}_{2}\right)$ and 21.0 (5-Me); $m / z(\mathrm{EI}) 210\left(\mathrm{M}^{+}-15,1 \%\right), 184$ (93), 141 (72), 125 (3), 86 (8), 77 (100) and 59 (7).
( $\boldsymbol{E}$ )-1-( $\boldsymbol{p}$-Tolylsulfonylamino)hex-3-ene 24b. ( $E$ )-Hex-3-en-1ol ( $0.50 \mathrm{~g}, 5.0 \mathrm{mmol}$ ) was reacted with Mitsunobu reagent 17b ( $1.35 \mathrm{~g}, 5 \mathrm{mmol}$ ) under the conditions described above to give the $(E)$-Boc-sulfonamide 23b as a yellow semi-solid which was not further purified but which was immediately deprotected.

Crude ( $E$ )-Boc-sulfonamide 23b from the Mitsunobu reaction was treated with trifluoroacetic acid ( $5.7 \mathrm{ml}, 50 \mathrm{mmol}$ ) as described above and the crude product purified by cc (hexane-
ethyl acetate $60: 40$ ) to yield the sulfonamide $\mathbf{2 4 b}(0.86 \mathrm{~g}, 67 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.70$ (hexane-ethyl acetate $60: 40$ ); $v_{\max } / \mathrm{cm}^{-1} 3272(\mathrm{NH}), 2968$ (s), 2923 (s), 1598 (w) and 1315 (s); $\delta_{\mathrm{H}}(250) 7.73(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 7.30(2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $5.45(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $6.2,4-\mathrm{H}), 5.16(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $7.1,3-\mathrm{H}), 4.62(1 \mathrm{H}, \mathrm{m}, \mathrm{NH}), 2.99-2.91\left(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{CH}_{2}\right)$, $2.42(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.15-2.08\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{CH}_{2}\right), 1.93(2 \mathrm{H}, \mathrm{qd}$, $J 7.4$ and $\left.6.2,5-\mathrm{CH}_{2}\right)$ and $0.90(3 \mathrm{H}, \mathrm{t}, J 7.4,6-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8)$ 143.3, 139.9 (both Ar-C), 135.9 (=CH), 129.6, 127.1 (both Ar-CH), 124.3 (=CH), $42.6\left(1-\mathrm{CH}_{2}\right), 32.3,25.4$ (both $\left.\mathrm{CH}_{2}\right), 21.5$ (Ar-Me) and 13.6 (6-Me); m/z (ES) $254\left(\mathrm{M}^{+}+1,100 \%\right), 252$ (22), 224 (13), 184 (6) and 83 (4) [Found: $\mathrm{M}^{+}+1,254.1214$. $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{NO}_{2} \mathrm{~S}$ requires $\left.M, 254.1215\right]$.
( $\boldsymbol{E}$ )-1-(Phenylsulfonylamino)hex-3-ene 24c. ( $E$ )-Hex-3-en-1-ol $(0.21 \mathrm{~g}, 2.0 \mathrm{mmol})$ was reacted with Mitsunobu reagent $\mathbf{1 7 a}$ ( $0.54 \mathrm{~g}, 5 \mathrm{mmol}$ ) under the conditions described above to give the ( $E$ )-Boc-sulfonamide 23c as a yellow semi-solid which was not further purified but which was immediately deprotected.

Crude ( $E$ )-Boc-sulfonamide 23c was treated with trifluoroacetic acid $(2.3 \mathrm{ml}, 20 \mathrm{mmol})$ as described above and the crude product purified by cc (hexane-ethyl acetate $60: 40$ ) to afford the title compound $\mathbf{2 4 c}(0.35 \mathrm{~g}, 70 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.70$ (hexane-ethyl acetate $60: 40$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 3275$ (NH), 2970 (s), 2925 (s), $1660(\mathrm{w})$, and 1317 (s); $\delta_{\mathrm{H}}$ (250) 7.91$7.46(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 5.44(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $6.2,4-\mathrm{H}), 5.22(1 \mathrm{H}$, $\mathrm{dt}, J 15.5$ and $7.1,3-\mathrm{CH}), 3.02-2.91$ ( $2 \mathrm{H}, \mathrm{m}, 1-\mathrm{CH}_{2}$ ), 2.20-2.08 $\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{CH}_{2}\right), 1.93\left(2 \mathrm{H}, \mathrm{qd}, J 7.4\right.$ and $\left.6.2,5-\mathrm{CH}_{2}\right)$ and 0.90 (3H, t, J 7.4, 6-Me); $\delta_{\mathrm{C}}$ (67.8) 139.8 (Ar-C), 135.4 (Ar-CH), 132.3 ( $=\mathrm{CH}$ ), 129.2, 126.8 (both Ar-CH), 124.1 (=CH), 42.6 ( $1-\mathrm{CH}_{2}$ ), 32.2, $25.2\left(\right.$ both $\mathrm{CH}_{2}$ ) and $13.3(6-\mathrm{Me})$; $m / z(\mathrm{EI}) 240$ $\left(\mathrm{M}^{+}+1,5 \%\right), 198$ (6), 170 (90), 141 (97), 125 (6), 82 (7), 77 (100), 69 (14), 51 (47) and 41 (44) [Found: $\mathrm{M}^{+}+1,240.1048$. $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{NO}_{2} \mathrm{~S}$ requires $\left.M, 240.1058\right]$.
( $E$ )-2-( $\boldsymbol{p}$-Tolylsulfonylamino)non-4-ene 24d. Racemic ( $E$ )-non-4-en-2-ol ( $0.60 \mathrm{~g}, 4.2 \mathrm{mmol}$ ) was reacted with Mitsunobu reagent $\mathbf{1 7 b}(1.14 \mathrm{~g}, 4.2 \mathrm{mmol})$ under the conditions described above to give the sulfonamide $\mathbf{2 3 d}(1.20 \mathrm{~g}, 74 \%)$ as a pale yellow, viscous oil: $R_{\mathrm{f}} 0.67$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 3010$ (w), 2980 (s), 2963 (s), 2880 (s), 1735 (C=O), 1384 (s) and 1295 (s); $\delta_{\mathrm{H}}(250) 7.80(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 7.29(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H})$, $5.46(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $6.5,5-\mathrm{H}), 5.29(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $6.6,4-\mathrm{H}), 4.56(1 \mathrm{H}, \mathrm{qt}, J 6.8$ and $6.6,2-\mathrm{H}), 2.68-2.33(2 \mathrm{H}$, $\left.\mathrm{m}, 3-\mathrm{CH}_{2}\right), 2.43(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 1.46(3 \mathrm{H}, \mathrm{d}, J 6.8,1-\mathrm{Me})$, $1.37\left(9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}} \mathrm{Bu}\right), 1.33-1.26\left(4 \mathrm{H}, \mathrm{m}, 7-\mathrm{and} 8-\mathrm{CH}_{2}\right)$ and 0.88 $(3 \mathrm{H}, \mathrm{t}, J 8.6,9-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 150.6(\mathrm{C}=\mathrm{O}), 143.6,136.9$ (both Ar-C), 133.7 ( $=\mathrm{CH}$ ), 128.9, 127.8 (both Ar-CH), 126.4 (=CH), 83.8 ('Bu-C), $55.5(2-\mathrm{CH}), 37.9,32.2,31.4$ (all $\mathrm{CH}_{2}$ ),
 14.0 ( $9-\mathrm{Me}$ ).
( $E$ )-Boc-Sulfonamide $\mathbf{2 3 d}$ ( $1.98 \mathrm{~g}, 5.0 \mathrm{mmol}$ ) was deprotected as described in the general procedure and the product purified by cc (dichloromethane) to yield the title compound $\mathbf{2 4 d}(1.15 \mathrm{~g}$, $78 \%$ ) as a pale yellow, viscous oil: $R_{\mathrm{f}} 0.36$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 3274(\mathrm{NH}), 2963(\mathrm{~s}), 2935(\mathrm{~s}), 2864$ (w), 1603 (w) and 1459 (s); $\delta_{\mathrm{H}}(250) 7.75(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 7.30$ ( $2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $5.38(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $6.6,5-\mathrm{H}), 5.10(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $7.2,4-\mathrm{H}), 4.33(1 \mathrm{H}, \mathrm{d}, J 7.2, \mathrm{NH}), 3.31(1 \mathrm{H}, \mathrm{dqt}, J 7.2,6.5$ and $6.2,2-\mathrm{H}), 2.43(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.04(2 \mathrm{H}, \mathrm{dd}, J 7.2$ and 6.2 , 3- $\mathrm{CH}_{2}$ ), 1.94-1.91 ( $2 \mathrm{H}, \mathrm{m}, 6-\mathrm{CH}_{2}$ ), 1.38-1.13 ( $4 \mathrm{H}, \mathrm{m}, 7$ - and $\left.8-\mathrm{CH}_{2}\right), 1.09(3 \mathrm{H}, \mathrm{d}, J 6.5,1-\mathrm{Me})$ and $0.90(3 \mathrm{H}, \mathrm{t}, J 7.0,9-\mathrm{Me})$; $\delta_{\mathrm{C}}(100) 143.6,138.3$ (Ar-C), $135.5(=\mathrm{CH}$ ), 130.0, 127.5 (both Ar-CH), 124.9 (=CH), 49.9 (2-CH), 40.5, 32.6, 31.8, 22.6 (all $\mathrm{CH}_{2}$ ), 21.7 ( $\mathrm{Ar}-\mathrm{Me}$ ), $19.0(1-\mathrm{Me})$ and 14.3 ( $9-\mathrm{Me}$ ); $m / z(\mathrm{FAB})$ $296\left(\mathrm{M}^{+}+1,23 \%\right), 289$ (15), 254 (6), 198 (18), 154 (100), 137 (70), 120 (12), 107 (21), 95 (6) and 77 (17) [Found: C, 63.5; $\mathrm{H}, 8.6 ; \mathrm{N}, 4.5 ; \mathrm{M}^{+}+1$, 296.1693. $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{NO}_{2} \mathrm{~S}$ requires $\mathrm{C}, 64.0$; $\mathrm{H}, 8.5 ; \mathrm{N}, 4.7 \% ; \mathrm{C}_{16} \mathrm{H}_{26} \mathrm{NO}_{2} \mathrm{~S}$ requires $\left.M, 296.1684\right]$.
(2R)-(E)-2-(p-Tolylsulfonylamino)non-4-ene 24e. (2R)-(E)-Non-4-en-2-ol $(0.34 \mathrm{~g}, 2.40 \mathrm{mmol})$ was reacted with Mitsunobu reagent 17b ( $0.65 \mathrm{~g}, 2.40 \mathrm{mmol}$ ) under the conditions described above to give $(2 R)-(E)-2-(N$-tert-butoxycarbonyl-p-tolylsulfonylamino)non-4-ene $\quad \mathbf{2 3 e}$ as a yellow, semi-solid which was not further purified but was immediately deprotected.

Crude $(R)-(E)$-Boc-sulfonamide 23e was deprotected with trifluoroacetic acid ( $2.9 \mathrm{ml}, 24 \mathrm{mmol}$ ) as described in the general procedure to afford the title compound $\mathbf{2 4 e}(0.50 \mathrm{~g}, 71 \%)$ as a pale yellow, viscous oil: $[a]_{\mathrm{D}}=+33.5\left(c 0.01, \mathrm{CHCl}_{3}\right.$ ); other spectral data were identical to those displayed by the racemic sulfonamide 24d (see above).
( $\boldsymbol{E}$ )-3-( $\boldsymbol{p}$-Tolylsulfonylamino)dec-5-ene 24f. ( $E$ )-Boc-Sulfonamide $23 \mathrm{f}(0.50 \mathrm{~g}, 1.2 \mathrm{mmol})$ was deprotected as described above and purified by cc (dichloromethane) to afford the sulfonamide $\mathbf{2 4 f}(0.34 \mathrm{~g}, 92 \%)$ as a viscous, colourless oil: $R_{\mathrm{f}} 0.36$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3280(\mathrm{NH}), 2959$ (s), 2927 (s), 2872 (w), 1598 (w), 1454 (s) and 1327 (s); $\delta_{\mathrm{H}}(400) 7.77(2 \mathrm{H}, \mathrm{d}$, $J 8.3, \mathrm{Ar}-\mathrm{H}), 7.28(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 5.34(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $6.6,6-\mathrm{H}), 5.10(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $7.3,5-\mathrm{H}), 4.91(1 \mathrm{H}, \mathrm{d}, J 7.9$, $\mathrm{NH}), 3.14(1 \mathrm{H}, \mathrm{dtt}, J 7.9,7.3$ and $6.5,3-\mathrm{H}), 2.41(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Ar}-\mathrm{Me}), 2.03\left(2 \mathrm{H}\right.$, app. dd, $J 7.3$ and $\left.6.5,4-\mathrm{CH}_{2}\right), 1.96-1.82$ $\left(2 \mathrm{H}, \mathrm{m}, 7-\mathrm{CH}_{2}\right), 1.44\left(2 \mathrm{H}, \mathrm{dq}, J 7.3\right.$ and $\left.6.9,2-\mathrm{CH}_{2}\right), 1.28-1.23$ $\left(4 \mathrm{H}, \mathrm{m}, 8-\mathrm{and} 9-\mathrm{CH}_{2}\right), 0.87(3 \mathrm{H}, \mathrm{t}, J 6.9,1-\mathrm{Me})$ and $0.81(3 \mathrm{H}, \mathrm{t}$, $J 7.4,10-\mathrm{Me}) ; \delta_{\mathrm{C}}(100) 142.8,138.3$ (both Ar-C), 134.6 (=CH), 129.3, 126.9 (both Ar-CH), 124.4 (=CH), 54.9 (3-CH), 37.1, 32.1, 31.3, 27.3, $22.0\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 21.3(\mathrm{Ar}-\mathrm{Me}), 13.7(10-\mathrm{Me})$ and 9.6 ( $1-\mathrm{Me}$ ), $m / z(\mathrm{ES}) 310\left(\mathrm{M}^{+}+1,100\right), 279$ (6), 139 (13) and 83 (12) [Found: $\mathrm{M}^{+}+1$, 310.1843. $\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{NO}_{2} \mathrm{~S}$ requires $M$, 310.1841].
( $\boldsymbol{E}$ )-3-(Methylsulfonylamino)dec-5-ene 24g. ( $E$ )-Boc-Sulfonamide $\mathbf{2 3 g}(0.20 \mathrm{~g}, 0.65 \mathrm{mmol})$ was deprotected as described above and purified by cc (dichloromethane) to yield the title compound $\mathbf{2 4 g}(0.12 \mathrm{~g}, 77 \%)$ as a pale yellow, viscous oil: $R_{\mathrm{f}} 0.39$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3286(\mathrm{NH}), 2960(\mathrm{~s}), 2930(\mathrm{~s})$, $2874(\mathrm{w}), 1581(\mathrm{w}), 1434(\mathrm{~s}), 1413(\mathrm{~s})$ and $1379(\mathrm{~s}) ; \delta_{\mathrm{H}}(270) 5.39$ $(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $6.6,6-\mathrm{H}), 5.22(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and $7.2,5-\mathrm{H})$, $4.22(1 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{NH}), 3.15(1 \mathrm{H}, \mathrm{dtt}, J 8.3,8.0$ and $6.9,3-\mathrm{H})$, $2.79\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{Me}\right), 2.08\left(2 \mathrm{H}, \mathrm{dd}, J 7.2\right.$ and $\left.6.9,4-\mathrm{CH}_{2}\right), 1.87$ $\left(2 \mathrm{H}, \mathrm{m}, 7-\mathrm{CH}_{2}\right), 1.52\left(2 \mathrm{H}, \mathrm{dq}, J 8.0\right.$ and $\left.7.1,2-\mathrm{CH}_{2}\right), 1.24-1.03$ $\left(4 \mathrm{H}, \mathrm{m}, 8-\mathrm{and} 9-\mathrm{CH}_{2}\right), 0.80(3 \mathrm{H}, \mathrm{t}, J 7.2,10-\mathrm{Me})$ and $0.74(3 \mathrm{H}$, $\mathrm{t}, J 7.1,1-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 135.3,124.5$ (both $\left.=\mathrm{CH}\right), 55.5(3-\mathrm{CH})$, 41.8 ( $\mathrm{SO}_{2} \mathrm{Me}$ ), 38.0, 32.2, 31.4, 27.9, 22.2 (all $\mathrm{CH}_{2}$ ), 13.8 (10Me ) and $10.0(1-\mathrm{Me}) ; m / z(\mathrm{ES}) 234\left(\mathrm{M}^{+}+1,100 \%\right)$, 231 (33), 172 (70) and 62 (35) [Found: $\mathrm{M}^{+}+1,234.1530 . \mathrm{C}_{11} \mathrm{H}_{24} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 234.1528$ ].
( $\boldsymbol{E}$ )-3-( $\boldsymbol{p}$-Tolylsulfonylamino)undec-5-ene 24h. (E)-Undec-5-en-3-ol ( $1.00 \mathrm{~g}, 6.3 \mathrm{mmol}$ ) was reacted with Mitsunobu reagent 17b ( $1.71 \mathrm{~g}, 6.3 \mathrm{mmol}$ ) under the conditions described in the general procedure to give the Boc-sulfonamide 23h as an orange semi-solid which was not further purified but was immediately deprotected.

Crude ( $E$ )-Boc-sulfonamide 23h was treated with trifluoroacetic acid ( $7.2 \mathrm{ml}, 62 \mathrm{mmol}$ ) as described above and the crude product purified by cc (dichloromethane) to afford the sulfonamide $\mathbf{2 4 h}(0.14 \mathrm{~g}, 68 \%)$ as a viscous, colourless oil: $R_{\mathrm{f}} 0.37$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3280(\mathrm{NH}), 2958$ (s), 2927 (s), $2870(\mathrm{w}), 1600(\mathrm{w}), 1453(\mathrm{~s})$ and $1328(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 7.77(2 \mathrm{H}, \mathrm{d}$, $J 8.2, \mathrm{Ar}-\mathrm{H}), 7.28(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 5.37(1 \mathrm{H}, \mathrm{dt}, J 15.4$ and $6.6,6-\mathrm{H}), 5.09(1 \mathrm{H}, \mathrm{dt}, J 15.4$ and $7.4,5-\mathrm{H}), 4.55(1 \mathrm{H}, \mathrm{d}, J 7.9$, $\mathrm{NH}), 3.14(1 \mathrm{H}, \mathrm{dtt}, J 7.9,7.4$ and $6.6,3-\mathrm{H}), 2.42(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Ar}-\mathrm{Me}), 2.03\left(2 \mathrm{H}, \mathrm{dd}, J 7.4\right.$ and $\left.6.6,4-\mathrm{CH}_{2}\right), 1.91-1.77(2 \mathrm{H}, \mathrm{m}$, $\left.7-\mathrm{CH}_{2}\right), 1.43\left(2 \mathrm{H}, \mathrm{dq}, J 7.4\right.$ and $\left.6.9,2-\mathrm{CH}_{2}\right), 1.39-1.17(6 \mathrm{H}, \mathrm{m}$, $8-, 9-$ and $\left.10-\mathrm{CH}_{2}\right), 0.87(3 \mathrm{H}, \mathrm{t}, J 6.9,1-\mathrm{Me})$ and $0.82(3 \mathrm{H}, \mathrm{t}$, $J 7.4,10-\mathrm{Me}) ; \delta_{\mathrm{C}}(100) 143.0,138.2$ (both Ar-C), 135.0 (=CH), 129.5, 126.9 (both $\mathrm{Ar}-\mathrm{CH}$ ), 124.3 ( $=\mathrm{CH}$ ), 54.9 (3-CH), 37.1,
32.5, 31.4, 29.1, 27.4, $22.4\left(\right.$ all $_{2} \mathrm{CH}_{2}$ ), 21.3 ( $\mathrm{Ar}-\mathrm{Me}$ ), 14.0 ( $10-\mathrm{Me}$ ) and 9.8 (1-Me) [Found: $(\mathrm{APcI}) \mathrm{M}^{+}+1,324.1994 . \mathrm{C}_{18} \mathrm{H}_{30} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 324.1997]$.
( $E$ )-5-( $\boldsymbol{p}$-Tolylsulfonylamino)tridec-7-ene 24i. ( $E$ )-Boc-Sulfonamide $\mathbf{2 3 i}(2.20 \mathrm{~g}, 4.8 \mathrm{mmol})$ was deprotected as described in the general procedure and the product purified by cc (hexaneethyl acetate $9: 1$ ) to afford the title compound $\mathbf{2 4 i}(1.30 \mathrm{~g}, 77 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.43$ (hexane-ethyl acetate $9: 1$ ); $v_{\max } / \mathrm{cm}^{-1} 3278$ (NH), 2955 (w), 2929 (s), 2858 (s), 1778 (w), $1598(\mathrm{w}), 1457(\mathrm{~s}), 1425(\mathrm{~s})$ and $1326(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.77(2 \mathrm{H}, \mathrm{d}$, $J 8.3$, Ar-H), $7.28(2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $5.30(1 \mathrm{H}, \mathrm{dt}, J 15.1$ and $6.6,8-\mathrm{H}), 5.05(1 \mathrm{H}, \mathrm{dt}, J 15.1$ and $7.2,7-\mathrm{H}), 4.72(1 \mathrm{H}, \mathrm{d}, J 7.9$, $\mathrm{NH}), 3.21(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.47(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.15-1.72(6 \mathrm{H}$, $\mathrm{m}, 4-, 6-$ and $\left.9-\mathrm{CH}_{2}\right), 1.52-1.03(10 \mathrm{H}, \mathrm{m}, 2-$, $3-, 10-$, 11- and $\left.12-\mathrm{CH}_{2}\right), 0.97(3 \mathrm{H}, \mathrm{t}, J 7.2,1-\mathrm{Me})$ and $0.86(3 \mathrm{H}, \mathrm{t}, J 7.1$, $13-\mathrm{Me}$ ); $\delta_{\mathrm{C}}(100) 143.1,138.4$ (both Ar-C), 135.3 (=CH), 129.5, 127.1 (both Ar-CH), 124.3 (=CH), 54.1 (5-CH), 34.9, 34.8, 31.5, 31.4, 27.5, 27.4, 24.9, $22.6\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 22.5(\mathrm{Ar}-\mathrm{Me})$ and 13.9 (1- and $13-\mathrm{Me}) ; m / z$ (ES) $352\left(\mathrm{M}^{+}+1,100 \%\right), 350$ (46), 312 (42), 172 (78) and 60 (39) [Found: $\mathrm{M}^{+}+1,352.2310$. $\mathrm{C}_{20} \mathrm{H}_{34} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 352.2310$ ].
(E)-4-(p-Tolylsulfonylamino)-1-phenylhex-1-ene 24j. (E)-BocSulfonamide $\mathbf{2 3 j}$ ( $3.00 \mathrm{~g}, 7 \mathrm{mmol}$ ) was deprotected as described above using trifluoroacetic acid ( $7.98 \mathrm{~g}, 70 \mathrm{mmol}$ ) and the product purified by cc (hexane-ethyl acetate $9: 1$ ) to yield the title compound $\mathbf{2 4 j}(2.20 \mathrm{~g}, 95 \%)$ as a colourless, crystalline solid, $\mathrm{mp} 76-77^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.26$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 3280(\mathrm{NH}), 2965$ (s), 2931 (s), 2875 (w), 1597 (w), 1448 (s), 1422 (s) and $1324(\mathrm{~s}) ; \delta_{\mathrm{H}}$ (250) $7.74(2 \mathrm{H}, \mathrm{d}, J 8.2$, Ar-H), $7.31-7.19(7 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 6.28(1 \mathrm{H}, \mathrm{d}, J 15.8,1-\mathrm{H}), 5.86(1 \mathrm{H}, \mathrm{dt}$, $J 15.8$ and $7.4,2-\mathrm{H}), 4.69(1 \mathrm{H}, \mathrm{d} . J 7.2, \mathrm{NH}), 3.28(1 \mathrm{H}$, dtt, $J 7.2,7.0$ and $6.7,4-\mathrm{H}), 2.38(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.27(2 \mathrm{H}, \mathrm{dd}, J 7.4$ and $\left.6.7,3-\mathrm{CH}_{2}\right), 1.54\left(2 \mathrm{H}, \mathrm{qd}, J 7.6\right.$ and $\left.7.0,5-\mathrm{CH}_{2}\right)$ and 0.85 $(3 \mathrm{H}, \mathrm{t}, J 7.6,6-\mathrm{Me}) ; \delta_{\mathrm{C}}(100) 141.6,136.4,135.1$ (all Ar-C), 131.9 (=CH), 128.0, 126.9, 125.8, 125.5, 124.5, (all Ar-CH), $123.5(=\mathrm{CH}), 53.6(4-\mathrm{CH}), 36.3,26.3\left(\right.$ both $\left.\mathrm{CH}_{2}\right), 19.9(\mathrm{Ar}-\mathrm{Me})$ and $8.3(6-\mathrm{Me}) ; m / z(\mathrm{ES}) 330\left(\mathrm{M}^{+}+1,100 \%\right), 176$ (7) and 159 (19) [Found: C, 69.3; H, 7.3; N, 4.45. $\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{NO}_{2} \mathrm{~S}$ requires C, 69.3; H, 7.0; N, 4.25\%].
(E)-4-Phenylsulfonylamino-1-phenylhex-1-ene 24 k . (E)-1-Phenylhex-1-en-4-ol ( $2.47 \mathrm{~g}, 14 \mathrm{mmol}$ ) was reacted with the Mitsunobu reagent $\mathbf{1 7 a}(3.59 \mathrm{~g}, 14 \mathrm{mmol})$ under the conditions described in the general procedure to give a crude product $\mathbf{2 3 k}$ which was deprotected immediately by treatment with trifluoroacetic acid ( $16 \mathrm{~g}, 140 \mathrm{mmol}$ ) as described above. Purification of the residue by cc (hexane-ethyl acetate $9: 1$ ) gave the benzenesulfonamide $\mathbf{2 4 k}(3.175 \mathrm{~g}, 72 \%)$ as a colourless, crystalline solid, $\mathrm{mp} 82-83{ }^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.24$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 3270(\mathrm{NH}), 2980$ (s), 2930 (s), 2875 (w), 1600 (w), 1450 (s), $1410(\mathrm{~s})$ and $1320(\mathrm{~s}) ; \delta_{\mathrm{H}} 7.85-7.14(10 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{Ph}), 6.24$ $(1 \mathrm{H}, \mathrm{d}, J 15.6,1-\mathrm{H}), 5.79(1 \mathrm{H}, \mathrm{dt}, J 15.6$ and $7.3,2-\mathrm{H}), 4.87$ $(1 \mathrm{H}, \mathrm{d}, J 7.1, \mathrm{NH}), 3.23(1 \mathrm{H}, \mathrm{dtt}, J 7.1,6.9$ and $6.7,4-\mathrm{H}), 2.33$ $\left(2 \mathrm{H}\right.$, dd, $J 7.3$ and $\left.6.7,3-\mathrm{CH}_{2}\right), 1.55(2 \mathrm{H}, \mathrm{qd}, J 7.6$ and 6.9 , $\left.5-\mathrm{CH}_{2}\right)$ and $0.87(3 \mathrm{H}, \mathrm{t}, J 7.6,6-\mathrm{Me}) ; m / z(\mathrm{ES}) 316\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ) [Found: C, $68.5 ; \mathrm{H}, 6.9 ; \mathrm{N}, 4.5 . \mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{2} \mathrm{~S}$ requires C, 68.5; H, 6.7; N, 4.4\%].
( $Z$ )-1-( $p$-Tolylsulfonylamino)hex-3-ene 26a. ( $Z$ )-Boc-Sulfonamide $25 \mathrm{a}(1.30 \mathrm{~g}, 3.7 \mathrm{mmol})$ was deprotected as described above and the crude product purified by cc (dichloromethane) to give the sulfonamide $\mathbf{2 6 a}(0.70 \mathrm{~g}, 72 \%)$ as a viscous, colourless oil: $R_{\mathrm{f}} 0.43$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3284(\mathrm{NH}), 2964$ (s), 2934 (s), 1656 (w), 1598 (w), 1426 (s) and 1325 (s); $\delta_{\mathrm{H}}$ (400) 7.76 ( $2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}$ ), $7.30(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 5.49(1 \mathrm{H}, \mathrm{dt}$ $J 10.8$ and $7.3,4-\mathrm{H}), 5.25(1 \mathrm{H}$, dt $J 10.8$ and $7.3,3-\mathrm{H}), 5.01(1 \mathrm{H}$, $\mathrm{t}, J 6.1, \mathrm{NH}), 2.95\left(2 \mathrm{H}, \mathrm{td}, J 6.9\right.$ and $\left.6.1,1-\mathrm{CH}_{2}\right), 2.42(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Ar}-\mathrm{Me}), 2.20\left(2 \mathrm{H}, \mathrm{dt}, J 7.3\right.$ and $\left.6.9,2-\mathrm{CH}_{2}\right), 2.01(2 \mathrm{H}, \mathrm{qd}, J 7.6$
and $\left.7.3,5-\mathrm{CH}_{2}\right)$ and $0.92(3 \mathrm{H}, \mathrm{t}, J 7.6,6-\mathrm{Me}) ; \delta_{\mathrm{C}}(100) 143.2$, 136.5 (both Ar-C), 135.0 (=CH), 129.5, 127.0 (both Ar-CH), $123.1(=\mathrm{CH}), 42.7\left(1-\mathrm{CH}_{2}\right), 27.2,21.3\left(\right.$ both $\left.\mathrm{CH}_{2}\right), 20.4(\mathrm{Ar}-\mathrm{Me})$ and $14.0(6-\mathrm{Me}) ; m / z(\mathrm{FAB}) 254\left(\mathrm{M}^{+}+1,100 \%\right), 184(33), 154$ (61), 137 (53), 107 (18), 91 (23) and 77 (15) [Found: $\mathrm{M}^{+}+1$, 254.1240. $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{NO}_{2} \mathrm{~S}$ requires $\left.M, 254.1215\right]$.
( $Z$ )-1-(Methylsulfonylamino)hex-3-ene 26b. ( $Z$ )-Boc-Sulfonamide $\mathbf{2 5 b}(0.21 \mathrm{~g}, 0.80 \mathrm{mmol})$ was deprotected as described above and purified by cc (dichloromethane) to yield the title compound $\mathbf{2 6 b}(0.16 \mathrm{~g}, 71 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.38$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3287(\mathrm{NH}), 2964$ (s), 2875 (s), $1655(\mathrm{w}), 1437(\mathrm{~s})$ and $1317(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 5.49(1 \mathrm{H}, \mathrm{dt}, J 10.7$ and $7.3,4-\mathrm{H}), 5.24(1 \mathrm{H}, \mathrm{dt}, J 10.7$ and $7.3,3-\mathrm{H}), 4.92-4.79(1 \mathrm{H}, \mathrm{br}$, $\mathrm{s}, \mathrm{NH}), 3.08\left(2 \mathrm{H}, \mathrm{td}, J 6.9\right.$ and $\left.6.4,1-\mathrm{CH}_{2}\right), 2.91\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{Me}\right)$, $2.27\left(2 \mathrm{H}, \mathrm{dt}, J 7.3\right.$ and $\left.6.9,2-\mathrm{CH}_{2}\right), 2.01(2 \mathrm{H}, \mathrm{qd}, J 7.5$ and 7.3 , $\left.5-\mathrm{CH}_{2}\right)$ and $0.93(3 \mathrm{H}, \mathrm{t}, J 7.5,6-\mathrm{Me}) ; \delta_{\mathrm{C}}(100) 135.0,124.0$ (both $=\mathrm{CH}), 42.8\left(1-\mathrm{CH}_{2}\right), 39.9\left(\mathrm{SO}_{2} \mathrm{Me}\right), 27.7,20.5\left(\right.$ both $\left.\mathrm{CH}_{2}\right)$ and $14.0(\mathrm{Me}) ; m / z(\mathrm{FAB}) 178\left(\mathrm{M}^{+}+1,43 \%\right), 154$ (100), 137 (81), 108 (27), 89 (20), 77 (22) and 55 (18).
( $Z$ )-3-(Tolylsulfonylamino)dec-5-ene 26c. ( $Z$ )-Dec-5-en-3-ol $(0.40 \mathrm{~g}, 0.26 \mathrm{mmol})$ was reacted with Mitsunobu reagent 17b $(0.69 \mathrm{~g}, 0.26 \mathrm{mmol})$ under the conditions described above to afford $N$-Boc-sulfonamide $\mathbf{2 5 c}$ as a yellow, semi-solid which was not further purified but which was immediately deprotected.

Crude ( $Z$ )-Boc-sulfonamide 25c was treated with trifluoroacetic acid ( $2.9 \mathrm{ml}, 26.0 \mathrm{mmol}$ ) as described above and the crude product purified by cc (dichloromethane) to afford the title compound 26c $(0.05 \mathrm{~g}, 74 \%)$ as a viscous, colourless oil: $R_{\mathrm{f}}$ 0.38 (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3274(\mathrm{NH}), 2964$ (s), 2938 (s), $2862(\mathrm{w}), 1601(\mathrm{w})$ and $1500(\mathrm{~s}) ; \delta_{\mathrm{H}}(270) 7.83(2 \mathrm{H}, \mathrm{d}$, $J 7.9, \mathrm{Ar}-\mathrm{H}), 7.33(2 \mathrm{H}, \mathrm{d}, J 7.9, \mathrm{Ar}-\mathrm{H}), 5.47-5.19(3 \mathrm{H}, \mathrm{m}, \mathrm{NH}$, 5- and $6-\mathrm{H}), 3.25-3.17(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.45(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me})$, 2.21-2.16 ( $2 \mathrm{H}, \mathrm{m}, 4-\mathrm{CH}_{2}$ ), 2.07-1.97 ( $2 \mathrm{H}, \mathrm{m}, 7-\mathrm{CH}_{2}$ ), 1.59$1.29\left(6 \mathrm{H}, \mathrm{m}, 2-, 8-\mathrm{and} 9-\mathrm{CH}_{2}\right), 0.91(3 \mathrm{H}, \mathrm{t}, J 6.9,1-\mathrm{Me})$ and $0.84(3 \mathrm{H}, \mathrm{t}, J 7.4,10-\mathrm{Me})$; $\delta_{\mathrm{C}}(67.8) 143.2$, 138.6 (both Ar-C), $133.4(=\mathrm{CH}), 129.7,127.3$ (both Ar-CH), 124.3 $(=\mathrm{CH}), 55.6(3-\mathrm{CH}), 32.4,31.9,27.4,27.2,22.5\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 21.7$ (Ar-Me), $14.2(10-\mathrm{Me})$ and $9.9(1-\mathrm{Me}) ; m / z(\mathrm{FAB}) 310\left(\mathrm{M}^{+}+1\right.$, $53 \%$ ), 212 (100), 155 (41), 137 (19), 91 (36), 83 (20) and 54 (16) [Found: $\mathrm{M}^{+}+1,310.1846 . \mathrm{C}_{17} \mathrm{H}_{28} \mathrm{NO}_{2} \mathrm{~S}$ requires $M$, 310.1840].
(2RS,3RS)-(E)-2-(p-Tolylsulfonylamino)-3-methyloct-4-ene 27b. ( $2 S R, 3 R S$ )-(E)-3-Methyloct-4-en-2-ol [(E)-syn-alkenol] ( $0.80 \mathrm{~g}, 5.62 \mathrm{mmol}$ ), derived from cis-2,3-epoxybutane, was reacted with Mitsunobu reagent $\mathbf{1 7 b}(1.53 \mathrm{~g}, 5.62 \mathrm{mmol})$ under the conditions described in the general procedure to afford the (E)-anti-Boc-sulfonamide $27 \mathbf{a}$ as an orange semi-solid which was not further purified but which was immediately deprotected.

A solution of crude ( $E$ )-anti-Boc-sulfonamide 27a in dichloromethane ( 50 ml ) was deprotected as described above using trifluoroacetic acid $(5.7 \mathrm{~g}, 56 \mathrm{mmol})$. The product was purified by cc (dichloromethane) to afford the sulfonamide 27b $(1.32 \mathrm{~g}$, $80 \%$ ) as a colourless, viscous oil: $R_{\mathrm{f}} 0.48$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3282(\mathrm{NH}), 2973$ (s), 2931 (s), 2875 (w), 1581 (w), 1454 (s), 1328 (s) and $1166(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.74(2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $7.29(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 5.37(1 \mathrm{H}, \mathrm{dt}, J 15.2$ and 6.6 , $5-\mathrm{H}), 5.12(1 \mathrm{H}$, ddd, $J 15.2,8.3$ and 1.2, $4-\mathrm{H}), 4.34(1 \mathrm{H}, \mathrm{d}, J 8.7$, NH), 3.29-3.20 (1H, m, 2-H), 2.43 (3H, s, Ar-Me), 2.16-2.07 $(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{CH}), 1.93\left(2 \mathrm{H}, \mathrm{td}, J 7.1\right.$ and $\left.6.6,6-\mathrm{CH}_{2}\right), 1.35(2 \mathrm{H}, \mathrm{qt}$, $J 7.4$ and $\left.7.1,7-\mathrm{CH}_{2}\right), 0.97(3 \mathrm{H}, \mathrm{d}, J 6.7,1-\mathrm{Me}), 0.90(3 \mathrm{H}, \mathrm{d}$, $J 6.8,3-\mathrm{Me})$ and $0.86(3 \mathrm{H}, \mathrm{t}, J 7.4,8-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 143.6,139.8$ (both Ar-C), 133.1 (Ar-CH), 130.1, 129.6 (both $=\mathrm{CH}), 127.6$ (Ar-CH), $53.6(2-\mathrm{CH}), 42.2(3-\mathrm{CH}), 34.6,22.5$ (both $\left.\mathrm{CH}_{2}\right), 21.5$ (Ar-Me), 17.4 ( $1-\mathrm{Me}$ ), 17.1 (3-Me) and 13.6 ( $8-\mathrm{Me}$ ); m/z (FAB) $296\left(\mathrm{M}^{+}+1,93 \%\right), 198$ (100), 155 (48), 125 (84), 107 (12), 91 (47), 83 (41) and 69 (78) [Found: C, 64.6; H, 8.6; N, 4.55;
$\mathrm{M}^{+}+1$, 296.1734. $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{NO}_{2} \mathrm{~S}$ requires $\mathrm{C}, 65.0 ; \mathrm{H}, 8.5 ; \mathrm{N}$, $4.7 \% ; \mathrm{C}_{16} \mathrm{H}_{26} \mathrm{NO}_{2} \mathrm{~S}$ requires $\left.M, 296.1684\right]$.
(2SR,3RS)-( $E$ )-2-( $p$-Tolylsulfonylamino)-3-methyloct-4-ene 28b. $(2 S R, 3 S R)$-( $E$ )-3-Methyloct-4-en-2-ol [(E)-anti-alkenol] $(0.30 \mathrm{~g}, 2.20 \mathrm{mmol})$, derived from trans-2,3-epoxybutane, was reacted with Mitsunobu reagent $\mathbf{1 7 b}(0.57 \mathrm{~g}, 2.20 \mathrm{mmol})$ under the conditions described in the general procedure to give the ( $E$ )-syn-Boc-sulfonamide 28a as an orange semi-solid which was not further purified but which was immediately deprotected.

A solution of crude ( $E$ )-syn-Boc-sulfonamide 28a in dichloromethane ( 20 ml ) was deprotected using trifluoroacetic acid ( $2.4 \mathrm{ml}, 21 \mathrm{mmol}$ ) as described above and the product purified by cc (dichloromethane) to give the title compound 28b ( $0.53 \mathrm{~g}, 86 \%$ yield) as a colourless, viscous oil: $R_{\mathrm{f}} 0.49$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 3282(\mathrm{NH}), 2973$ (s), 2931 (s), 2882 (s), $1588(\mathrm{w}), 1440(\mathrm{~s})$ and $1328(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 7.76(2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $7.29(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 5.40(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and 6.8 , $5-\mathrm{H}), 5.74(1 \mathrm{H}$, ddd, $J 15.3,7.9$ and $1.2,4-\mathrm{H}), 4.76(1 \mathrm{H}, \mathrm{d}, J 7.3$, NH), 3.14 ( $1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ ), 2.42 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), $2.10(1 \mathrm{H}, \mathrm{m}$, $3-\mathrm{H}), 1.90\left(2 \mathrm{H}, \mathrm{td}, J 7.3\right.$ and $\left.6.8,6-\mathrm{CH}_{2}\right), 1.32(2 \mathrm{H}, \mathrm{tq}, J 7.3$ and $\left.7.3,7-\mathrm{CH}_{2}\right), 0.99(3 \mathrm{H}, \mathrm{d}, J 6.6,1-\mathrm{Me}), 0.90(3 \mathrm{H}, \mathrm{d}, J 6.8,3-\mathrm{Me})$ and $0.86(3 \mathrm{H}, \mathrm{t}, J 7.3,8-\mathrm{Me}) ; \delta_{\mathrm{C}}(100)$ 143.0, 138.0 (both Ar-C), 132.4 (Ar-CH), 130.8, 129.4 (both $=\mathrm{CH}$ ), 127.0 (Ar-CH), 53.9 (2-CH), 42.1 (3-CH), 34.5, 22.4 (both $\mathrm{CH}_{2}$ ), 21.4 (Ar-Me), 18.8 $(1-\mathrm{Me}), 16.2(3-\mathrm{Me})$ and $13.5(8-\mathrm{Me}) ; m / z(\mathrm{FAB}) 296\left(\mathrm{M}^{+}+1\right.$, 46\%), 198 (50), 154 (57), 137 (48), 125 (43), 107 (27), 91 (72), 83 (32) and 69 (100) [Found: $\mathrm{M}^{+}+1$, 296.1697. $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 296.1684]$.

## (1RS,2RS)-2-(Hex-1-ynyl)-1-(p-tolylsulfonylamino)cyclo-

hexane 30b. To a stirred solution of phthalimide $\mathbf{3 0 a}(1.0 \mathrm{~g}, 3.5$ mmol ) in ethanol ( 4 ml ) was added hydrazine monohydrate $(3.85 \mathrm{ml})$. The resulting solution was heated at reflux for 2 h , then cooled and acidified with concentrated hydrochloric acid. The phthalhydrazide was removed by filtration and the solid washed with ethanol ( $4 \times 10 \mathrm{ml}$ ). The filtrate was diluted with water ( 50 ml ), and the solution reduced to 20 ml under reduced pressure and filtered. The solvent was then completely removed and the remaining amine salt dissolved in dichloromethane $(10 \mathrm{ml})$. The resulting solution was treated with toluene-psulfonyl chloride ( $0.74 \mathrm{~g}, 3.85 \mathrm{mmol}$ ) and triethylamine ( 1.1 ml , 7.7 mmol ) and stirred at ambient temperature for 16 h then washed successively with 2 M hydrochloric acid ( $2 \times 5 \mathrm{ml}$ ), saturated aqueous sodium hydrogencarbonate ( 5 ml ) and brine $(10 \mathrm{ml})$. The remaining organic solution was dried and evaporated and the crude product purified by cc (hexane-ether $80: 20$ ) to afford the toluene-p-sulfonamide $\mathbf{3 0 b}(0.55 \mathrm{~g}, 47 \%)$ as a pale yellow oil: $v_{\text {max }} / \mathrm{cm}^{-1} 3372(\mathrm{NH}), 1738$ (w), 1599 (w), 1415 (s) and $1331(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 7.77(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 7.27(2 \mathrm{H}, \mathrm{d}$, $J 8.2, \mathrm{Ar}-\mathrm{H}), 4.89(1 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{NH}), 3.20-3.17(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H})$, $2.60(1 \mathrm{H}, \mathrm{app}$ br s, 2-H), $2.41(3 \mathrm{H}, \mathrm{s}$, Ar-Me), 2.15 ( $2 \mathrm{H}, \mathrm{td}, J 6.9$ and $\left.2.2,3^{\prime}-\mathrm{CH}_{2}\right), 1.80-1.25(12 \mathrm{H}, \mathrm{m})$ and $0.91(3 \mathrm{H}, \mathrm{t}, J 7.1$, Me ); $\delta_{\mathrm{C}}(100) 143.0,138.5$ (both Ar-C), 129.5, 126.7 (both ArCH), 84.9, 77.8 (both C), 55.6, 33.7 (both CH), 31.0, 30.4, 29.7, 24.6, $21.9\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 21.4(\mathrm{Ar}-\mathrm{Me}), 20.6,18.2\left(\right.$ both $\left.\mathrm{CH}_{2}\right)$ and $13.6(\mathrm{Me}) ; \mathrm{m} / \mathrm{z}(\mathrm{ES}) 334\left(\mathrm{M}^{+}+1,100 \%\right), 189$ (42) and 124 (19) [Found: $\mathrm{M}^{+}+1,334.1841 . \mathrm{C}_{19} \mathrm{H}_{28} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 334.1843$ ].

## ( $E$ )-( $1 R S, 2 R S$ )-2-(Hex-1-en-1-yl)-1-( $p$-tolylsulfonylamino)-

 cyclohexane 31. To a stirred solution of sulfonamide 30b ( 0.10 $\mathrm{g}, 0.30 \mathrm{mmol}$ ) in THF ( 10 ml ) was added lithium aluminium hydride ( 1 M solution in THF, $3.0 \mathrm{ml}, 3.0 \mathrm{mmol}$ ) and the resulting solution refluxed for 48 h . The solution was then cooled and excess reagent was quenched by the dropwise addition of water. The resulting white solid was removed by filtration. The filtrate was evaporated, and the residue dissolved in dichloromethane ( 10 ml ) and treated with triethylamine ( $50 \mu \mathrm{l}, 0.30 \mathrm{mmol}$ ) and toluene-p-sulfonyl chloride ( $78 \mathrm{mg}, 0.30 \mathrm{mmol}$ ). The resultingsolution was stirred at ambient temperature for 2 days then diluted with dichloromethane and washed successively with 2 M hydrochloric acid ( $3 \times 2 \mathrm{ml}$ ), saturated aqueous sodium hydrogencarbonate ( 3 ml ) and brine ( 3 ml ). The organic solution was then dried and evaporated and the residue purified by cc (hexane-ether $80: 20$ ) to furnish the ( $E$ )-toluene-p-sulfonamide $31(40 \mathrm{mg}, 41 \%)$ as a viscous oil: $v_{\text {max }} / \mathrm{cm}^{-1} 1600(\mathrm{~s}), 1415(\mathrm{~s})$ and 1178 (s); $\delta_{\mathrm{H}}(400) 7.73(2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{Ar}-\mathrm{H}), 7.29(2 \mathrm{H}, \mathrm{d}, J 8.0$, Ar-H), $5.32\left(1 \mathrm{H}, \mathrm{dt}, J 15.5\right.$ and $\left.6.5,2^{\prime}-\mathrm{H}\right), 5.23(1 \mathrm{H}, \mathrm{dd}, J 15.5$ and $\left.7.5,1^{\prime}-\mathrm{H}\right), 4.40(1 \mathrm{H}, \mathrm{d}, J 7.3, \mathrm{NH}), 3.30-3.26(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H})$, 2.43 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), $2.20-2.13(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 1.95-1.90(2 \mathrm{H}, \mathrm{m}$, $\left.3^{\prime}-\mathrm{CH}_{2}\right), 1.63-1.20(12 \mathrm{H}, \mathrm{m})$ and $0.91(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me}) ; \delta_{\mathrm{C}}(100)$ 142.7, 137.9 (both Ar-C), 133.6 (=CH), 129.3, 128.1 (both Ar-CH), 126.8 (=CH), 53.5, 42.2 (both CH), 32.2, 31.3, 30.2, 28.4, $22.6\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 22.0(\mathrm{Ar}-\mathrm{Me}), 21.1\left(\mathrm{CH}_{2}\right)$ and $13.7(\mathrm{Me})$.
(Z)-(1RS,2RS)-2-(Hex-1-en-1-yl)-1-(p-tolylsulfonylamino)cyclohexane 32. The tosylated acetylene $\mathbf{3 0 b}$ ( $54 \mathrm{mg}, 0.16 \mathrm{mmol}$ ) was dissolved in ethyl acetate ( 1 ml ) and $5 \%$ palladium on barium sulfate ( $7 \mathrm{mg}, 0.016 \mathrm{mmol}$ ) was added. The resulting suspension was stirred under an atmosphere of hydrogen for 3 h . The catalyst was removed by filtration and the solvent evaporated to give a crude product which was purified by cc (hexane-ether $80: 20$ ) to afford the ( $Z$ )-toluene-p-sulfonamide $32(50 \mathrm{mg}, 95 \%)$ as a pale yellow oil: $v_{\text {max }} / \mathrm{cm}^{-1} 1721(\mathrm{~m}), 1600$ (s), 1415 (s) and 1178 (s); $\delta_{\mathrm{H}}(400) 7.74$ ( $2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}$ ), 7.28 $(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 5.48-5.38\left(2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\right.$ and $\left.2^{\prime}-\mathrm{H}\right), 4.49(1 \mathrm{H}$, d, $J 6.6, \mathrm{NH}), 3.35-3.30(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 1-\mathrm{H}), 2.61-2.58(1 \mathrm{H}, \mathrm{m}$, $2-\mathrm{H}), 2.43$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), $1.88-1.85$ ( $2 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{CH}$ ), $1.62-1.11$ ( $12 \mathrm{H}, \mathrm{m}$ ) and 0.88 ( $3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me}$ ); $\delta_{\mathrm{C}}(100)$ 142.9, 137.9 (both Ar-C), $133.8(=\mathrm{CH}), 129.5,126.9$ (both Ar-CH), $126.7(=\mathrm{CH})$, 54.2, 37.1 (both CH), 31.8, 30.3, 30.0, 27.2, 23.8, 22.3 (all $\mathrm{CH}_{2}$ ), 21.5 (Ar-Me), $21.1\left(\mathrm{CH}_{2}\right)$ and $13.9(\mathrm{Me})$.

2-Methyloct-4-enoic acid 34. To a stirred solution of hex-1-en-3-ol ( $5.00 \mathrm{~g}, 50.0 \mathrm{mmol}$ ), triethylamine ( $6.05 \mathrm{ml}, 50.0 \mathrm{mmol}$ ) and 4-dimethylaminopyridine ( 10 mg ) in dichloromethane at $0^{\circ} \mathrm{C}$ was added dropwise propionyl chloride $(4.77 \mathrm{ml}, 55.0$ $\mathrm{mmol})$. The reaction mixture was heated to reflux for 2 h , then cooled and washed with aqueous 2 M hydrochloric acid $(2 \times 30$ $\mathrm{ml})$. The organic solution was dried and evaporated to leave a yellow oil which was purified by cc (dichloromethane) to afford the propionate $33(7.20 \mathrm{~g}, 85 \%)$ as a colourless oil: $R_{\mathrm{f}} 0.76$ (dichloromethane): $v_{\max } / \mathrm{cm}^{-1} 2961$ (s), $2940(\mathrm{~s}), 2876$ (w), 1739 (C=O), $1463(\mathrm{~s})$ and $1362(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 5.76(1 \mathrm{H}$, ddd, $J 16.0,9.4$ and $6.2,2-\mathrm{H}), 5.29-5.11\left(3 \mathrm{H}, \mathrm{m}, 1-\mathrm{CH}_{2}\right.$ and $\left.3-\mathrm{H}\right), 2.32(2 \mathrm{H}, \mathrm{q}$, $\left.J 7.6,3^{\prime}-\mathrm{CH}_{2}\right), 1.64-1.17\left(4 \mathrm{H}, \mathrm{m}, 4-\mathrm{CH}_{2}\right.$ and $\left.5-\mathrm{CH}_{2}\right), 1.14(3 \mathrm{H}$, $\mathrm{t}, J 7.6,4-\mathrm{Me})$ and $0.91(3 \mathrm{H}, \mathrm{t}, J 7.3,6-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 173.7$ $(\mathrm{C}=\mathrm{O}), 136.7(2-\mathrm{CH}), 116.7\left(1-\mathrm{CH}_{2}\right), 74.3(3-\mathrm{CH}), 36.3,27.8$, 18.3 (all $\left.\mathrm{CH}_{2}\right), 13.8(4-\mathrm{Me})$ and 9.1 (6-Me).

To neat diisopropylamine ( $2.31 \mathrm{ml}, 16.7 \mathrm{mmol}$ ), stirred at $-78^{\circ} \mathrm{C}$, was added dropwise butyllithium ( 1.6 M solution in hexanes, $5.10 \mathrm{ml}, 11.1 \mathrm{mmol}$ ) and the mixture warmed to ambient temperature during 1 h . The hexane was removed by a stream of nitrogen to leave a cream paste which was dissolved in THF ( 25 ml ). The resulting solution of LDA was cooled to $-78^{\circ} \mathrm{C}$ and unsaturated ester $33(1.56 \mathrm{~g}, 10.0 \mathrm{mmol})$ was added dropwise. The reaction mixture was maintained at $-78^{\circ} \mathrm{C}$ for 10 minutes, after which trimethylsilyl chloride ( $1.20 \mathrm{~g}, 11.1$ mmol , freshly distilled from calcium hydride) was added dropwise. The reaction mixture was warmed to ambient temperature during 0.5 h then refluxed for 1 h . The reaction mixture was cooled to ambient temperature, poured into aqueous 2 M sodium hydroxide ( 30 ml ) and the mixture washed with diethyl ether $(3 \times 20 \mathrm{ml})$. The aqueous layer was acidified with concentrated hydrochloric acid, then extracted with dichloromethane $(3 \times 30 \mathrm{ml})$. The combined dichloromethane extracts were dried and evaporated to give the acid $34(1.03 \mathrm{~g}, 66 \%)$ as a clear oil: $R_{\mathrm{f}} 0.47$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 3112(\mathrm{OH}), 2959$ (s), 2663 $(\mathrm{s}), 1706(\mathrm{C}=\mathrm{O}), 1463(\mathrm{~s})$ and $1418(\mathrm{~s}) ; \delta_{\mathrm{H}}(270) 5.56-5.33(2 \mathrm{H}$,
$\mathrm{m}, 4-\mathrm{and} 5-\mathrm{H}), 2.59-2.34\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{CH}_{2}\right), 2.21-1.87(3 \mathrm{H}, \mathrm{m}$, $2-\mathrm{H}$ and $6-\mathrm{CH}_{2}$ ), 1.46-1.26 ( $2 \mathrm{H}, \mathrm{m}, 7-\mathrm{CH}_{2}$ ), $1.18(3 \mathrm{H}, \mathrm{d}, J 7.0$, $\left.1^{\prime}-\mathrm{Me}\right)$ and $0.90(3 \mathrm{H}, \mathrm{t}, J 7.3,8-\mathrm{Me})$; $\delta_{\mathrm{C}}(67.8) 182.8(\mathrm{C}=\mathrm{O})$, 133.2, 126.5 (both $=\mathrm{CH}), 67.8(2-\mathrm{CH}), 36.4,34.6,22.5\left(\mathrm{all} \mathrm{CH}_{2}\right)$, $16.1(1-\mathrm{Me})$ and $13.5(8-\mathrm{Me}){ }^{53}$
(E)-2-(tert-Butoxycarbonylamino)oct-4-ene 35a. A stirred solution of the foregoing acid $34(2.60 \mathrm{~g}, 17.0 \mathrm{mmol})$, diphenylphosphoryl azide ( $3.59 \mathrm{ml}, 17.0 \mathrm{ml}, 17.0 \mathrm{mmol}$ ) and triethylamine ( $2.44 \mathrm{ml}, 17.0 \mathrm{mmol}$ ) in tert-butyl alcohol ( 50 ml ) was refluxed for 18 h . The reaction mixture was cooled, and the tertbutyl alcohol removed under reduced pressure. The residue was dissolved in benzene ( 50 ml ) and the solution washed successively with water ( 30 ml ), saturated aqueous sodium carbonate $(60 \mathrm{ml})$ and brine $(30 \mathrm{ml})$. The organic solution was then dried and evaporated to leave a clear oil which was purified by cc (dichloromethane) to afford the $N$-Boc amine $\mathbf{3 5 a}(2.79 \mathrm{~g}, 72 \%)$ as a clear oil: $R_{\mathrm{f}} 0.44$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 3342(\mathrm{NH})$, 2964 (s), 2872 (s), 1693 (C=O), 1523 (s) and 1248 ( s$) ; \delta_{\mathrm{H}}(250)$ $5.42(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $6.6,5-\mathrm{H}), 5.37(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and 6.6 , $4-\mathrm{H}), 4.42-4.28(1 \mathrm{H}, \mathrm{m}, \mathrm{NH}), 3.72-3.57(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.10(1 \mathrm{H}$, $\mathrm{dd}, J 6.6$ and $6.0,3-\mathrm{H}), 1.96\left(2 \mathrm{H}, \mathrm{td}, J 7.2\right.$ and $\left.6.6,6-\mathrm{CH}_{2}\right), 1.42$ $\left(9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}} \mathrm{Bu}\right), 1.36\left(2 \mathrm{H}, \mathrm{qt}, J 7.4\right.$ and $\left.7.2,7-\mathrm{CH}_{2}\right), 1.07(3 \mathrm{H}, \mathrm{t}$, $J 6.6,1-\mathrm{Me})$ and $0.86(3 \mathrm{H}, \mathrm{t}, J 7.4,8-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 133.8$, 125.7 (both $=\mathrm{CH}), 47.4(2-\mathrm{CH}), 40.0,34.7,22.5\left(\right.$ all CH$\left._{2}\right), 20.5$ ( $\left.1^{\prime}-\mathrm{Me}\right)$ and 13.6 ( $8-\mathrm{Me}$ ); m/z (FAB) $228\left(\mathrm{M}^{+}+1,15 \%\right), 172$ (100), 154 (10), 144 (29), 88 (33), 68 (27) and 56 (46) [Found: $\mathrm{M}^{+}+1,228.1959 . \mathrm{C}_{13} \mathrm{H}_{26} \mathrm{NO}_{2}$ requires $\left.M, 228.1963\right]$.
( $\boldsymbol{E}$ )-2-Aminooct-4-ene 35b. To a stirred solution of the $N$-Boc-amine 35 a ( $0.40 \mathrm{~g}, 1.80 \mathrm{mmol}$ ) in dichloromethane $(20 \mathrm{ml})$ was added trifluoroacetic acid $(2.05 \mathrm{~g}, 18 \mathrm{mmol})$ and the solution stirred at ambient temperature for 3 h , then neutralised with saturated aqueous sodium hydrogencarbonate, followed by extraction with dichloromethane ( $3 \times 30 \mathrm{ml}$ ). The combined organic solutions were dried (sodium sulfate) and evaporated to afford the amine $\mathbf{3 5 b}(0.19 \mathrm{~g}, 86 \%)$ as a yellow oil: $R_{\mathrm{f}} 0.18$ (methanol-dichloromethane 1:9); $v_{\text {max }} / \mathrm{cm}^{-1} 3370(\mathrm{NH}), 2945$ (s), $2837(\mathrm{~s}), 1460(\mathrm{~s})$ and $1342(\mathrm{~s}) ; \delta_{\mathrm{H}}(270) 5.58(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and $6.7,=\mathrm{CH}), 5.44(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and $7.1,=\mathrm{CH}), 3.77-3.65$ $(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.08-1.83\left(4 \mathrm{H}, \mathrm{m}, 3-\mathrm{and} 6-\mathrm{CH}_{2}\right), 1.38-1.23(2 \mathrm{H}$, $\left.\mathrm{m}, 7-\mathrm{CH}_{2}\right), 1.05(3 \mathrm{H}, \mathrm{d}, J 6.4,1-\mathrm{Me})$ and $0.81(3 \mathrm{H}, \mathrm{t}, J 7.3$, $8-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 138.2,122.8$ (both $=\mathrm{CH}$ ), $49.9(2-\mathrm{CH}), 38.3$, 35.2, $22.7\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 18.9(1-\mathrm{Me})$ and $14.2(8-\mathrm{Me}) ; \mathrm{m} / \mathrm{z}(\mathrm{FAB})$ $128\left(\mathrm{M}^{+}+1,23 \%\right), 73$ (10), 69 (17) and 43 (100) [Found: $\mathrm{M}^{+}+1,128.1565 . \mathrm{C}_{8} \mathrm{H}_{18} \mathrm{~N}$ requires $\left.M, 128.1557\right]$.
( $E$ )-2-Trifluoroacetylaminooct-4-ene 35c. A solution of amine $\mathbf{3 5 b}$ ( $0.28 \mathrm{~g}, 2.20 \mathrm{mmol}$ ), trifluoroacetic anhydride ( 0.24 $\mathrm{ml}, 2.20 \mathrm{mmol}$ ) and triethylamine ( $0.34 \mathrm{ml}, 2.20 \mathrm{mmol}$ ) in dichloromethane ( 20 ml ) was stirred at ambient temperature for 2 h . The reaction mixture was then neutralised with 2 M hydrochloric acid and extracted with dichloromethane ( $3 \times 20 \mathrm{ml}$ ). The combined organic solutions were dried and evaporated to give a mixture of mono- and diacylated products which were separated by cc (dichloromethane) to afford the title compound $35 \mathrm{c}(0.31 \mathrm{~g}, 63 \%)$ as a pale yellow oil: $R_{\mathrm{f}} 0.37$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 3621(\mathrm{NH}), 3021(\mathrm{~s}), 2975(\mathrm{~s}), 1723$ (C=O), 1527 (s) and $1383(\mathrm{~s}) ; \delta_{\mathrm{H}}(270) 5.49(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and $6.8,5-\mathrm{H}), 5.35$ $(1 \mathrm{H}, \mathrm{dt}, J 15.3$ and $7.1,4-\mathrm{H}), 4.12-4.01(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.21(2 \mathrm{H}$, dd, $J 7.1$ and $\left.6.9,3-\mathrm{CH}_{2}\right), 1.97\left(2 \mathrm{H}, \mathrm{td}, J 7.2\right.$ and $\left.6.8,6-\mathrm{CH}_{2}\right)$, $1.40-1.28\left(2 \mathrm{H}, \mathrm{m}, 7-\mathrm{CH}_{2}\right), 1.20(3 \mathrm{H}, \mathrm{d}, J 6.6,1-\mathrm{Me})$ and 0.87 $(3 \mathrm{H}, \mathrm{t}, J 7.2,8-\mathrm{Me}) ; m / z(\mathrm{FAB}) 224\left(\mathrm{M}^{+}+1,9 \%\right), 209(12), 197$ (34), 135 (100), 109 (59), 83 (29) and 69 (94) [Found: $\mathrm{M}^{+}+1$, 224.1272. $\mathrm{C}_{10} \mathrm{H}_{17} \mathrm{~F}_{3} \mathrm{NO}$ requires $M, 224.1262$ ].
( E)-2-Acetylaminooct-4-ene 35d. A solution of amine 35b $(0.10 \mathrm{~g}, 0.78 \mathrm{mmol})$, acetic anhydride $(0.07 \mathrm{ml}, 0.78 \mathrm{mmol})$, triethylamine ( $0.11 \mathrm{ml}, 0.78 \mathrm{mmol}$ ) and 4-dimethylaminopyridine ( 10 mg ) in dichloromethane ( 10 ml ) was stirred at
ambient temperature for 18 h . The reaction mixture was then neutralised with 2 M hydrochloric acid and extracted with dichloromethane ( $3 \times 10 \mathrm{ml}$ ). The combined organic solutions were dried and evaporated to afford the amide $\mathbf{3 5 d}(0.92 \mathrm{~g}, 70 \%)$ as a pale yellow oil: $R_{\mathrm{f}} 0.52$ (methanol-dichloromethane $1: 9$ ); $v_{\max } / \mathrm{cm}^{-1} 3454(\mathrm{NH}), 2956(\mathrm{~s}), 2932(\mathrm{~s}), 1682(\mathrm{C}=\mathrm{O}), 1430(\mathrm{~s})$ and $1351(\mathrm{~s}) ; \delta_{\mathrm{H}}(270) 5.39(1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $6.6,5-\mathrm{H}), 5.23$ ( $1 \mathrm{H}, \mathrm{dt}, J 15.5$ and $6.3,4-\mathrm{H}$ ), 3.83-3.99 ( $1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ ), $2.04(2 \mathrm{H}$, dd, $J 6.6$ and $\left.6.3,3-\mathrm{CH}_{2}\right), 1.91\left(2 \mathrm{H}, \mathrm{td}, J 7.2\right.$ and $\left.6.6,6-\mathrm{CH}_{2}\right)$, $1.85(3 \mathrm{H}, \mathrm{s}, \mathrm{C}(\mathrm{O}) \mathrm{Me}), 1.27\left(2 \mathrm{H}\right.$, hex, $\left.J 7.2,7-\mathrm{CH}_{2}\right), 1.03(2 \mathrm{H}, \mathrm{d}$, $J 6.6,1-\mathrm{Me})$ and $0.79(3 \mathrm{H}, \mathrm{t}, J 7.2,8-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 169.8$ $(\mathrm{C}(\mathrm{O}) \mathrm{Me}), 134.1,125.5$ (both $=\mathrm{CH}$ ), 44.9 (2-CH), 39.5, 34.7 (both $\left.\mathrm{CH}_{2}\right), 24.3(\mathrm{C}(\mathrm{O}) \mathrm{Me}), 22.6\left(\mathrm{CH}_{2}\right), 20.2(1-\mathrm{Me})$ and 13.6 ( $8-\mathrm{Me}$ ); $m / z$ (FAB) $170\left(\mathrm{M}^{+}+1,24 \%\right), 155$ (30), 154 (100), 136 (70), 107 (25) and 77 (21).
( $E$ )-2-Methoxycarbonylaminooct-4-ene 35e. A solution of amine $\mathbf{3 5 b}$ ( $0.10 \mathrm{~g}, 0.78 \mathrm{mmol}$ ), methyl chloroformate ( 0.06 ml , 0.86 mmol , freshly distilled from calcium hydride) and triethylamine ( $0.12 \mathrm{ml}, 0.86 \mathrm{mmol}$ ) in dichloromethane ( 20 ml ) was stirred at ambient temperature for 5 h then neutralised with 2 M hydrochloric acid and extracted with dichloromethane ( $3 \times 20$ ml ). The combined organic solutions were dried and evaporated to give a mixture of mono- and diacylated products which were separated by cc (methanol-dichloromethane $5: 95$ ) to afford the carbamate $35 \mathrm{e}(0.08 \mathrm{~g}, 53 \%)$ as a pale yellow oil: $R_{\mathrm{f}} 0.14$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 3437$ (NH), 2961 (s), 2931 (s), $1713(\mathrm{C}=\mathrm{O}), 1514(\mathrm{~s})$ and $1435(\mathrm{~s}) ; \delta_{\mathrm{H}}(270) 5.55-5.31(2 \mathrm{H}, \mathrm{m}$, $4-$ and $5-\mathrm{H}), 3.92-3.79(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 3.62(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.11$ $\left(2 \mathrm{H}, \mathrm{dd}, J 6.5\right.$ and $\left.6.2,3-\mathrm{CH}_{2}\right), 1.95(2 \mathrm{H}, \mathrm{td}, J 7.3$ and 6.6 , $\left.6-\mathrm{CH}_{2}\right), 1.35\left(2 \mathrm{H}\right.$, hex, $\left.J 7.3,7-\mathrm{CH}_{2}\right), 1.15(3 \mathrm{H}, \mathrm{d}, J 6.6,1-\mathrm{Me})$ and $0.89(3 \mathrm{H}, \mathrm{t}, J 7.3,8-\mathrm{Me}) ; \delta_{\mathrm{C}}(67.8) 156.2(C(\mathrm{O}) \mathrm{OMe}), 134.6$, 124.9 (both $=\mathrm{CH})$, 54.6 (C(O)OMe), 47.0 (2-CH), 43.7, 34.6, 22.5 (all $\left.\mathrm{CH}_{2}\right), 20.1(1-\mathrm{Me})$ and 13.6 ( $8-\mathrm{Me}$ ) [Found (ES): $\mathrm{M}^{+}+1,186.1491 . \mathrm{C}_{10} \mathrm{H}_{20} \mathrm{NO}_{2}$ requires $\left.M, 186.1494\right]$.

## Iodocyclisations with sodium hydrogen carbonate as base: general procedure

To a stirred mixture of the cyclisation precursor ( $1 \mathrm{mmol}, 1$ eq.) and sodium hydrogen carbonate ( $0.25 \mathrm{~g}, 3 \mathrm{mmol}, 3$ eq.) in dry acetonitrile ( 1 ml ) was added portionwise iodine ( 0.76 g , $3 \mathrm{mmol}, 3$ eq.). The reaction mixture was stirred at ambient temperature until no starting material remained according to TLC analysis (typically $5-30 \mathrm{~min}$ ). The reaction mixture was then treated with saturated aqueous sodium thiosulfate $(1-2 \mathrm{ml})$ and the resulting pale yellow solution extracted with dichloromethane ( $3 \times 2 \mathrm{ml}$ ). The combined organic solutions were dried, filtered and the solvent evaporated to yield the crude product which was purified by column chromatography. Any isomer ratios quoted were determined from ${ }^{1} \mathrm{H}$ NMR spectra for the mixtures; the peaks used for the determination of the ratios are stated in each case.

Cyclisation of 4-(phenylsulfonylamino)pent-1-ene 24a. Homoallylic sulfonamide $\mathbf{2 4 a}(0.06 \mathrm{~g}, 0.25 \mathrm{mmol})$ was reacted under the conditions described above for 1 h . The crude product was purified by cc (dichloromethane) to afford an inseparable mixture of isomers in the ratio $3: 2$ together with a trace of another unidentified product ( $0.04 \mathrm{~g}, 42 \%$ ). The isomer ratio was determined by measurement of the alkyl methyl peaks. Data for the two predominant isomers: $R_{\mathrm{f}} 0.55$ (hexane-ethyl acetate $1: 1$ ); $\delta_{\mathrm{H}}(250) 7.79-7.91(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$, both isomers), 7.65-7.51 ( 6 H , m , Ar-H, both isomers), $3.93-3.84(1 \mathrm{H}, \mathrm{m}$, minor isomer), $3.79-3.64(3 \mathrm{H}, \mathrm{m}$, both isomers), 3.61-3.50 $(2 \mathrm{H}, \mathrm{m}$, both isomers), $3.46-3.38(1 \mathrm{H}, \mathrm{m}$, minor isomer), $3.28-3.17(2 \mathrm{H}, \mathrm{m}$, major isomer), $2.62-2.50(1 \mathrm{H}, \mathrm{m}$, minor isomer), 2.31-2.19 ( $1 \mathrm{H}, \mathrm{m}$, major isomer), $1.99-1.84(1 \mathrm{H}, \mathrm{m}$, minor isomer), $1.61-$ $1.52(1 \mathrm{H}, \mathrm{m}$, major isomer), $1.38(3 \mathrm{H}, \mathrm{d}, J 5.9$, minor isomer) and $1.32\left(3 \mathrm{H}, \mathrm{d}, J 6.3\right.$, major isomer); $\delta_{\mathrm{C}}(100) 138.4$ (Ar-C,
minor isomer), 135.8 (Ar-C, major isomer), 133.4 ( $\mathrm{Ar}-\mathrm{CH}$, minor isomer), 129.4 (Ar-CH, minor isomer), 129.3 ( $\mathrm{Ar}-\mathrm{CH}$, major isomer), 129.2 ( $\mathrm{Ar}-\mathrm{CH}$, minor isomer), 128.3 ( $\mathrm{Ar}-\mathrm{CH}$, major isomer), 127.4 ( $\mathrm{Ar}-\mathrm{CH}$, major isomer), $59.8(\mathrm{CH}$, major isomer), $58.6\left(\mathrm{CH}_{2}\right.$, minor isomer), $56.8(\mathrm{CH}$, minor isomer), $55.5\left(\mathrm{CH}\right.$, major isomer), $46.7\left(\mathrm{CH}_{2}\right.$, minor isomer $), 32.6\left(\mathrm{CH}_{2}\right.$ major isomer), 22.6 ( Me , minor isomer), 22.3 ( Me , major isomer) and $10.2\left(\mathrm{CH}_{2}\right.$, major isomer); $m / z(\mathrm{EI}) 351\left(\mathrm{M}^{+}, 2 \%\right)$, $336\left(\mathrm{M}^{+}-\mathrm{Me}, 30 \%\right), 224$ (25), 210 (96), 141 (74) and 77 (100) [ $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{INO}_{2} \mathrm{~S}$ requires $\left.M, 351\right]$.
(2SR,3RS)-2-Ethyl-3-iodo-1-( $p$-tolylsulfonyl)pyrrolidine
36b. ( $E$ )-Homoallylic tosylamide $\mathbf{2 4 b}(0.20 \mathrm{~g}, 0.79 \mathrm{mmol})$ was cyclised under the general conditions above, and the reaction was complete within 10 min . The crude product was obtained as a colourless solid, which was recrystallised (hexane-ethyl acetate $9: 1$ ) to give the iodopyrrolidine $\mathbf{3 6 b}(0.25 \mathrm{~g}, 84 \%)$ as a colourless, crystalline solid, mp $109-110^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.52$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2950(\mathrm{~s}), 2923$ (s), 2856 (w), 1592 (w), 1456 (w) and 1338 (s); $\delta_{\mathrm{H}}$ (400) 7.75 (2H, d, J 8.3, Ar-H), $7.32(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 4.14$ (1H, ddd, $J 5.1,2.1$ and 1.4, $3-\mathrm{H}), 3.88(1 \mathrm{H}$, ddd, $J 9.5,3.9$ and $1.4,2-\mathrm{H}), 3.55(1 \mathrm{H}$, ddd, $J 9.8,7.2$ and $\left.2.5,5-\mathrm{H}_{\alpha}\right), 3.42\left(1 \mathrm{H}\right.$, ddd, $J 9.8,9.8$ and $\left.6.2,5-\mathrm{H}_{\beta}\right)$, $2.41(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.25(1 \mathrm{H}$, dddd, $J 14.0,9.8,5.1$ and 2.5 , $\left.4-\mathrm{H}_{\alpha}\right), 1.95\left(1 \mathrm{H}\right.$, dddd, $14.0,7.2,6.2$ and $\left.2.1,4-\mathrm{H}_{\beta}\right), 1.83(1 \mathrm{H}$, dqd, $J 9.5,7.5$ and $\left.3.9,1^{\prime}-\mathrm{H}_{\alpha}\right), 1.46(1 \mathrm{H}$, ddq, $J 9.5,9.5$ and 7.5 , $\left.1^{\prime}-\mathrm{H}_{\beta}\right)$ and $0.96\left(3 \mathrm{H}, \mathrm{t}, J 7.5,2^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{C}}(100) 143.1,132.6$ (both Ar-C), 129.5, 127.8 (both Ar-CH), 72.9 (2-CH), $47.6\left(5-\mathrm{CH}_{2}\right)$, $35.9\left(4-\mathrm{CH}_{2}\right), 30.3\left(1^{\prime}-\mathrm{CH}_{2}\right), 23.8(\mathrm{Ar}-\mathrm{Me}), 21.4(3-\mathrm{CH})$ and 10.1 ( $2^{\prime}-\mathrm{Me}$ ); $m / z$ (ES) $380\left(\mathrm{M}^{+}+1,100 \%\right.$ ), 329 (7), 279 (8), 254 (20), 83 (12) and 60 (23) [Found: $\mathrm{M}^{+}+1,380.0181$. $\mathrm{C}_{13} \mathrm{H}_{19} \mathrm{INO}_{2} \mathrm{~S}$ requires $\left.M, 380.0187\right]$.
(2SR,3RS)-1-Phenylsulfonyl-2-ethyl-3-iodopyrrolidine 36c. ( $E$-Homoallylic sulfonamide $\mathbf{2 4 c}(0.20 \mathrm{~g}, 0.84 \mathrm{mmol})$ was cyclised under the general conditions described above and the reaction was complete within 10 min . The crude product was obtained as a colourless solid, which was recrystallised (hexane-ethyl acetate $9: 1$ ) to afford the iodopyrrolidine 36c ( $0.26 \mathrm{~g}, 84 \%$ ) as a colourless crystalline solid, mp $101-102{ }^{\circ} \mathrm{C}$ : $R_{\mathrm{f}} 0.56$ (dichloromethane); $v_{\text {max }} \mathrm{cm}^{-1} /\left(\mathrm{CHCl}_{3}\right) 2951$ (s), 2920 (s), $2856(\mathrm{w}), 1592(\mathrm{w}), 1452(\mathrm{~s})$ and $1338(\mathrm{~s}) ; \delta_{\mathrm{H}}$ (400) 7.93-7.53 $(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 4.17(1 \mathrm{H}$, ddd, $J 5.2,2.1$ and $1.4,3-\mathrm{H}), 3.90(1 \mathrm{H}$, ddd, $J 9.4,3.8$ and $1.4,2-\mathrm{H}), 3.62(1 \mathrm{H}$, ddd, $J 9.8,7.2$ and 2.5 , $\left.5-\mathrm{H}_{\alpha}\right), 3.43\left(1 \mathrm{H}\right.$, ddd, $J 9.8,9.8$ and $\left.6.2,5-\mathrm{H}_{\beta}\right), 2.27(1 \mathrm{H}$, dddd, $J 14.0,9.8,7.2$ and $\left.5.2,4-\mathrm{H}_{\alpha}\right), 1.96(1 \mathrm{H}$, dddd, 14.0, 6.2, 2.5 and $\left.2.1,4-\mathrm{H}_{\beta}\right), 1.83\left(1 \mathrm{H}, \mathrm{dqd}, J 14.0,7.5\right.$ and $\left.3.8,1^{\prime}-\mathrm{H}_{\alpha}\right), 1.46(1 \mathrm{H}$, ddq, $J 14.0,9.4$ and $\left.7.5,1^{\prime}-\mathrm{H}_{\beta}\right)$ and $0.98\left(3 \mathrm{H}, \mathrm{t}, J 7.5,2^{\prime}-\mathrm{Me}\right)$; $\delta_{\mathrm{C}}(100) 136.3$ (Ar-C), 132.9, 129.5, 127.8 (all Ar-CH), 73.1 $(2-\mathrm{CH}), 47.7\left(5-\mathrm{CH}_{2}\right), 35.9\left(4-\mathrm{CH}_{2}\right), 30.8\left(1^{\prime}-\mathrm{CH}_{2}\right), 23.8(3-\mathrm{CH})$ and $10.2\left(2^{\prime}-\mathrm{Me}\right) ; m / z$ (EI) $336\left(\mathrm{M}^{+}-\mathrm{C}_{2} \mathrm{H}_{5}, 28 \%\right)$, 208 (39), 141 (65), 128 (63), 86 (98) and 77 (100) [Found: C, 39.6; H, 4.5; $\mathrm{N}, 3.8 . \mathrm{C}_{12} \mathrm{H}_{16} \mathrm{INO}_{2} \mathrm{~S}$ requires $\left.\mathrm{C}, 39.5 ; \mathrm{H}, 4.4 ; \mathrm{N}, 3.8 \%\right]$.
(2SR,3RS,5SR)- and (2RS,3SR,5SR)-2-Butyl-5-ethyl-3-iodo1 -(p-tolylsulfonyl)pyrrolidine $\mathbf{3 6 f}$ and 37 f . ( $E$ )-Homoallylic sulfonamide $24 \mathrm{f}(0.05 \mathrm{~g}, 0.16 \mathrm{mmol})$ was cyclised under the general conditions described above; the reaction was complete after 25 min . The resulting crude oil was purified by cc (dichloromethane) to yield the iodopyrrolidine as a yellow oil $(0.05 \mathrm{~g}, 76 \%)$ in a ratio of isomers of $3: 1$. The isomer ratio was determined by measurement of the aryl-methyl peaks in the ${ }^{1} \mathrm{H}$ NMR spectrum. Comparison of the spectral data for the mixture with those for the potassium carbonate and base-free reactions (see later) revealed that the major isomer was the 2,5-trans-pyrrolidine $\mathbf{3 6 f}$ while the minor product was the 2,5-cis-pyrrolidine $\mathbf{3 7 f}$.
(2SR,3RS,5SR)- and (2RS,3SR,5SR)-5-Ethyl-3-iodo-2-phenyl-1-(p-tolylsulfonyl)pyrrolidine $\mathbf{3 6 j}$ and $\mathbf{3 7 j}$. ( $E$ )-Homo-
allylic tosylamide $\mathbf{2 4 j}(0.04 \mathrm{~g}, 0.12 \mathrm{mmol})$ was subjected to the iodocyclisation conditions described in the general procedure and the reaction was complete after 40 min . The resulting crude product was purified by cc (dichloromethane) to yield the title compounds as a colourless solid $(0.04 \mathrm{~g}, 76 \%)$ in an isomer ratio of $2: 1$. The isomer ratio was measured by comparison of the aryl-methyl peaks in the ${ }^{1} \mathrm{H}$ NMR spectrum. Comparison of the spectral data (see below) for the mixture with those for the potassium carbonate and base-free reactions showed that the major isomer was the 2,5-trans-pyrrolidine $\mathbf{3 6 j}$ and the minor isomer was the corresponding 2,5-cis-pyrrolidine $\mathbf{3 7 j}$.

Cyclisation of ( $Z$ )-1-( $p$-tolylsulfonylamino)hex-3-ene 26a. $(Z)$-Homoallylic tosylamide $26 \mathbf{a}(0.20 \mathrm{~g}, 0.79 \mathrm{mmol})$ was reacted under the general conditions described above and the reaction appeared complete after 1 h . The crude product was obtained as a yellow oil, which was purified by cc (dichloromethane) to give an inseparable mixture of isomers ( 0.12 g , $40 \%$ ) in the ratio $2: 1$. The isomer ratio was determined by measurement of the signals corresponding to the alkyl methyl groups in the ${ }^{1} \mathrm{H}$ NMR spectrum. Comparison of the ${ }^{1} \mathrm{H}$ NMR spectrum with that for iodocyclisation of the corresponding $(E)$-tosylamide 24b revealed that the minor product was the 2,3-trans-iodopyrrolidine 36b (see above). Data for mixture: $R_{\mathrm{f}}$ 0.52 (dichloromethane); $\delta_{\mathrm{H}}(270) 7.80-7.73(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$, both isomers), 7.45-7.30 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$, both isomers), 4.20-4.04 $(2 \mathrm{H}, \mathrm{m}$, both isomers $), 3.89-3.70(2 \mathrm{H}, \mathrm{m}$, both isomers $), 3.55-$ $2.95(4 \mathrm{H}, \mathrm{m}$, both isomers), $2.41(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$, minor isomer), $2.38(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$, major isomer), $2.27-1.39(8 \mathrm{H}, \mathrm{m}$, both isomers), $1.04(3 \mathrm{H}, \mathrm{t}, J 7.3$, major isomer) and $0.96(3 \mathrm{H}, \mathrm{t}, J 7.5$, $2^{\prime}-\mathrm{Me}$, minor isomer).

Cyclisation of ( $Z$ )-1-(methylsulfonylamino)hex-3-ene 26b. $(Z)$-Homoallylic mesylamide 26b $(0.02 \mathrm{~g}, 0.07 \mathrm{mmol})$ in acetonitrile $(0.2 \mathrm{ml})$ was reacted as described in the general procedure for 1 h . The crude product was purified by cc (dichloromethane) to give an inseparable mixture of isomers in the ratio $1: 1(0.06 \mathrm{~g}, 45 \%)$. The ratio of isomers was determined by integration of the $-\mathrm{SO}_{2} \mathrm{Me}$ peaks in the ${ }^{1} \mathrm{H}$ NMR spectrum. Data for the mixture: $R_{\mathrm{f}} 0.53$ (dichloromethane); $\delta_{\mathrm{H}}(250) 4.79-4.68(1 \mathrm{H}, \mathrm{m}), 4.31-4.20(1 \mathrm{H}, \mathrm{m}), 4.08-3.96(1 \mathrm{H}$, $\mathrm{m}), 3.62-3.15(4 \mathrm{H}, \mathrm{m}), 3.00\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{Me}\right), 2.98(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{SO}_{2} \mathrm{Me}\right), 2.11-1.34(8 \mathrm{H}, \mathrm{m}), 1.06(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$ and 0.97 $(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$.

Cyclisation of ( $Z$ )-3-( $p$-tolylsulfonylamino)dec-5-ene 26c. $(Z)$-Homoallylic tosylamide $26 \mathbf{c}(0.10 \mathrm{~g}, 0.34 \mathrm{mmol})$ was cyclised under the conditions described in the general procedure above for 1 h . The crude product was purified by cc (dichloromethane) and gave an inseparable mixture of isomers in the ratio $3: 2: 2(0.09 \mathrm{~g}, 65 \%)$. The two minor isomers were found (by comparison of ${ }^{1} \mathrm{H}$ NMR data for the mixture) to be identical to the products ( $\mathbf{3 6 f}$ and $\mathbf{3 7 f}$ ) obtained from the cyclisation of the corresponding ( $E$ )-homoallylic tosylamide $\mathbf{2 4 f}$ (see above). The major isomer was tentatively assigned the all-cis-structure 44.

## Potassium carbonate as base: general procedure

A mixture of the cyclisation precursor ( $1.0 \mathrm{mmol}, 1 \mathrm{eq}$.) and potassium carbonate ( $0.41 \mathrm{~g}, 3 \mathrm{mmol}, 3$ eq.) in dry acetonitrile $(1 \mathrm{ml})$ and water $(0.01 \mathrm{ml})$ was stirred for 0.5 h , after which time iodine ( $0.76 \mathrm{~g}, 3 \mathrm{mmol}, 3$ eq.) was added portionwise. The reaction was followed by TLC analysis, and once complete (typically $0.5-1 \mathrm{~h}$ ), the mixture was quenched with saturated aqueous sodium thiosulfate $(1 \mathrm{ml})$. The resulting layers were separated and the aqueous layer extracted with dichloromethane $(3 \times 2$ ml ). The combined organic solutions were dried and evaporated to yield the crude product which was purified by column chromatography.
(2SR,3RS)-2-Ethyl-3-iodo-1-( $p$-tolylsulfonyl)pyrrolidine 36b. (E)-Homoallylic sulfonamide 24b ( $0.20 \mathrm{~g}, 0.79 \mathrm{mmol}$ ) was cyclised under the conditions described above using potassium carbonate as the base, and the reaction was complete after 20 min . The crude product was purified by recrystallisation (hexane-ethyl acetate $9: 1$ ) to yield the iodopyrrolidine $\mathbf{3 6 b}(0.22$ $\mathrm{g}, 73 \%$ ) as a colourless crystalline solid, mp $108-110^{\circ} \mathrm{C}$; other spectral and analytical data were identical to the sample prepared using sodium hydrogen carbonate as the base (see above).
(2SR,3RS,5SR)-2-Butyl-3-iodo-5-methyl-1-(p-tolylsulfonyl)pyrrolidine 36d. ( $E$ )-Homoallylic tosylamide $24 d$ ( $0.20 \mathrm{~g}, 0.68$ mmol ) was subjected to the cyclisation conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$, and the reaction was complete after 25 min . The resulting crude product was purified by cc (dichloromethane) then recrystallised (hexane-ethyl acetate $9: 1)$ to afford the iodopyrrolidine $\mathbf{3 6 d}(0.24 \mathrm{~g}, 85 \%)$ as a colourless, crystalline solid, $\mathrm{mp} 84-85^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.52$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2957$ (s), 2928 (s), 2856 (w), 1597 (w), 1458 (s) and $1335(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.84(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 7.30(2 \mathrm{H}, \mathrm{d}$, $J 8.3$, Ar-H), $4.29-4.20(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{and} 3-\mathrm{H}), 4.15(1 \mathrm{H}, \mathrm{dqd}$, $J 9.0,6.6$ and $3.0,5-\mathrm{H}), 2.89(1 \mathrm{H}$, ddd, $J 15.2,9.0$ and 6.1 , $\left.4-\mathrm{H}_{\alpha}\right), 2.43(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.10(1 \mathrm{H}, \mathrm{ddd}, J 15.2,3.0$ and 3.0, $\left.4-\mathrm{H}_{\beta}\right), 2.05-1.98\left(2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}\right), 1.42(3 \mathrm{H}, \mathrm{d}, J 6.6,5-\mathrm{Me})$, $1.37-1.21\left(4 \mathrm{H}, \mathrm{m}, 2^{\prime}-\right.$ and $\left.3^{\prime}-\mathrm{CH}_{2}\right)$ and $0.91(3 \mathrm{H}, \mathrm{t}, J 6.9$, $4^{\prime}-\mathrm{Me}$ ); $\delta_{\mathrm{C}}$ (67.8) 142.8, 139.6 (both Ar-C), 129.5, 126.9 (both $\mathrm{Ar}-\mathrm{CH}), 74.3(2-\mathrm{CH}), 55.2(5-\mathrm{CH}), 43.2\left(4-\mathrm{CH}_{2}\right), 34.4\left(1^{\prime}-\mathrm{CH}_{2}\right)$, $28.2\left(2^{\prime}-\mathrm{CH}_{2}\right), 22.3\left(3^{\prime}-\mathrm{CH}_{2}\right), 22.0(5-\mathrm{Me}), 21.3(3-\mathrm{CH}), 21.2$ (Ar-Me) and $13.8\left(4^{\prime}-\mathrm{Me}\right) ; m / z(\mathrm{FAB}) 422\left(\mathrm{M}^{+}+1,68 \%\right), 391$ (7), 364 (39), 294 (47), 198 (100), 155 (69), 136 (56), 91 (86) and 77 (30) [Found: C, 45.7; H, 6.0; N, 3.6; $\mathrm{M}^{+}+1,422.0596$. $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{INO}_{2} \mathrm{~S}$ requires $\mathrm{C}, 45.6 ; \mathrm{H}, 5.7 ; \mathrm{N}, 3.3 \% ; \mathrm{C}_{16} \mathrm{H}_{25} \mathrm{INO}_{2} \mathrm{~S}$ requires $M, 422.0566]$.
(2SR,3RS,5SR)-2-Butyl-5-ethyl-3-iodo-1-( $p$-tolylsulfonyl)pyrrolidine 36 f . ( $E$ )-Homoallylic tosylamide $24 \mathrm{f}(0.20 \mathrm{~g}, 0.64$ mmol ) was cyclised under the conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ and the reaction was complete after 30 min. The crude product was purified by cc (dichloromethane) then recrystallised (hexane-ethyl acetate $9: 1$ ) to yield the title compound $\mathbf{3 6 f}(0.21 \mathrm{~g}, 74 \%)$ as a colourless, crystalline solid: mp $72-73{ }^{\circ} \mathrm{C} ; R_{\mathrm{f}} 0.52$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2950$ (s), 2927 (s), 2872 (w), 1465 (w), 1337 (s); $\delta_{\mathrm{H}}$ (400) 7.82 ( $2 \mathrm{H}, \mathrm{d}$, $J 8.3$, Ar-H), 7.29 (2H, d, J 8.3, Ar-H), 4.27-4.19 (2H, m, 2- and $3-\mathrm{H}), 3.86(1 \mathrm{H}$, dddd, $J 11.2,8.7,3.3$ and $3.2,5-\mathrm{H}), 2.73(1 \mathrm{H}$, ddd, $J 15.2,8.7$ and 6.6, $\left.4-\mathrm{H}_{\alpha}\right), 2.41(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.25(1 \mathrm{H}$, ddd, $J 15.2,3.2$ and $\left.2.2,4-\mathrm{H}_{\beta}\right), 2.09-1.50\left(4 \mathrm{H}, \mathrm{m}, 1^{\prime}-\right.$ and $\left.1^{\prime \prime}-\mathrm{CH}_{2}\right), 1.47-1.28\left(4 \mathrm{H}, \mathrm{m}, 2^{\prime}-\right.$ and $\left.3^{\prime}-\mathrm{CH}_{2}\right), 0.88(3 \mathrm{H}, \mathrm{t}, J 7.2$, $\left.2^{\prime \prime}-\mathrm{Me}\right)$ and $0.79\left(3 \mathrm{H}, \mathrm{t}, J 7.4,4^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{H}}\left(\mathrm{d}_{6}\right.$-acetone, 400) 7.85 $(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 7.37(2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $4.48(1 \mathrm{H}$, ddd, $J 6.6,2.2$ and $1.6,3-\mathrm{H}), 4.23(1 \mathrm{H}$, ddd, $J 10.1,3.2$ and $1.6,2-\mathrm{H})$, $3.89(1 \mathrm{H}$, dddd, $J 11.2,8.7,3.3$ and $3.2,5-\mathrm{H}), 2.88(1 \mathrm{H}$, ddd, $J 15.3,8.7$ and $\left.6.6,4-\mathrm{H}_{\alpha}\right), 2.59(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.23(1 \mathrm{H}$, ddd, $J 15.3,3.2$ and $\left.2.2,4-\mathrm{H}_{\beta}\right), 2.20-1.95\left(2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}\right), 1.72(1 \mathrm{H}$, ddq, $J 14.1,11.2$ and $\left.7.3,1^{\prime \prime}-\mathrm{H}_{\alpha}\right), 1.49(1 \mathrm{H}$, dqd, $J 14.1,7.3$ and $\left.3.3,1^{\prime \prime}-\mathrm{H}_{\beta}\right), 1.47-1.15\left(4 \mathrm{H}, \mathrm{m}, 2^{\prime}-\right.$ and $\left.3^{\prime}-\mathrm{CH}_{2}\right), 0.92(3 \mathrm{H}, \mathrm{t}$, $\left.J 7.3,2^{\prime \prime}-\mathrm{Me}\right)$ and $0.86\left(3 \mathrm{H}, \mathrm{t}, J 7.5,4^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{C}}(67.8) 142.9$, 139.6 (both Ar-C), 129.3, 127.0 (both Ar-CH), 74.2 (2-CH), $61.4(5-\mathrm{CH}), 39.3\left(4-\mathrm{CH}_{2}\right), 34.5\left(1^{\prime}-\mathrm{CH}_{2}\right), 31.5\left(1^{\prime \prime}-\mathrm{CH}_{2}\right), 26.4$, 25.9 (both $\mathrm{CH}_{2}$ ), 21.4 (3-CH), 21.3 ( $\mathrm{Ar}-\mathrm{Me}$ ), $13.9\left(4^{\prime}-\mathrm{Me}\right)$ and 10.9 ( $2^{\prime \prime}-\mathrm{Me}$ ); $m / z(\mathrm{ES}) 436\left(\mathrm{M}^{+}+1\right), 391$ (10), 102 (28), 83 (72) and 60 (100) [Found C, $46.9 ; \mathrm{H}, 6.15 ; \mathrm{N}, 3.3 . \mathrm{C}_{17} \mathrm{H}_{26} \mathrm{INO}_{2} \mathrm{~S}$ requires $\mathrm{C}, 46.9 ; \mathrm{H}, 6.0 ; \mathrm{N}, 3.3 \%$ ].

## (2SR,3RS,5SR)-2-Butyl-5-ethyl-3-iodo-1-methylsulfonyl-

 pyrrolidine 36g. ( $E$ )-Homoallylic sulfonamide $\mathbf{2 4 g}$ ( $0.01 \mathrm{~g}, 0.04$ $\mathrm{mmol})$ in acetonitrile $(0.2 \mathrm{ml})$ was cyclised under the conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ and the reaction was complete after 25 min . The resulting crude product was purified by cc (dichloromethane) to afford the iodopyrrolidine$\mathbf{3 6 g}(0.01 \mathrm{~g}, 72 \%)$ as a pale yellow, viscous oil: $R_{\mathrm{f}} 0.59$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 2952$ (s), 2931 (s), 2875 (w), 1461 (w) and $1341(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 4.29(1 \mathrm{H}$, ddd, $J 6.3,2.3$ and 1.7, $3-\mathrm{H}), 4.05(1 \mathrm{H}$, ddd, $J 10.3,3.3$ and $1.7,2-\mathrm{H}), 3.88(1 \mathrm{H}$, dddd, $J 11.9,9.0,3.2$ and $2.9,5-\mathrm{H}), 3.06\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{Me}\right), 2.41-2.29$ $\left(2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}\right), 2.74\left(1 \mathrm{H}\right.$, ddd, $J 15.3,9.0$ and $\left.6.3,4-\mathrm{H}_{\alpha}\right), 2.32$ $\left(1 \mathrm{H}\right.$, ddd, $J 15.3,3.2$ and $\left.2.3,4-\mathrm{H}_{\beta}\right), 2.09(1 \mathrm{H}$, ddq, $J 15.0,11.9$ and $\left.7.3,1^{\prime \prime}-\mathrm{H}_{\alpha}\right), 1.43\left(1 \mathrm{H}\right.$, dqd, $J 15.0,7.3$ and $\left.2.9,1^{\prime \prime}-\mathrm{H}_{\beta}\right), 1.38-$ $1.24\left(4 \mathrm{H}, \mathrm{m}, 2^{\prime}-\right.$ and $\left.3^{\prime}-\mathrm{CH}_{2}\right), 0.95\left(3 \mathrm{H}, \mathrm{t}, J 7.3,2^{\prime \prime}-\mathrm{Me}\right)$ and 0.88 $\left(3 \mathrm{H}, \mathrm{t}, J 7.5,4^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{C}}(67.8) 73.9(2-\mathrm{CH}), 61.4(5-\mathrm{CH}), 42.4$ $\left(\mathrm{SO}_{2} \mathrm{Me}\right), 38.5\left(4-\mathrm{CH}_{2}\right), 35.2\left(1^{\prime \prime}-\mathrm{CH}_{2}\right), 28.4,26.2,22.5$ (all $\mathrm{CH}_{2}$ ), 22.1 ( $3-\mathrm{CH}$ ), $13.9\left(2^{\prime \prime}-\mathrm{Me}\right)$ and $11.1\left(4^{\prime}-\mathrm{Me}\right) ; \mathrm{m} / \mathrm{z}$ (ES) $360\left(\mathrm{M}^{+}+1,38 \%\right), 254$ (36), 234 (39), 193 (30), 83 (66) and 64 (100) [Found: $\mathrm{M}^{+}+1,360.0494 . \mathrm{C}_{11} \mathrm{H}_{23} \mathrm{INO}_{2} \mathrm{~S}$ requires $M$, 360.0499].
(2SR,3RS,5SR)-5-Ethyl-3-iodo-2-pentyl-1-(p-tolylsulfonyl)pyrrolidine 36h. (E)-Homoallylic sulfonamide $\mathbf{2 4 h}(0.05 \mathrm{~g}, 0.16$ $\mathrm{mmol})$ in acetonitrile ( 0.3 ml ) was cyclised under the conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ and the reaction was complete after 25 min . The resulting crude product was purified by cc (dichloromethane) to yield the title compound $\mathbf{3 6 h}$ $(0.06 \mathrm{~g}, 79 \%)$ as a pale yellow, viscous oil: $R_{\mathrm{f}} 0.49$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2952$ (s), 2931 (s), 2875 (w), 1461 (w) and $1342(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 7.84(2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), 7.28 ( $2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), 4.29-4.21 ( $2 \mathrm{H}, \mathrm{m}, 2-$ and $3-\mathrm{H}$ ), $3.85(1 \mathrm{H}$, dddd, $J 11.3$, 8.8, 3.2 and $3.1,5-\mathrm{H}), 2.71\left(1 \mathrm{H}\right.$, ddd, $J 15.2,8.8$ and $\left.6.6,4-\mathrm{H}_{\alpha}\right)$, $2.42(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.25\left(1 \mathrm{H}\right.$, ddd, $J 15.2,3.2$ and $\left.2.2,4-\mathrm{H}_{\beta}\right)$, $2.09-1.50\left(4 \mathrm{H}, \mathrm{m}, 1^{\prime}-\right.$ and $\left.1^{\prime \prime}-\mathrm{CH}_{2}\right), 1.49-1.16\left(6 \mathrm{H}, \mathrm{m}, 2^{\prime}-\right.$, $3^{\prime}-$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 0.94\left(3 \mathrm{H}, \mathrm{t}, J 6.9,2^{\prime \prime}-\mathrm{Me}\right)$ and $0.88(3 \mathrm{H}, \mathrm{t}, J 7.5$, $5^{\prime}-\mathrm{Me}$ ); $\delta_{\mathrm{C}}(67.8) 142.9,140.0$ (both Ar-C), 129.3, 127.0 (both Ar-CH), 74.2 (2-CH), 61.4 (5-CH), 39.3, 34.5, 31.5, 26.4, 25.9, 22.4 (all $\left.\mathrm{CH}_{2}\right), 21.5(3-\mathrm{CH}), 21.3(\mathrm{Ar}-\mathrm{Me}), 13.9(11-\mathrm{Me})$ and 11.0 (1-Me) [Found: C, 48.3; H, 6.3; N 3.1. $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{INO}_{2} \mathrm{~S}$ requires $\mathrm{C}, 48.1 ; \mathrm{H}, 6.2$; $\mathrm{N}, 3.1 \%]$.
(2SR,3RS,5SR)-5-Butyl-3-iodo-2-pentyl-1-(p-tolylsulfonyl)pyrrolidine 36i. ( $E$ )-Homoallylic sulfonamide $24 \mathrm{i}(0.30 \mathrm{~g}, 0.85$ $\mathrm{mmol})$ was cyclised under the general conditions $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$, and the reaction was complete after 35 min . The resulting crude product was purified by cc (dichloromethane) to yield the title compound $36 \mathrm{i}(0.33 \mathrm{~g}, 78 \%)$ as a viscous, pale yellow oil: $R_{\mathrm{f}} 0.62$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2950$ (s), 2935 (s), 2871 (w), 1461 (w), 1358 (s) and 1342 (s); $\delta_{\mathrm{H}}$ (400) 7.82 ( $2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $7.29(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 4.24-4.19(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{and} 3-\mathrm{H}), 3.94$ $(1 \mathrm{H}$, dddd, $J 11.2,8.7,3.4$ and $3.2,5-\mathrm{H}), 3.71(1 \mathrm{H}$, ddd, $J 15.2$, 8.7 and $\left.6.6,4-\mathrm{H}_{\alpha}\right), 2.42(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.20(1 \mathrm{H}$, ddd, $J 15.2$, 3.4 and $\left.2.3,4-\mathrm{H}_{\beta}\right), 2.06-2.01\left(2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}\right), 1.71-1.61(2 \mathrm{H}$, $\left.\mathrm{m}, 1^{\prime \prime}-\mathrm{CH}_{2}\right), 1.61-1.11\left(10 \mathrm{H}, \mathrm{m}, 2^{\prime}-, 2^{\prime \prime}-, 3^{\prime}-, 3^{\prime \prime}-\right.$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right)$, $0.89(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Me})$ and $0.83(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{Me}) ; \delta_{\mathrm{C}}(67.8)$ 142.9, 139.9 (both Ar-C), 129.0, 127.0 (both Ar-CH), 73.9 $(2-\mathrm{CH}), 60.0(5-\mathrm{CH}), 39.8\left(4-\mathrm{CH}_{2}\right), 34.4\left(1^{\prime}-\mathrm{CH}_{2}\right), \quad 33.0$ (1"- $\mathrm{CH}_{2}$ ), 31.6, 29.4, 26.8, 22.5, 22.4 (all $\mathrm{CH}_{2}$ ), 21.6 (3-CH), $21.4(\mathrm{Ar}-\mathrm{Me}), 13.9(\mathrm{Me})$ and $13.8(\mathrm{Me}) ; m / z(\mathrm{ES}) 478\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ), 391 (5), 254 (96) and 60 (7) [Found: $\mathrm{M}^{+}+1,478.1280$. $\mathrm{C}_{20} \mathrm{H}_{33} \mathrm{INO}_{2} \mathrm{~S}$ requires $M, 478.1279$ ].
(2SR,3RS,5SR)-5-Ethyl-3-iodo-2-phenyl-1-(p-tolylsulfonyl)pyrrolidine $\mathbf{3 6 j}$. ( $E$ )-Homoallylic tosylamide $\mathbf{2 4 j}(1.00 \mathrm{~g}, 3.04$ mmol ) was subjected to the iodocyclisation conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$, and the reaction was complete after 45 min . The resulting orange solid was purified by cc (dichloromethane) followed by recrystallisation (hexane-ethyl acetate $9: 1$ ) to afford the 2-phenyliodopyrrolidine $\mathbf{3 6 j}(1.12 \mathrm{~g}, 83 \%)$ as a pale cream solid: mp $156-157^{\circ} \mathrm{C}$; $R_{\mathrm{f}} 0.52$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3286$ (s), 3062 (s), 2966 (s), 2876 (w), 2251 (w), 1700 (w), 1676 (w) and 1446 (s); $\delta_{\mathrm{H}}(400) 7.59(2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), $7.31-7.16$ ( $7 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ), $5.32(1 \mathrm{H}, \mathrm{d}, J 2.3,2-\mathrm{H}), 4.28-4.18(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{and} 5-\mathrm{H}), 2.74$ $\left(1 \mathrm{H}\right.$, ddd, $J 15.0,8.7$ and $\left.6.7,4-\mathrm{H}_{\alpha}\right), 2.40(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.25$
( 1 H , ddd, $J 15.0,3.0$ and $2.4,4-\mathrm{H}_{\beta}$ ), 1.98-1.84 ( $2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}$ ) and $0.91\left(3 \mathrm{H}, \mathrm{t}, J 7.4,2^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{H}}\left(270, \mathrm{~d}_{6}\right.$-benzene) 8.05-7.13 $(9 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 5.70(1 \mathrm{H}, \mathrm{d}, J 2.3,2-\mathrm{H}), 4.28$ ( 1 H , dddd, $J 11.2$, 8.3, 3.3 and $3.2,5-\mathrm{H}), 3.98(1 \mathrm{H}$, ddd, $J 6.6,3.3$ and $2.3,3-\mathrm{H}$ ), $2.72\left(1 \mathrm{H}, \mathrm{ddd}, J 14.8,11.2\right.$ and $\left.6.6,4-\mathrm{H}_{\alpha}\right), 2.62(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me})$, 2.31-2.19 ( $2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}$ ), $2.10(1 \mathrm{H}$, ddd, $J 14.8,3.3$ and 3.3, $4-\mathrm{H}_{\beta}$ ) and $0.93\left(3 \mathrm{H}, \mathrm{t}, J 7.3,2^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{C}}(100)$ 143.7, 140.4, 134.5 (all Ar-C), 129.6, 128.5, 128.0, 127.9, 126.6 (all Ar-CH), 74.8 ( $2-\mathrm{CH}$ ), $62.9(5-\mathrm{CH}), 41.8\left(4-\mathrm{CH}_{2}\right), 28.7\left(1^{\prime}-\mathrm{CH}_{2}\right), 26.2(3-\mathrm{CH})$, 21.6 (Ar-Me) and 10.7 ( $2^{\prime}-\mathrm{Me}$ ); m/z (ES) 456 ( $\mathrm{M}^{+}+1,100 \%$ ), 282 (10), 276 (47) and 104 (21) [Found: C, 50.2; H, 5.0; N, 3.1. $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{INO}_{2} \mathrm{~S}$ requires C, $50.1 ; \mathrm{H}, 4.9 ; \mathrm{N}, 3.1 \%$ ].
(2SR,3RS,4SR,5RS)-4,5-Dimethyl-3-iodo-2-propyl-1-( $p$ tolylsulfonyl)pyrrolidine 38. (E)-anti-Homoallylic tosylamide 27b $(0.20 \mathrm{~g}, 0.7 \mathrm{mmol})$ was subjected to the iodocyclisation conditions described above $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ and the reaction was complete after 40 min . The resulting crude product was purified by cc (dichloromethane) followed by recrystallisation (hexaneethyl acetate $9: 1$ ) to yield the iodopyrrolidine $38(0.22 \mathrm{~g}, 76 \%)$ as a colourless, crystalline solid, $\mathrm{mp} 102-103^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.62$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2950(\mathrm{~s}), 2932(\mathrm{~s}), 2873$ (w), 1462 (w) and 1339 (s); $\delta_{\mathrm{H}}(400) 7.72$ ( $2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}$ ), $7.33(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 3.87(1 \mathrm{H}$, ddd, $J 8.7,6.2$ and 3.6 , $2-\mathrm{H}), 3.72(1 \mathrm{H}, \mathrm{dq}, J 7.2$ and $6.9,5-\mathrm{H}), 3.61(1 \mathrm{H}, \mathrm{dd}, J 11.6$ and 8.7, 3-H), 2.45 (3H, s, Ar-Me), 1.95-1.89 (2H, m, 1'-CH2), 1.60 $(1 \mathrm{H}, \mathrm{ddq}, 11.6,7.2$ and $6.8,4-\mathrm{H}), 1.48-1.36\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{CH}_{2}\right)$, $1.11(3 \mathrm{H}, \mathrm{d}, J 6.9,5-\mathrm{Me}), 0.97\left(3 \mathrm{H}, \mathrm{t}, J 7.2,3^{\prime}-\mathrm{Me}\right)$ and 0.99 (3H, d, $J 6.8,4-\mathrm{Me}$ ); $\delta_{\mathrm{C}}$ (67.8) 143.6, 134.6 (both Ar-C), 129.8, 127.4 (both Ar-CH), 69.4 (2-CH), 58.6 (5-CH), 46.5 ( $4-\mathrm{CH}$ ), $35.9\left(1^{\prime}-\mathrm{CH}_{2}\right), 30.5(3-\mathrm{CH}), 21.6$ (Ar-Me), $17.67\left(2^{\prime}-\mathrm{CH}_{2}\right), 17.2$ (5-Me), $14.2\left(3^{\prime}-\mathrm{Me}\right)$ and $12.7(4-\mathrm{Me}) ; m / z(\mathrm{FAB}) 422\left(\mathrm{M}^{+}+1\right.$, $36 \%$ ), 378 (13), 294 (28), 251 (8), 198 (37), 155 (29), 123 (16), 109 (24), 91 (52) and 69 (90) [Found: C, 46.0; H, 5.7; N, 3.0; $\mathrm{M}^{+}+1$, 422.0618. $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{INO}_{2} \mathrm{~S}$ requires C, $45.6 ; \mathrm{H}, 5.8 ; \mathrm{N}$, $3.3 \% ; \mathrm{C}_{16} \mathrm{H}_{25} \mathrm{INO}_{2} \mathrm{~S}$ requires $\left.M, 422.0650\right]$.

## (2RS,3SR,4RS,5RS)-4,5-Dimethyl-3-iodo-2-propyl-1-( $p$ -

tolylsulfonyl)pyrrolidine 39. (E)-syn-Homoallylic tosylamide 28b $(0.20 \mathrm{~g}, 0.7 \mathrm{mmol})$ was cyclised under the conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$, and the reaction was complete after 45 min . The crude product was purified by cc (dichloromethane) to yield the iodopyrrolidine $39(0.23 \mathrm{~g}$, $79 \%$ ) as a colourless crystalline solid, $\mathrm{mp} 108-109{ }^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.64$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2953$ (s), 2928 (s), 2879 (w), 1458 (w) and 1345 (s); $\delta_{\mathrm{H}}(250) 7.75$ ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), 7.28 ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), $4.31(1 \mathrm{H}$, ddd, $J 7.6,6.3$ and 4.1 , $2-\mathrm{H}), 3.50(1 \mathrm{H}, \mathrm{dd}, J 9.2$ and $6.3,3-\mathrm{H}), 3.20(1 \mathrm{H}, \mathrm{dq}, J 9.4$ and $6.5,5-\mathrm{H}), 2.42$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 2.10-1.95 ( $2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}$ ), 1.71 ( $1 \mathrm{H}, \mathrm{ddq}, J 9.4,9.2$ and $6.6,4-\mathrm{H}$ ), 1.49-1.38 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{2}^{\prime}-\mathrm{CH}_{2}$ ), $1.31(3 \mathrm{H}, \mathrm{d}, J 6.5,5-\mathrm{Me}), 0.97(3 \mathrm{H}, \mathrm{d}, J 6.6,4-\mathrm{Me})$ and 0.94 ( $3 \mathrm{H}, \mathrm{t}, J 7.3,3^{\prime}-\mathrm{Me}$ ); $\delta_{\mathrm{C}}$ (67.8) 143.0, 141.3 (both $\mathrm{Ar}-\mathrm{C}$ ), 129.6, 126.5 (both Ar-CH), 70.6 ( $2-\mathrm{CH}$ ), 61.8 ( $5-\mathrm{CH}$ ), 52.8 $(4-\mathrm{CH}), 36.8\left(1^{\prime}-\mathrm{CH}_{2}\right), 30.3(3-\mathrm{CH}), 21.5$ (Ar-Me), 18.3 (2'$\mathrm{CH}_{2}$ ), $16.8(5-\mathrm{Me}), 15.2(4-\mathrm{Me})$ and $14.1\left(3^{\prime}-\mathrm{Me}\right) ; m / z(\mathrm{FAB})$ $422\left(\mathrm{M}^{+}+1,1 \%\right), 421\left(\mathrm{M}^{+}, 39 \%\right), 378$ (13), 294 (28), 279 (5), 251 (8), 198 (29), 155 (29), 91 (60), 71 (44) and 69 (100) [Found: $\mathrm{C}, 45.7 ; \mathrm{H}, 5.9 ; \mathrm{N}, 3.0 \% ; \mathrm{M}^{+}, 421.0595 . \mathrm{C}_{16} \mathrm{H}_{24} \mathrm{INO}_{2}$ requires $M, 421.0574]$.

Cyclisation of ( $Z$ )-1-( $p$-tolylsulfonylamino)hex-3-ene 26a. $(Z)$-Homoallylic tosylamide 26a ( $0.02 \mathrm{~g}, 0.07 \mathrm{mmol}$ ) in acetonitrile ( 0.2 ml ) was reacted under the conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ for 1 h . Purification of the crude product by cc (dichloromethane) afforded a mixture of isomers in a ratio $2: 1(0.01 \mathrm{~g}, 52 \%)$. The isomer ratio was determined from measurement of the alkyl methyl peaks from the ${ }^{1} \mathrm{H}$ NMR spectrum. Data for mixture: $R_{\mathrm{f}} 0.58$ (methanoldichloromethane $2: 98)$; $\delta_{\mathrm{H}}(250) 7.75-7.72(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$, both isomers), $7.32-7.29$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$, both isomers), $5.01-4.85$
( $1 \mathrm{H}, \mathrm{m}$, minor isomer), 4.79-4.70 ( $1 \mathrm{H}, \mathrm{m}$, major isomer), 4.22$4.15(1 \mathrm{H}, \mathrm{m}$, major isomer), 3.98-3.89 ( $1 \mathrm{H}, \mathrm{m}$, minor isomer), $3.21-1.95(4 \mathrm{H}, \mathrm{m}$, both isomers), $2.38(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$, minor isomer), $2.36(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$, major isomer), 2.12-1.51 ( $8 \mathrm{H}, \mathrm{m}$, both isomers), $1.06(3 \mathrm{H}, \mathrm{t}, J 7.3$, minor isomer) and $0.94(3 \mathrm{H}, \mathrm{t}$, $J 7.2$, major isomer). These data indicate that the major isomer was the 2,3-trans-iodopyrrolidine $\mathbf{3 6 b}$.

Attempted cyclisation of ( $\boldsymbol{Z}$ )-1-(methylsulfonylamino)hex-3ene 26b. ( $Z$ )-Homoallylic mesylamide $26 \mathrm{~b}(0.02 \mathrm{~g}, 0.11 \mathrm{mmol})$ in acetonitrile $(0.2 \mathrm{ml})$ was reacted under the conditions described above $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ for 1 h , but no reaction was observed. The reaction was repeated and left stirring for 24 h at ambient temperature but, again, no reaction was found to have occurred.

Cyclisation of ( $Z$ )-3-( $p$-tolylsulfonylamino)dec-5-ene 26c. ( $Z$ )Homoallylic tosylamide $26 \mathrm{c}(0.10 \mathrm{~g}, 0.33 \mathrm{mmol})$ was reacted under the conditions described above $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ for 1 h . The crude product was purified by cc (dichloromethane) to give an inseparable mixture of isomers in the ratio of $c a .5: 1(0.09 \mathrm{~g}$, $63 \%$ ). The isomer ratio was determined by integration of the aromatic methyl groups in the ${ }^{1} \mathrm{H}$ NMR spectrum. The major isomer was found, by comparison with spectral data from the cyclisation of the corresponding ( $E$ )-precursor 24f, to be the foregoing ( $2 S R, 3 R S, 5 S R$ )-iodopyrrolidine $36 \mathbf{f}$ (see above). Data for the mixture: $R_{\mathrm{f}} 0.53$ (dichloromethane); the minor isomer showed distinctive resonances at $\delta_{\mathrm{H}}(250) 7.69(2 \mathrm{H}$, d, $J 8.3$, Ar-H), 7.38 ( $2 \mathrm{H}, \mathrm{d}, J 8.3$, Ar-H), 4.65-4.57 ( $1 \mathrm{H}, \mathrm{m}$ ), 4.42$4.32(1 \mathrm{H}, \mathrm{m}), 4.12-4.19(1 \mathrm{H}, \mathrm{m})$ and $2.43(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me})$; comparisons with the spectral data displayed by the pyrrolidine $\mathbf{3 7 f}$ suggested this was not the minor product which was hence assigned the all-cis-structure 44.

Cyclisation of ( $E$ )-2-(tert-butoxycarbonylamino)oct-4-ene 35a. ( $E$ )-Homoallylic amide $35 \mathrm{a}(0.03 \mathrm{~g}, 0.13 \mathrm{mmol}$ ) was reacted under the conditions described above $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ for 30 min . The crude product was purified by cc (methanoldichloromethane $5: 95$ ) to afford the carbamate as a mixture of isomers $\mathbf{4 5}$ in the ratio $2: 1(0.04 \mathrm{~g}, 79 \%)$ as a viscous yellow oil: $R_{\mathrm{f}} 0.78$ (methanol-dichloromethane 5:95); $v_{\max } / \mathrm{cm}^{-1} 3433$ (NH), 3248 (NH), 2932 (w), 2874 (w), 1714 (C=O), 1459 (s) and $1342(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 6.68(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}$, minor isomer), $6.52(1 \mathrm{H}$, br s , NH , major isomer), $4.31-3.98(4 \mathrm{H}, \mathrm{m}$, both isomers), 3.70-3.51 $(2 \mathrm{H}, \mathrm{m}$, both isomers), $2.41-2.15(2 \mathrm{H}, \mathrm{m}$, minor isomer), $2.10-1.23(10 \mathrm{H}, \mathrm{m}$, both isomers), $1.27(3 \mathrm{H}, \mathrm{d}, J 6.6$, Me , minor isomer), 1.25 ( 3 H , d. $J 6.7, \mathrm{Me}$, major isomer), 0.90 $(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me}$, minor isomer) and $0.89(3 \mathrm{H}, \mathrm{t}, J 7.3$, major isomer); $\delta_{\mathrm{C}}$ (67.8) 154.3 ( $\mathrm{C}=\mathrm{O}$, major isomer), 154.1 ( $\mathrm{C}=\mathrm{O}$, minor isomer), $79.6(\mathrm{CH}$, major isomer), $76.9(\mathrm{CH}$, minor isomer), $46.3\left(\mathrm{CH}_{2}\right.$, major isomer), $43.8\left(\mathrm{CH}_{2}\right.$, minor isomer), $38.0\left(\mathrm{CH}\right.$, major isomer), $37.7\left(\mathrm{CH}_{2}\right.$, minor isomer), $37.6\left(\mathrm{CH}_{2}\right.$, major isomer), $37.2\left(\mathrm{CH}\right.$, minor isomer), $35.5\left(\mathrm{CH}_{2}\right.$, major isomer), $32.9\left(\mathrm{CH}_{2}\right.$, minor isomer), 29.1 ( Me , minor isomer), $22.9\left(\mathrm{CH}_{2}\right.$, both isomers), $22.0(\mathrm{Me}$, major isomer), $13.5(\mathrm{Me}$, minor isomer) and 13.4 (Me, major isomer); $m / z$ (FAB) 298 $\left(\mathrm{M}^{+}+1,100 \%\right), 154$ (48), 136 (54), 88 (67), 69 (56) and 55 (76) [Found: $\mathrm{M}^{+}+1,298.0298 . \mathrm{C}_{9} \mathrm{H}_{17} \mathrm{INO}_{2}$ requires $M, 298.0304$ ].

Cyclisation of ( $E$ )-2-trifluoroacetylaminooct-4-ene 35c. ( $E$ )Homoallylic amide $35 \mathrm{c}(0.05 \mathrm{~g}, 0.23 \mathrm{mmol}$ ) was reacted under the conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ for 1 h . The crude product was purified by cc (dichloromethane) to afford a mixture of isomers 46c in a ratio of $3: 1(0.05 \mathrm{~g}, 57 \%)$, as a yellow oil. The isomer ratio was determined by integration of the alkyl-methyl peaks in the ${ }^{1} \mathrm{H}$ NMR spectrum. Data for mixture: $R_{\mathrm{f}} 0.47$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2980$ (s), 2951 (s), $1730,1504(\mathrm{~s})$ and $1346(\mathrm{~s}) ; \delta_{\mathrm{H}}(250)$ 4.12-4.09 ( $2 \mathrm{H}, \mathrm{m}$, major isomer), 4.01-3.97 ( $1 \mathrm{H}, \mathrm{m}$, major isomer), $3.62-3.60(1 \mathrm{H}, \mathrm{m}$, minor isomer), 3.58-3.49 ( $2 \mathrm{H}, \mathrm{m}$, minor isomer), $2.36(1 \mathrm{H}$, ddd,
$J$ 13.8, 4.3 and 2.9, major isomer), $1.85-1.40(11 \mathrm{H}, \mathrm{m}$, both isomers), $1.32(3 \mathrm{H}, \mathrm{d}, J 6.7$, Me, major isomer), $1.30(3 \mathrm{H}, \mathrm{d}$, $J 6.6$, Me, minor isomer) and $0.96(6 \mathrm{H}, \mathrm{t}, J 7.3$, Me, both isomers); $\delta_{\mathrm{C}}$ (67.8) 118.3 (C, major isomer), $78.6(\mathrm{CH}$, major isomer), 75.7 (minor isomer), 48.8 ( CH , major isomer), 46.3 $\left(\mathrm{CH}\right.$, minor isomer), $37.1\left(\mathrm{CH}\right.$, major isomer), $36.9\left(\mathrm{CH}_{2}\right.$, minor isomer), $36.8\left(\mathrm{CH}_{2}\right.$, major isomer), $36.6(\mathrm{CH}$, minor isomer), $34.5\left(\mathrm{CH}_{2}\right.$, major isomer), $33.1\left(\mathrm{CH}_{2}\right.$, minor isomer), $32.2\left(\mathrm{CH}_{2}\right.$, minor isomer), $22.5\left(\mathrm{Me}\right.$, major isomer), $22.3\left(\mathrm{CH}_{2}\right.$, major isomer), 21.9 ( Me , minor isomer) and 13.1 ( Me , both isomers). The sample decomposed before a mass spectrum could be obtained.

Cyclisation of 2-methoxycarbonylaminooct-4-ene 35e. (E)Homoallylic carbamate $35 \mathrm{e}(0.05 \mathrm{~g}, 0.23 \mathrm{mmol})$ was reacted under the conditions described in the general procedure $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ for 1 h . The crude product was purified by cc (methanol-dichloromethane $5: 95$ ) to afford a mixture of isomers 46e in a ratio of $2: 1(0.03 \mathrm{~g}, 48 \%)$, as a yellow oil. The isomer ratio was determined by integration of the-OMe peaks in the ${ }^{1} \mathrm{H}$ NMR spectrum of the mixture. Data for the mixture: $R_{\mathrm{f}} 0.23$ (methanol-dichloromethane $5: 95$ ); $\delta_{\mathrm{H}}(250)$ 4.52-4.48 $(1 \mathrm{H}, \mathrm{m}$, major isomer), $4.46-3.98(4 \mathrm{H}, \mathrm{m}$, both isomers), $3.92-$ $3.86(1 \mathrm{H}, \mathrm{m}$, minor isomer), $3.69(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$, major isomer), $3.68(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$, minor isomer), $2.80-2.71(1 \mathrm{H}, \mathrm{m}$, major isomer), 2.31-2.11 ( 2 H , minor isomer), 2.03-1.20 ( $9 \mathrm{H}, \mathrm{m}$, both isomers), $1.22(3 \mathrm{H}, \mathrm{d}, J 6.6$, Me, minor isomer), $1.20(3 \mathrm{H}, \mathrm{d}$, $J 6.6$, Me, major isomer), $0.95(3 \mathrm{H}, \mathrm{t}, J 7.3$, major isomer) and $0.93(3 \mathrm{H}, \mathrm{t}, J 7.3$, minor isomer).

## Iodocyclisations with no base present: general procedure

To a stirred solution of the cyclisation precursor ( 1.0 mmol , 1 eq .) in acetonitrile ( 1 ml ) was added portionwise iodine ( $3.0 \mathrm{mmol}, 3$ eq.) and the reaction stirred at ambient temperature until complete according to TLC analysis (typically 2-20 $\mathrm{min})$. TLC samples were taken every 30 seconds. The acetonitrile was removed under reduced pressure and the residue partitioned between dichloromethane ( 1 ml ) and saturated aqueous sodium thiosulfate ( 1 ml ). The resulting two layers were separated, and the aqueous layer extracted with dichloromethane ( $3 \times 2 \mathrm{ml}$ ). The combined organic solutions were dried and evaporated to give crude products which were purified by column chromatography.

## (2SR,3RS)-2-Ethyl-3-iodo-1-( $p$-tolylsulfonyl)pyrrolidine

36b. Homoallylic tosylamide 24b ( $0.20 \mathrm{~g}, 0.80 \mathrm{mmol}$ ) was cyclised under the general conditions with no base and the reaction was complete within 2 min . The crude product was purified by cc (dichloromethane) followed by recrystallisation (hexaneethyl acetate $9: 1$ ) to afford the title compound $\mathbf{3 6 b}(0.27 \mathrm{~g}, 88 \%)$ as a colourless crystalline solid, mp 108-109 ${ }^{\circ} \mathrm{C}$; other spectral and analytical data were identical to those displayed by samples prepared using either sodium hydrogen carbonate or potassium carbonate as the base (see above).
(2RS,3SR,5SR)-2-Butyl-3-iodo-5-methyl-1-(p-tolylsulfonyl)pyrrolidine 37d. ( $E$ )-Homoallylic tosylamide $24 \mathrm{~d}(0.2 \mathrm{~g}, 0.7$ mmol ) was cyclised under the general conditions (no base), and the reaction was complete after 3 min . The crude product was purified by cc (dichloromethane) then recrystallised (hexaneethyl acetate $9: 1$ ) to afford the pyrrolidine $\mathbf{3 7 d}(0.25 \mathrm{~g}, 86 \%)$ as a colourless, crystalline solid $\mathrm{mp} 102-103{ }^{\circ} \mathrm{C}: R_{\mathrm{f}} 0.56$ (dichloromethane): $v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2959$ (s), 2924 (s), 2875 (w) and 1342 (s); $\delta_{\mathrm{H}}(270) 7.77(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 7.33(2 \mathrm{H}, \mathrm{d}$, $J 8.3, \mathrm{Ar}-\mathrm{H}), 4.09(1 \mathrm{H}, \mathrm{ddd}, J 5.7,3.4$ and $3.2,3-\mathrm{H}), 4.01-3.97$ $(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 3.93(1 \mathrm{H}$, dqd, $J .5,6.6$ and $6.2,5-\mathrm{H}), 2.43(2 \mathrm{H}$, s, Ar-Me), $2.20\left(1 \mathrm{H}\right.$, ddd, $J 13.9,7.5$ and $\left.5.7,4-\mathrm{H}_{a}\right), 2.15(1 \mathrm{H}$, ddd, $J 13.9,6.2$ and $3.4,4-\mathrm{H}_{\beta}$ ), 1.43 ( $3 \mathrm{H}, \mathrm{d}, J 6.6,5-\mathrm{Me}$ ), 2.08$1.09\left(6 \mathrm{H}, \mathrm{m}, 1^{\prime}-, 2^{\prime}\right.$ - and $\left.3^{\prime}-\mathrm{CH}_{2}\right)$ and $0.92\left(3 \mathrm{H}, \mathrm{t}, J 7.3,4^{\prime}-\mathrm{Me}\right)$;
$\delta_{\mathrm{C}}$ (67.8) 143.2, 134.2 (both Ar-C), 129.7, 128.1 (both Ar-CH), 72.9 (2-CH), $56.4(5-\mathrm{CH}), 45.0\left(4-\mathrm{CH}_{2}\right), 37.5\left(1^{\prime}-\mathrm{CH}_{2}\right), 28.0$ $\left(2^{\prime}-\mathrm{CH}_{2}\right), 22.6\left(3^{\prime}-\mathrm{CH}_{2}\right), 22.2$ (Ar-Me), 22.1 (3-CH), 21.6 (5-Me) and $14.0\left(4^{\prime}-\mathrm{Me}\right) ; m / z(\mathrm{FAB}) 422\left(\mathrm{M}^{+}+1,100 \%\right), 410$ (11), 364 (52), 335 (20), 294 (55), 237 (23), 198 (81), 154 (74), 136 (64), 91 (75) and 77 (29) [Found: C, 45.7; H, 6.0; N, 3.5; $\mathrm{M}^{+}+1$, 422.0682. $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{INO}_{2} \mathrm{~S}$ requires $\mathrm{C}, 45.6 ; \mathrm{H}, 5.7$; N , $3.3 \%, \mathrm{C}_{16} \mathrm{H}_{25} \mathrm{INO}_{2} \mathrm{~S}$ requires $M, 422.0652$ ].
(2S,3R,5R)-2-Butyl-3-iodo-5-methyl-1-( $p$-tolylsulfonyl)pyrrolidine 37e. (2R)-(E)-Homoallylic tosylamide $24 \mathrm{e}(20 \mathrm{mg}$, 0.07 mmol ) in acetonitrile ( 0.2 ml ) was cyclised under the conditions described in the general procedure (no base) to give a yellow oil which was purified by recrystallisation (hexane-ethyl acetate $9: 1$ ) to afford the iodopyrrolidine $\mathbf{3 7 e}(25 \mathrm{mg}, 85 \%)$ as a colourless, crystalline solid, $\mathrm{mp} 112-114^{\circ} \mathrm{C}:[a]_{\mathrm{D}}+14.0(c$ 0.015 in $\mathrm{CHCl}_{3}$ ); other spectral and analytical data, except the melting point, were identical to those obtained for the racemic product $\mathbf{3 7 d}$ (see above).
(2RS,3SR,5SR)-2-Butyl-5-ethyl-3-iodo-1-(p-tolylsulfonyl)-
pyrrolidine 37f. i) By iodocyclisation of sulfonamide 24f. (E)Homoallylic tosylamide $\mathbf{2 4 f}(0.20 \mathrm{~g}, 0.64 \mathrm{mmol})$ was cyclised under the general conditions described above (no base); the reaction was complete after 3 min . The crude product was purified by cc (dichloromethane) then recrystallised (ethyl acetatehexane $15: 85$ ) to give the iodopyrrolidine $37 \mathrm{f}(0.23 \mathrm{~g}, 78 \%)$ as a pale orange, crystalline solid: mp $94-95^{\circ} \mathrm{C} ; R_{\mathrm{f}} 0.56$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2918$ (s), 2930 (s), 2889 (s), 1597 (w), 1335 (s) and 1304 (s); $\delta_{\mathrm{H}}$ (250) 7.77 ( $2 \mathrm{H}, \mathrm{d}, J 8.2$, Ar-H), $7.34(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 4.07(1 \mathrm{H}$, ddd, $J 5.6,3.5$ and 3.4 , $3-\mathrm{H}), 3.95(1 \mathrm{H}$, ddd, $J 9.1,3.6$ and $3.4,2-\mathrm{H}), 3.72(1 \mathrm{H}$, dddd, $J 8.0,7.5,6.8$ and 4.2, 5-H), $2.43(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.10(1 \mathrm{H}$, ddd, $J 14.2,7.5$ and $\left.5.6,4-\mathrm{H}_{\omega}\right), 2.02(1 \mathrm{H}, \mathrm{ddd}, J 14.2,6.8$ and 3.5 , $\left.4-\mathrm{H}_{\beta}\right), 1.77\left(1 \mathrm{H}, \mathrm{ddq}, J 13.3,8.0\right.$ and $\left.7.2,1^{\prime \prime}-\mathrm{H}_{\alpha}\right), 1.44(1 \mathrm{H}, \mathrm{dqd}$, $J$ 13.3, 7.2 and $\left.4.2,1^{\prime \prime}-\mathrm{H}_{8}\right), 1.41-1.28\left(6 \mathrm{H}, \mathrm{m}, 1^{\prime}-2^{\prime}-\right.$ and $\left.3^{\prime}-\mathrm{CH}_{2}\right), 0.92(3 \mathrm{H}, \mathrm{t} . J 7.2, \mathrm{Me})$ and $0.90(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Me})$; $\delta_{\mathrm{C}}(100)$ 141.4, 136.3 (both Ar-C), 129.4, 127.9, (both Ar-CH), $72.7(2-\mathrm{CH}), 61.9(5-\mathrm{CH}), 41.9\left(4-\mathrm{CH}_{2}\right), 37.1\left(1^{\prime}-\mathrm{CH}_{2}\right), 28.4$ ( $1^{\prime \prime}-\mathrm{CH}_{2}$ ), 24.5, 22.5 (both $\mathrm{CH}_{2}$ ), 22.3 (Ar-Me), 21.9 (3-CH), $14.0\left(4^{\prime}-\mathrm{Me}\right)$ and 9.9 ( $2^{\prime \prime}-\mathrm{Me}$ ); $m / z$ (FAB) 435 ( $\mathrm{M}^{+}, 1 \%$ ), 406 (19), 378 (80), 326 (7), 251 (47), 212 (48), 187 (7) and 91 (100) [Found: C, 47.3; H, 6.2; N, 3.2; M ${ }^{+}$, 435.0745. $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{INSO}_{2}$ requires C, $46.9 ; \mathrm{H}, 6.0 ; \mathrm{N}, 3.3 \% ; M, 435.0729]$.
ii) By acid-catalysed isomerisation of iodopyrrolidine 36f. The ( $2 S R, 3 R S, 5 S R$ )-iodopyrrolidine $36 f(0.02 \mathrm{~g}, 0.05 \mathrm{mmol})$ in dry acetonitrile ( 0.2 ml ) was treated with $57 \%$ aqueous hydrogen iodide ( $2 \mu \mathrm{l}$ ) and the mixture stirred at ambient temperature for 20 min . The standard sodium thiosulfate work-up described above afforded the 2,5 -cis-pyrrolidine $37 \mathrm{f}(0.02 \mathrm{~g}, 94 \%)$ as a colourless solid, $\mathrm{mp} 95-97^{\circ} \mathrm{C}$ (mixed mp 94-96 ${ }^{\circ} \mathrm{C}$ ); other spectral and analytical data were identical to those displayed by the sample prepared by the iodocyclisation conditions described immediately above.

## (2SR,3RS,5SR)-2-Butyl-5-ethyl-3-iodo-1-methylsulfonyl-

pyrrolidine $\mathbf{3 6 g}$. ( $E$ )-Homoallylic methanesulfonamide $\mathbf{2 4 g}$ $(0.01 \mathrm{~g}, 0.04 \mathrm{mmol})$ was cyclised under the conditions described in the general procedure (no base) and the reaction was complete after 3 min . The reaction was left for a further 30 min in an attempt to effect isomerisation. The crude product was purified by cc (dichloromethane) to afford the title compound $\mathbf{3 6 g}$ ( 0.01 $\mathrm{g}, 78 \%$ ) as a viscous, pale yellow oil: spectral and analytical data were identical to those obtained for the same substrate, cyclised in the presence of potassium carbonate.
(2RS,3SR,5SR)-5-Ethyl-3-iodo-2-pentyl-1-( $p$-tolylsulfonyl)pyrrolidine 37 h . ( $E$ )-Homoallylic tosylamide $24 \mathrm{~h}(0.05 \mathrm{~g}, 0.16$ mmol ) was cyclised under the general conditions described above (no base); the reaction was complete after 3 min . The
crude product was purified by cc (dichloromethane) to afford the title compound $37 \mathrm{~h}(0.05 \mathrm{~g}, 74 \%)$ as a pale orange, viscous oil: $R_{\mathrm{f}} 0.56$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 2919$ (s), 2930 (s), 2886 (s), $1711(\mathrm{w}), 1595(\mathrm{w}), 1337(\mathrm{~s})$ and $1301(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.75(2 \mathrm{H}$, d, $J 8.2$, Ar-H), 7.33 ( $2 \mathrm{H}, \mathrm{d}, J 8.2$, Ar-H), 4.05 ( 1 H, ddd, $J 5.7$, 3.5 and $3.4,3-\mathrm{H}), 3.94(1 \mathrm{H}$, ddd, J 9.1, 3.6 and $3.4,2-\mathrm{H}), 3.72$ ( 1 H , dddd, $J 8.0,7.5,6.8$ and $4.2,5-\mathrm{H}), 2.44$ (3H, s, Ar-Me), $2.10\left(1 \mathrm{H}\right.$, ddd, $J 14.2,7.5$ and $\left.5.7,4-\mathrm{H}_{\alpha}\right), 2.02(1 \mathrm{H}$, ddd, $J 14.2$, 6.8 and $\left.3.5,4-\mathrm{H}_{\beta}\right), 1.77\left(1 \mathrm{H}\right.$, ddq, $J 13.4,8.0$ and $\left.7.1,1^{\prime \prime}-\mathrm{H}_{\alpha}\right)$, $1.44\left(1 \mathrm{H}, \mathrm{dqd}, J 13.4,7.1\right.$ and $\left.4.2,1^{\prime \prime}-\mathrm{H}_{\beta}\right), 1.43-1.20(8 \mathrm{H}, \mathrm{m}$, $1^{\prime}-, 2^{\prime}-, 3^{\prime}-$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 0.92(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$ and $0.90(3 \mathrm{H}, \mathrm{t}$, $J 7.1, \mathrm{Me}$ ); $\delta_{\mathrm{C}}(100) 141.2,136.2$ (both Ar-C), 129.4, 127.9, (both $\mathrm{Ar}-\mathrm{CH}$ ), $72.5(2-\mathrm{CH}), 61.9$ ( $5-\mathrm{CH}$ ), $41.9\left(4-\mathrm{CH}_{2}\right), 37.1$ ( $1^{\prime}-\mathrm{CH}_{2}$ ), $28.4\left(1^{\prime \prime}-\mathrm{CH}_{2}\right), 22.5\left(\mathrm{CH}_{2}\right), 22.3(\mathrm{Ar}-\mathrm{Me}), 22.1\left(\mathrm{CH}_{2}\right)$, $21.8(3-\mathrm{CH}), 14.2\left(5^{\prime}-\mathrm{Me}\right)$ and 9.93 ( $2^{\prime \prime}-\mathrm{Me}$ ) [Found: C, 48.6 ; H, 6.4; $\mathrm{N}, 3.0 . \mathrm{C}_{18} \mathrm{H}_{28} \mathrm{INO}_{2} \mathrm{~S}$ requires C, 48.1; H, 6.2; $\mathrm{N}, 3.1 \%$ ].
(2RS,3SR,5SR)-5-Butyl-3-iodo-2-pentyl-1-(p-tolylsulfonyl)pyrrolidine 37i. ( $E$ )-Homoallylic tosylamide $24 \mathrm{i}(0.30 \mathrm{~g}, 0.85$ mmol ) was subjected to the general iodocyclisation conditions (no base), and the reaction was complete after 15 min . The crude product was purified by cc (dichloromethane) to yield the iodopyrrolidine $37 \mathrm{i}(0.32 \mathrm{~g}, 76 \%)$ as a pale yellow, viscous oil: $R_{\mathrm{f}} 0.64$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2955$ (s), 2928 (s), 2858 (w), $1685(\mathrm{w}), 1598(\mathrm{w}), 1455(\mathrm{~s})$ and $1346(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.75(2 \mathrm{H}$, d, $J 8.3$, Ar-H), 7.33 ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), 4.05 ( 1 H, ddd, $J 5.1$, 3.4 and $3.3,3-\mathrm{H}), 3.94(1 \mathrm{H}$, ddd, J 8.7, 3.7 and $3.3,2-\mathrm{H}), 3.75$ ( 1 H , dddd, $J .5,7.8,6.9$ and $4.5,5-\mathrm{H}$ ), 2.43 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 2.13-1.74 (4H, m, 1'- and 4-CH2), 1.53-1.20 (12H, m, 1"-, 2'-, $2^{\prime \prime}-, 3^{\prime}-, 3^{\prime \prime}$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 0.94(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Me})$ and $0.88(3 \mathrm{H}, \mathrm{t}$, $J 7.3, \mathrm{Me}$ ); $\delta_{\mathrm{C}}(100)$ 143.5, 134.4 (both Ar-C), 129.6, 128.0 (both Ar-CH), 72.2 (2-CH), 60.6 (5-CH), 42.7 ( $4-\mathrm{CH}_{2}$ ), 37.1 ( $1^{\prime}-\mathrm{CH}_{2}$ ), $35.3,31.6,27.9,25.4,22.5,22.4$ (all $\mathrm{CH}_{2}$ ), 21.7 (3-CH), 21.5 (Ar-Me), $14.0(\mathrm{Me})$ and $13.9(\mathrm{Me}) ; m / z(\mathrm{ES}) 478\left(\mathrm{M}^{+}+1\right.$, $45 \%$ ), 419 (100), 391 (24), 352 (33), 254 (56), 83 (28) and 60 (22) [Found: $\mathrm{M}^{+}+1,478.1277 . \mathrm{C}_{20} \mathrm{H}_{33} \mathrm{INO}_{2} \mathrm{~S}$ requires $M, 478.1279$ ] [Found: C, $50.3 ; \mathrm{H}, 6.9 ; \mathrm{N}, 3.1 . \mathrm{C}_{20} \mathrm{H}_{32} \mathrm{INO}_{2} \mathrm{~S}$ requires C, 50.3 ; H, 6.8; N, 2.9\%].
(2RS,3SR,5SR)-5-Ethyl-3-iodo-2-phenyl-1-( $p$-tolylsulfonyl)pyrrolidine 37 j . ( $E$ )-Homoallylic tosylamide $\mathbf{2 4 j}(0.50 \mathrm{~g}, 1.52$ mmol ) was cyclised under the foregoing general conditions (no base), and the reaction was complete after 10 min . The crude product was purified by cc (dichloromethane) followed by recrystallisation (hexane-ethyl acetate $9: 1$ ) to yield the iodopyrrolidine $\mathbf{3 7 j}(0.42 \mathrm{~g}, 76 \%)$ as a colourless, flaky solid: mp $110-112{ }^{\circ} \mathrm{C} ; R_{\mathrm{f}} 0.54$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3061$ (w), 2966 (s), 2993 (s), 2876 (w), 1446 (w) and 1349 (s); $\delta_{\mathrm{H}}$ (400) 7.96 ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), $7.42-7.25$ ( $7 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ), 4.98 ( $1 \mathrm{H}, \mathrm{d}$, $J 4.9,2-\mathrm{H}), 4.11(1 \mathrm{H}$, ddd, $J 6.2,6.2$ and $4.9,3-\mathrm{H}), 3.92(1 \mathrm{H}$, dddd, $J$ 9.6, 6.2, 6.2 and $4.5,5-\mathrm{H}), 2.43$ ( $3 \mathrm{H}, \mathrm{s}$, Ar-Me), 2.28 $\left(1 \mathrm{H}, \mathrm{ddq}, J 13.5,9.6\right.$ and $\left.7.5,1^{\prime}-\mathrm{H}_{u}\right), 2.12(2 \mathrm{H}$, app. dd, $J 6.2$ and $\left.6.2,4-\mathrm{CH}_{2}\right), 1.75\left(1 \mathrm{H}, \mathrm{dqd}, J 13.5,7.5\right.$ and $\left.4.5,1^{\prime}-\mathrm{H}_{\beta}\right)$ and 1.01 ( $3 \mathrm{H}, \mathrm{t}, J 7.5,2^{\prime}-\mathrm{Me}$ ); $\delta_{\mathrm{C}}$ (100) 142.8, 140.1, 139.3 (all Ar-C), 129.1, 128.5, 127.8, 127.1, 126.4 (all Ar-CH), 76.6 (2-CH), 63.6 (5-CH), $38.4\left(4-\mathrm{CH}_{2}\right), 26.8\left(1^{\prime}-\mathrm{CH}_{2}\right), 24.7(3-\mathrm{CH}), 21.5(\mathrm{Ar}-\mathrm{Me})$ and $11.1\left(2^{\prime}-\mathrm{Me}\right) ; m / z(\mathrm{ES}) 456\left(\mathrm{M}^{+}+1,54 \%\right), 330(6), 276$ (15), 106 (100) and 74 (17) [Found: C, 50.4; H, 5.1; N, 3.2. $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{INO}_{2} \mathrm{~S}$ requires C, $50.1 ; \mathrm{H}, 4.9 ; \mathrm{N}, 3.1 \%$ ].
(2RS,3SR,5SR)-1-Phenylsulfonyl-5-ethyl-3-iodo-2-phenylpyrrolidine 37 k . The ( $E$ )-homoallylic benzenesulfonamide $\mathbf{2 4 k}$ ( $0.677 \mathrm{~g}, 2.15 \mathrm{mmol}$ ) was cyclised under the foregoing general conditions (no base); the reaction was complete after 10 min . The crude product was purified by cc (dichloromethane) followed by recrystallisation from hexane-ethyl acetate $(8: 1)$ to give the iodopyrrolidine $\mathbf{3 7 k}(0.768 \mathrm{~g}, 81 \%)$ as orange clusters: $\mathrm{mp} 102-104{ }^{\circ} \mathrm{C}$ (see discussion); $R_{\mathrm{f}} 0.51$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3061$ (w), 2966 (s), 2930 (s), 2876 (w), 1446 (w) and $1349(\mathrm{~s}) ; \delta_{\mathrm{H}} 7.85-7.31(10 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{Ph}), 5.01(1 \mathrm{H}, \mathrm{d}$,
$J 5.0,2-\mathrm{H}), 4.13(1 \mathrm{H}$, ddd, $J 6.2,6.2$ and $5.0,3-\mathrm{H}), 3.94(1 \mathrm{H}$, dddd, $J 9.8,6.2,6.2$ and $4.4,5-\mathrm{H}), 2.30(1 \mathrm{H}, \mathrm{ddq}, J 13.3,9.8$ and 7.5, $1^{\prime}-\mathrm{H}_{a}$ ), 2.14-2.07 ( $2 \mathrm{H}, \mathrm{m}, 4-\mathrm{CH}_{2}$ ), $1.75(1 \mathrm{H}, \mathrm{dqd}, J 13.3,7.3$ and $4.4,1^{\prime}-\mathrm{H}_{\beta}$ ) and $0.98\left(3 \mathrm{H}, \mathrm{t}, J 7.3,2^{\prime}-\mathrm{Me}\right)$; $\delta_{\mathrm{C}} 140.8,137.8$, (both $\mathrm{Ph}-\mathrm{C}$ ), 132.9, 128.9, 128.5, 127.9, 126.5 (all Ph-CH), 74.8 ( $2-\mathrm{CH}$ ), $62.8(5-\mathrm{CH}), 41.7\left(4-\mathrm{CH}_{2}\right), 28.7\left(1^{\prime}-\mathrm{CH}_{2}\right), 26.0(3-\mathrm{CH})$ and 10.7 ( $2^{\prime}-\mathrm{Me}$ ); $m / z$ (FAB) $442\left(\mathrm{M}^{+}+1,19 \%\right), 410(53), 380$ (46), 252 (63), 184 (66), 133 (100) and 77 (49) [Found: C, 49.2; $\mathrm{H}, 4.6 ; \mathrm{N}, 3.2 . \mathrm{C}_{18} \mathrm{H}_{20} \mathrm{INO}_{2} \mathrm{~S}$ requires C, 49.0; H, 4.6; N, 3.2\%].
(2SR,3RS,4SR,5RS)-4,5-Dimethyl-3-iodo-2-propyl-1-( $p$ tolylsulfonyl)pyrrolidine 38. ( $E$ )-Homoallylic tosylamide 27b $(0.10 \mathrm{~g}, 0.34 \mathrm{mmol})$ was cyclised under the general conditions (no base), and the reaction was complete after 5 min . The reaction was stirred at ambient temperature for a further 30 min to effect possible isomerisation. The crude product was purified by cc (dichloromethane) followed by recrystallisation (hexaneethyl acetate $9: 1$ ) to afford the title compound $\mathbf{3 8}(0.12 \mathrm{~g}, 82 \%)$ as a colourless solid, $\mathrm{mp} 103-104^{\circ} \mathrm{C}$ : the spectral and analytical data were identical to those obtained for the same substrate when cyclised in the presence of potassium carbonate.
(2SR,3RS,4SR,5SR)-4,5-Dimethyl-3-iodo-2-propyl-1-( $p$ -
tolylsulfonyl)pyrrolidine 39. ( $E$ )-Homoallylic tosylamide 28b $(0.10 \mathrm{~g}, 0.34 \mathrm{mmol})$ was cyclised under the general conditions (no base), and the reaction was complete after 4 min . The reaction was stirred at ambient temperature for a further 30 min to effect possible isomerisation. The crude product was purified by cc (dichloromethane) followed by recrystallisation (hexaneethyl acetate $9: 1)$ to yield the title compound $39(0.11 \mathrm{~g}, 78 \%)$ as a solid, mp 110-111 ${ }^{\circ} \mathrm{C}$. The spectral and analytical data were identical to those obtained from the potassium carbonate reaction described previously.

Cyclisation of ( $\boldsymbol{Z}$ )-3-( $\boldsymbol{p}$-tolylsulfonylamino)dec-5-ene 26c. The $(Z)$-homoallylic tosylamide $26 \mathrm{c}(0.05 \mathrm{~g}, 0.16 \mathrm{mmol})$ was reacted under the conditions described in the general procedure (no base) for 1 h . Purification by cc (dichloromethane) afforded an unresolvable mixture of isomers in the ratio $4: 1: 1(0.04 \mathrm{~g}$, $57 \%$ ). The two minor isomers were found (by comparison of the ${ }^{1} \mathrm{H}$ NMR spectrum of the mixture) to be the $(2 S R, 3 R S, 5 S R)$-pyrrolidine $\mathbf{3 6 f}$ and the $(2 R S, 3 S R, 5 S R)$ pyrrolidine $\mathbf{3 7 f}$ obtained from the cyclisation of the corresponding ( $E$ )-precursor $\mathbf{2 4 f}$ under acidic and basic conditions, respectively. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were sufficiently well resolved to assign the peaks for the major isomer: $R_{\mathrm{f}} 0.52$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2948(\mathrm{~s}), 2930(\mathrm{~s}), 2868$ (w), $1475(\mathrm{w})$ and $1340(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 7.69(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H})$, $7.31(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 3.80(1 \mathrm{H}$, ddd, $J 9.8,6.6$ and 3.8 , $3-\mathrm{H}), 3.51-3.44(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.27(1 \mathrm{H}$, ddd, $J 12.5,7.0$ and 6.6 , $2-\mathrm{H}), 2.45$ (3H, s, Ar-Me), 2.12-1.92 (4H, m), 1.45-1.28 (6H, $\mathrm{m}), 0.93(3 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{Me})$ and $0.90(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me}) ; \delta_{\mathrm{C}}(67.8)$ 142.8, 135.6 (both Ar-C), 129.9, 127.5 (both Ar-CH), 63.6, 62.3 (both CH ), 41.4, 33.6, 31.0, 28.4, 22.5 (all CH2), 21.5 ( $\mathrm{Ar}-\mathrm{Me}$ ), 20.9 (3-CH), 14.0 ( $\left.4^{\prime \prime}-\mathrm{Me}\right)$ and 10.2 ( $2^{\prime \prime}-\mathrm{Me}$ ); $m / z$ (FAB), 436 $\left(\mathrm{M}^{+}+1,100 \%\right), 308$ (95), 212 (84), 155 (69), 91 (86), 68 (49) and 56 (52) [Found: $\mathrm{M}^{+}+1,436.0770 . \mathrm{C}_{17} \mathrm{H}_{27} \mathrm{INO}_{2} \mathrm{~S}$ requires $M$, 436.0807], which was tentatively assigned the all-cis structure 44.
(2SR,3RS,3aSR,7aRS)-2-Butyl-3-iodo-1-(p-tolylsulfonyl)octahydroindole 40. ( $E$ )-Tosylamide $31(20 \mathrm{mg}, 0.06 \mathrm{mmol})$ was cyclised under the conditions described in the general procedure (no base), and the reaction was complete after 15 min . The crude product was purified by cc (hexane-ether $4: 1$ ) to afford the title compound $\mathbf{4 0}(20 \mathrm{mg}, 75 \%)$ as a yellow oil: $v_{\max } /$ $\mathrm{cm}^{-1} 2952(\mathrm{~s}), 2935(\mathrm{~s}), 2884(\mathrm{~s}), 1342(\mathrm{~s})$ and $1294(\mathrm{~s}) ; \delta_{\mathrm{H}}(400)$ 7.73 ( $2 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{Ar}-\mathrm{H}$ ), 7.33 ( $2 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{Ar}-\mathrm{H}$ ), 3.96 ( 1 H , dd, $J 11.7$ and $8.6,3-\mathrm{H}), 3.92-3.87(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 3.57(1 \mathrm{H}$, dt, $J 11.2$ and $6.8,7 \mathrm{a}-\mathrm{H}), 2.45$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 2.02-0.98 ( $15 \mathrm{H}, \mathrm{m}$ )
and $0.95(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Me}) ; \delta_{\mathrm{C}}(100)$ 143.6, 134.8 (both Ar-C), 129.8, 127.4 (both Ar-CH), 69.6, 59.7 (both CH), 47.3 (3a-CH), 33.9, 30.8 (both $\mathrm{CH}_{2}$ ), 27.2 (CHI), 26.9, 24.6, 24.4, 22.8 (all $\left.\mathrm{CH}_{2}\right), 21.6(\mathrm{Ar}-\mathrm{Me}), 20.1\left(\mathrm{CH}_{2}\right)$ and $14.0(\mathrm{Me}) ; m / z(\mathrm{ES}) 462$ $\left(\mathrm{M}^{+}+1,5 \%\right), 336$ (15), 124 (71) and 58 (100) [Found: $\mathrm{M}^{+}+1$, 462.0964. $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{INO}_{2} \mathrm{~S}$ requires $M, 462.0967$ ].
(2RS,3RS,3aSR,7aRS)-2-Butyl-3-iodo-1-( $p$-tolylsulfonyl)octahydroindole 41. ( $Z$ )-Tosylamide $32(12 \mathrm{mg}, 0.03 \mathrm{mmol}$ ) was cyclised under the conditions described in the general procedure (no base), and the reaction was complete after 3 h . The crude product was purified by cc (hexane-ether $80: 20$ ) to afford the title compound $\mathbf{4 1}(7 \mathrm{mg}, 43 \%)$ as a yellow oil: $v_{\max } /$ $\mathrm{cm}^{-1} 2950(\mathrm{~s}), 2937(\mathrm{~s}), 2879(\mathrm{~s}), 1350(\mathrm{~s})$ and $1300(\mathrm{~s}) ; \delta_{\mathrm{H}}(400)$ 7.74 (2H, d, $J 8.3, \mathrm{Ar}-\mathrm{H}), 7.29$ ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), 4.45 ( 1 H , dd, $J 12.7$ and $7.8,3-\mathrm{H}), 3.90-3.83(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{and} 7 \mathrm{a}-\mathrm{H}), 2.43$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.27-2.20(1 \mathrm{H}, \mathrm{m}), 2.01-1.87(2 \mathrm{H}, \mathrm{m}), 1.64$ $1.08(12 \mathrm{H}, \mathrm{m})$ and $0.83(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Me}) ; \delta_{\mathrm{C}}(100) 143.4,133.2$ (both Ar-C), 129.5, 127.2 (both Ar-CH), 60.4, 59.7 (both CH), $44.2(3 \mathrm{a}-\mathrm{CH}), 35.8\left(\mathrm{CH}_{2}\right), 30.9(\mathrm{CHI}), 28.8,28.0,25.4,23.8$, $22.7\left(\right.$ all $\left.\mathrm{CH}_{2}\right), 21.4(\mathrm{Ar}-\mathrm{Me}), 19.7\left(\mathrm{CH}_{2}\right)$ and $14.0(\mathrm{Me}) ; \mathrm{m} / \mathrm{z}$ (ES) $462\left(\mathrm{M}^{+}+1,3 \%\right), 124(42)$ and $58(100)$ [Found: $\mathrm{M}^{+}+1$, 462.0964].

Cyclisation of ( $E$ )-2-( $\boldsymbol{N}$-tert-butoxycarbonylamino)oct-4-ene 35a. ( $E$ )-Homoallylic amide $35 \mathrm{a}(0.03 \mathrm{~g}, 0.13 \mathrm{mmol})$ was cyclised under the conditions described in the general procedure (no base) for 30 min . The crude product was purified by cc (methanol-dichloromethane $5: 95$ ) to afford the homoallylic amine $\mathbf{3 5 b}$ ( $0.002 \mathrm{~g}, 15 \%$ ) as a yellow oil (spectral and analytical data were identical to the sample prepared earlier), and the carbamates as a mixture of isomers 45 in the ratio $2: 1(0.03 \mathrm{~g}$, $67 \%$ ) as a viscous yellow oil: spectral and analytical data were identical to those from the product of the same substrate with potassium carbonate as the base.

Cyclisation of ( $E$ )-2-acetylaminooct-4-ene 35d. ( $E$ )-Homoallylic amide $35 \mathrm{~d}(0.10 \mathrm{~g}, 0.4 \mathrm{mmol}$ ) in acetonitrile ( 2 ml ) was reacted under the conditions described in the general procedure (no base) for 1 h . The crude product was purified by cc (dichloromethane) to afford a mixture of isomers 46 d in the ratio $2: 1(0.07 \mathrm{~g}, 67 \%)$ as a yellow oil. The isomer ratio was determined by measurement of acetyl-methyl peaks from the ${ }^{1} \mathrm{H}$ NMR spectrum. Data for the mixture: $R_{\mathrm{f}} 0.49$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2948$ (s), 2929 (s), 1691, 1419 (s) and 1341 (s); $\delta_{\mathrm{H}}(250) 4.59-4.41(4 \mathrm{H}, \mathrm{m}$, both isomers), 4.25-4.12 ( $2 \mathrm{H}, \mathrm{m}$, both isomers), $2.67(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}$, minor isomer), $2.64(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}$, major isomer), 2.04-1.40 ( $12 \mathrm{H}, \mathrm{m}$, both isomers), $1.68(3 \mathrm{H}, \mathrm{d}$, $J 6.6$, Me, minor isomer), $1.62(3 \mathrm{H}, \mathrm{d}, J 6.6$, Me, major isomer), $1.02(3 \mathrm{H}, \mathrm{t}, J 7.3$, Me, minor isomer) and $1.01(3 \mathrm{H}, \mathrm{t}, J 7.3$, major isomer); $\delta_{\mathrm{C}}$ ( 100 , major isomer only) 83.2, 48.1, 37.0, 34.6, 32.7, 23.0, 21.8, 19.6 and 13.4.

Cyclisation of ( $E$ )-2-methoxycarbonylaminooct-4-ene 35 e. ( $E$ )-Homoallylic sulfonamide $35 \mathrm{e}(0.05 \mathrm{~g}, 0.23 \mathrm{mmol})$ was reacted under the conditions described in the general procedure (no base) for 1 h . The crude product was purified by cc (dichloromethane) to afford a mixture of isomers in the ratio $2: 1(0.03 \mathrm{~g}, 42 \%)$ as a yellow oil. The ratio of isomers and 46 e was determined by measurement of the OME resonances in the ${ }^{1} \mathrm{H}$ NMR spectrum. The two isomers were identical to those obtained from the same substrate when potassium carbonate was used as the base.

## Deiodination of iodopyrrolidines: general procedure

To a stirred solution of a 3-iodopyrrolidine ( $1 \mathrm{mmol}, 1$ eq.) in dry, degassed toluene ( 50 ml ) under nitrogen was added tributyltin hydride ( $2.91 \mathrm{~g}, 10 \mathrm{mmol}$ ) and the reaction mixture heated at reflux until complete reduction according to TLC analysis
(typically $2-5 \mathrm{~h}$ ). The reaction mixture was cooled and the toluene evaporated. The remaining crude product was dissolved in acetonitrile ( 10 ml ) and extracted with hexane ( $3 \times 10 \mathrm{ml}$ ) to remove any organotin residues. The acetonitrile was evaporated to yield the required product which was purified by column chromatography.

## (2SR,5SR)-2-Butyl-5-methyl-1-( $p$-tolylsulfonyl)pyrrolidine

49a. Iodopyrrolidine $36 \mathbf{d}(0.05 \mathrm{~g}, 0.12 \mathrm{mmol}$ ) was deiodinated as described above to give the title compound 49a ( $0.03 \mathrm{~g}, 88 \%$ ) as a colourless, viscous oil: $R_{\mathrm{f}} 0.32$ (dichloromethane); $v_{\text {max }}$ / $\mathrm{cm}^{-1} 2959(\mathrm{~s}), 2938(\mathrm{~s}), 2875(\mathrm{w})$ and $1355(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.73$ ( 2 H , d, $J 8.3$, Ar-H), 7.25 ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), 4.02 ( $1 \mathrm{H}, \mathrm{dqd}, J 6.5$, 6.3 and $1.2,5-\mathrm{H}), 3.84-3.77(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.41$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 2.08-1.09 ( $10 \mathrm{H}, \mathrm{m}, 1-, 2-, 3^{\prime}-, 3^{\prime}-$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 1.19(3 \mathrm{H}, \mathrm{d}$, $\left.J 6.3,1^{\prime \prime}-\mathrm{Me}\right)$ and $0.85\left(3 \mathrm{H}, \mathrm{t}, J 7.2,4^{\prime}-\mathrm{Me}\right)$; $\delta_{\mathrm{C}}$ (100) 142.4, 139.3 (both Ar-C), 129.3, 126.9 (both Ar-CH), 60.7 (5-CH), 56.3 (2-CH), $33.9\left(1^{\prime}-\mathrm{CH}_{2}\right), 31.5,28.5,27.7,22.5\left(\mathrm{all} \mathrm{CH}_{2}\right), 21.4$ (Ar-Me), 21.1 ( $1^{\prime \prime}-\mathrm{Me}$ ) and 14.0 ( $4^{\prime}-\mathrm{Me}$ ); $m / z$ (ES) 296 ( $\mathrm{M}^{+}+1$, $100 \%$ ), 125 (3), 102 (4) and 60 (7) [Found: $\mathrm{M}^{+}+1,296.1680$. $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{NO}_{2} \mathrm{~S}$ requires $M, 296.1684$ ].
(2SR,5SR)-2-Butyl-5-ethyl-1-( $p$-tolylsulfonyl)pyrrolidine 49b. 3-Iodopyrrolidine $\mathbf{3 6 f}(0.05 \mathrm{~g}, 0.12 \mathrm{mmol})$ was deiodinated as described above to give the title compound $\mathbf{4 9 b}(0.03 \mathrm{~g}, 84 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.32$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}$ $2959(\mathrm{~s}), 2940(\mathrm{~s}), 2877(\mathrm{~s})$ and $1355(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.71(2 \mathrm{H}, \mathrm{d}$, $J 8.2$, Ar-H), 7.23 ( $2 \mathrm{H}, \mathrm{d} J 8.2$, Ar-H), 3.84-3.70 ( $2 \mathrm{H}, \mathrm{m}, 2$ - and $5-\mathrm{H}), 2.40(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.03-1.05$ ( $12 \mathrm{H}, \mathrm{m}, 1-, 2-, 3^{\prime}, 3^{\prime}$ - and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 0.93(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$ and $0.90(3 \mathrm{H}, \mathrm{t}, J 6.9, \mathrm{Me})$; $\delta_{\mathrm{C}}(100)$ 142.1, 139.8 (both Ar-C), 129.0, 126.6 (both Ar-CH), 62.4 (5-CH), 56.3 (2-CH), 33.8, 31.9, 28.3, 27.4, 26.4, 22.3 (all $\mathrm{CH}_{2}$ ), 21.3 ( $\mathrm{Ar}-\mathrm{Me}$ ), 13.9 ( $4^{\prime}-\mathrm{Me}$ ) and 10.7 ( $\left.2^{\prime \prime}-\mathrm{Me}\right) ; ~ m / z(\mathrm{ES})$ $310\left(\mathrm{M}^{+}+1,100 \%\right), 281$ (35) and 91 (20) [Found: $\mathrm{M}^{+}+1$, 310.1837. $\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{NO}_{2} \mathrm{~S}$ requires $\left.M, 310.1841\right]$.

## (2SR,5SR)-5-Ethyl-2-pentyl-1-( $p$-tolylsulfonyl)pyrrolidine

49c. 3-Iodopyrrolidine $\mathbf{3 6 h}(0.05 \mathrm{~g}, 0.11 \mathrm{mmol})$ was deiodinated as described above to give the title compound $49 \mathrm{c}(0.03 \mathrm{~g}$, $84 \%$ ) as a colourless, viscous oil: $R_{\mathrm{f}} 0.34$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2958(\mathrm{~s}), 2940(\mathrm{~s}), 2875(\mathrm{~s})$ and $1357(\mathrm{~s}) ; \delta_{\mathrm{H}}(250) 7.72$ (2H, d, J 8.2, Ar-H), 7.25 (2H, d, J 8.2, Ar-H), 3.87-3.71 (2H, $\mathrm{m}, 2-\mathrm{and} 5-\mathrm{H}), 2.41$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 2.03-1.02 ( $14 \mathrm{H}, \mathrm{m}, 1^{\prime}-, 1^{\prime \prime}-$,, $2^{\prime}-, 3-, 3^{\prime}-, 4-$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 0.92(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$ and $0.89(3 \mathrm{H}$, t, J 6.9, Me); $\delta_{\mathrm{C}}(100)$ 142.2, 139.7 (both Ar-C), 129.3, 126.8 (both Ar-CH), 62.2 ( $5-\mathrm{CH}$ ), 56.3 (2-CH), 33.8, 31.6, 27.9, 27.4, 26.7, 26.1, $22.5($ all CH 2$), 21.4(\mathrm{Ar}-\mathrm{Me}), 14.0\left(5^{\prime}-\mathrm{Me}\right)$ and 10.64 (2"-Me); m/z (FAB) 324 ( ${ }^{+}+1,10 \%$ ), 291 (19), 235 (44), 179 (100) and 91 (9) [Found: $\mathrm{M}^{+}+1$, $324.1985 . \mathrm{C}_{18} \mathrm{H}_{30} \mathrm{NO}_{2} \mathrm{~S}$ requires $M 324.1997]$.

## (2SR,5SR)-5-Butyl-2-pentyl-1-( $p$-tolylsulfonyl)pyrrolidine

49d. Iodopyrrolidine $\mathbf{3 6 i}(0.07 \mathrm{~g}, 0.15 \mathrm{mmol})$ was deiodinated as described above to yield the title compound $\mathbf{4 9 d}(0.04 \mathrm{~g}, 82 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.48$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1}$ 2959 (s), 2936 (s), 2872 (w), 1464 (w) and 1355 (s); $\delta_{\mathrm{H}}$ (250) 7.72 ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), 7.26 ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), $3.81(2 \mathrm{H}, \mathrm{m}$, 2 - and $5-\mathrm{H}$ ), 2.42 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 1.92-1.13 ( $18 \mathrm{H}, 1^{\prime}-$-, 1"-, $2^{\prime}-$, $2^{\prime \prime}-, 3^{\prime}-, 3^{\prime \prime}-, 4-$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 0.88(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$ and 0.85 (3H, t, J 7.1, Me); $\delta_{\mathrm{C}}$ (100) 142.4, 140.1 (both Ar-C), 129.3, 126.9 (both Ar-CH), 60.9 ( $2-\mathrm{CH}$ and $5-\mathrm{CH}$ ), 33.8, 33.6, 31.6, 31.1, 28.6, 28.0, 26.1, 22.6 (all $\mathrm{CH}_{2}$ ), 21.4 (Ar-Me) and 14.0 ( $4^{\prime \prime}$ - and $\left.5^{\prime}-\mathrm{Me}\right) ; m / z(\mathrm{ES}) 352\left(\mathrm{M}^{+}+1,100 \%\right), 296$ (13), 291 (18), 235 (20), 102 (14) and 83 (17) [Found: $\mathrm{M}^{+}+1,352.2310$. $\mathrm{C}_{20} \mathrm{H}_{34} \mathrm{NO}_{2} \mathrm{~S}$ requires 352.2312].
(2RS,5SR)-5-Ethyl-2-phenyl-1-( $p$-tolylsulfonyl)pyrrolidine
49e. Iodopyrrolidine $\mathbf{3 6 j}(0.20 \mathrm{~g}, 0.60 \mathrm{mmol})$ was deiodinated as described above to yield the title compound $49 \mathrm{e}(0.16 \mathrm{~g}, 82 \%)$ as a viscous, colourless oil; $R_{\mathrm{f}} 0.34$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}$

2959 (s), 2945 (s), 2875 (w), 1461 (w) and 1349 (s); $\delta_{\mathrm{H}}(250) 7.37$ ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), $7.17-7.00(7 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 5.98(1 \mathrm{H}, \mathrm{dd}$, $J 7.9$ and $1.4,2-\mathrm{H}), 4.07-4.04(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.43-2.39(1 \mathrm{H}, \mathrm{m}$, $3-\mathrm{H}_{a}$ ), 2.35 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 2.25-2.09 ( $2 \mathrm{H}, \mathrm{m}, 4-\mathrm{and} 1^{\prime}-\mathrm{H}_{a}$ ), $1.82-1.67\left(2 \mathrm{H}, \mathrm{m}, 3-\right.$ and $\left.4-\mathrm{H}_{\beta}\right), 1.51$ ( $1 \mathrm{H}, \mathrm{dqd}, J 13.4,7.4$ and $2.9,1^{\prime}-\mathrm{H}_{\beta}$ ) and $0.90\left(3 \mathrm{H}, \mathrm{t}, J 7.4,2^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{C}}(100) 142.5,142.2$, 138.9 (all Ar-C), 128.8, 127.9, 126.8, 126.8, 126.5 (all Ar-CH), $63.9(2-\mathrm{CH}), 63.0(5-\mathrm{CH}), 33.3\left(1^{\prime}-\mathrm{CH}_{2}\right), 27.4\left(3-\mathrm{CH}_{2}\right), 27.2$ $\left(4-\mathrm{CH}_{2}\right), 21.3(\mathrm{Ar}-\mathrm{Me})$ and $10.6\left(1^{\prime}-\mathrm{Me}\right) ; m / z(\mathrm{ES}) 330\left(\mathrm{M}^{+}+1\right.$, $100 \%), 176$ (27), 174 (38), 102 (8), 85 (8) and 60 (19) [Found: $\mathrm{M}^{+}+1,330.1529 \mathrm{C}_{19} \mathrm{H}_{24} \mathrm{NO}_{2} \mathrm{~S}$ requires 330.1532].

## (2RS,5SR)-2-Butyl-5-methyl-1-(p-tolylsulfonyl)pyrrolidine

50a. Iodopyrrolidine $\mathbf{3 7 d}$ ( $0.05 \mathrm{~g}, 0.12 \mathrm{mmol}$ ) was deiodinated under the conditions described in the general procedure to afford the title compound $\mathbf{5 0 a}(0.03 \mathrm{~g}, 86 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.32$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2960$ (s), 2942 (s), 2875 (w) and 1357 (s); $\delta_{\mathrm{H}}(400) 7.71$ (2H, d, J 8.1, Ar-H), $7.29(2 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{Ar}-\mathrm{H}), 3.69-3.63(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.58-3.53$ $(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.42(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 1.89-1.27\left(10 \mathrm{H}, \mathrm{m}, 1^{\prime}-\right.$, $2^{\prime}-3$-, $3^{\prime}-$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 1.32\left(3 \mathrm{H}, \mathrm{d}, J 6.3,1^{\prime \prime}-\mathrm{Me}\right)$ and $0.92(3 \mathrm{H}$, $\mathrm{t}, J 7.2,4^{\prime}-\mathrm{Me}$ ); $\delta_{\mathrm{C}}(100) 143.4,143.2$ (both $\left.\mathrm{Ar}-\mathrm{C}\right), 129.5,127.4$ (both Ar-CH), 61.9 (5-CH), 57.2 (2-CH), 36.9, 32.1, 29.5, 28.4 (all $\left.\mathrm{CH}_{2}\right), 23.5\left(1^{\prime \prime}-\mathrm{Me}\right), 22.6\left(\mathrm{CH}_{2}\right), 21.4(\mathrm{Ar}-\mathrm{Me})$ and 14.0 ( $4^{\prime}-\mathrm{Me}$ ); $m / z$ (ES) $296\left(\mathrm{M}^{+}+1,100 \%\right), 276$ (6) and 221 (5) [Found: $\left.\mathrm{M}^{+}+1,296.1677\right]$.
(2RS,5SR)-2-Butyl-5-ethyl-1-(p-tolylsulfonyl)pyrrolidine 50b. Iodopyrrolidine $37 \mathrm{f}(0.05 \mathrm{~g}, 0.12 \mathrm{mmol})$ was deiodinated under the conditions described in the general procedure to afford the title compound $\mathbf{5 0 b}(0.03 \mathrm{~g}, 81 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}}$ 0.35 (dichloromethane); $v_{\max } / \mathrm{cm}^{-1} 2960$ (s), 2943 (s), 2874 (w) and $1355(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 7.72(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 7.32(2 \mathrm{H}, \mathrm{d}$, $J 8.2$, Ar-H), $3.55-3.45(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{and} 5-\mathrm{H}), 2.42(3 \mathrm{H}, \mathrm{s}$, Ar-Me), 1.91-1.12 ( $12 \mathrm{H}, \mathrm{m}, 1^{\prime}-$, $1^{\prime \prime}-, 2-, 3-$ - $3^{\prime}-$ and $4^{\prime}-\mathrm{CH}_{2}$ ), $0.92(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$ and $0.87(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Me}) ; \delta_{\mathrm{C}}(67.8)$ 142.8, 135.7 (both Ar-C), 129.5, 127.3 (both Ar-CH), 62.7, 61.4 (both CH), 37.2, 32.0, 29.8, 29.6, 25.8, 22.4 (all $\mathrm{CH}_{2}$ ), 21.5 (Ar-Me), $13.8\left(4^{\prime}-\mathrm{Me}\right)$ and $10.5\left(2^{\prime \prime}-\mathrm{Me}\right) ; m / z(\mathrm{ES}) 310\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ), 281 (26), 125 (43), and 91 (23) [Found: $\mathrm{M}^{+}+1$, 310.1836].

## (2RS,5SR)-5-Ethyl-2-pentyl-1-( $p$-tolylsulfonyl)pyrrolidine

50c. Iodopyrrolidine $\mathbf{3 7 h}(0.06 \mathrm{~g}, 0.11 \mathrm{mmol})$ was deiodinated by the general procedure to afford the pyrrolidine $50 \mathrm{c}(0.03 \mathrm{~g}$, $82 \%$ ) as a colourless, viscous oil: $R_{\mathrm{f}} 0.34$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2959(\mathrm{~s}), 2943(\mathrm{~s}), 2872(\mathrm{w})$ and $1356(\mathrm{~s}) ; \delta_{\mathrm{H}}(400) 7.71$ ( $2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 7.30(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar}-\mathrm{H}), 3.56-3.47(2 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{and} 5-\mathrm{H}), 2.41$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 1.91-1.12 ( $14 \mathrm{H}, \mathrm{m}, 1^{\prime}-, 1^{\prime \prime}-$, $2^{\prime}-, 3-, 3^{\prime}-, 4-$ and $\left.4^{\prime}-\mathrm{CH}_{2}\right), 0.92(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me})$ and $0.88(3 \mathrm{H}$, $\mathrm{t}, J 7.2$, Me); $\delta_{\mathrm{C}}(67.8) 142.9,135.4$ (both Ar-C), 129.5, 127.4 (both $\mathrm{Ar}-\mathrm{CH}$ ), 62.9, 61.7 (both CH ), 37.1, 31.7, 29.8, 29.5, 29.1, 25.9, 22.6 (all $\mathrm{CH}_{2}$ ), 21.5 ( $\mathrm{Ar}-\mathrm{Me}$ ), $14.0\left(5^{\prime}-\mathrm{Me}\right)$ and 10.5 ( $\left.2^{\prime \prime}-\mathrm{Me}\right) ; m / z(\mathrm{FAB}) 324\left(\mathrm{M}^{+}+1,7 \%\right), 297$ (22), 235 (43), 179 (100) and 91 (10) [Found: $\left.\mathrm{M}^{+}+1,324.1996\right]$.

## (2RS,5SR)-5-Butyl-2-pentyl-1-(p-tolylsulfonyl)pyrrolidine

50d. Iodopyrrolidine $\mathbf{3 7 i}(0.05 \mathrm{~g}, 0.11 \mathrm{mmol})$ was deiodinated as described in the general procedure to yield the title compound $50 \mathrm{~d}(0.03 \mathrm{~g}, 83 \%)$ as a colourless, viscous oil: $R_{\mathrm{f}} 0.50$ (dichloromethane); $v_{\text {max }} / \mathrm{cm}^{-1} 2959$ (s), 2931 (s), 2861 (w), 1461 (w) and 1335 (s); $\delta_{\mathrm{H}}(250) 7.71(2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}), 7.29$ ( $2 \mathrm{H}, \mathrm{d}$, J 8.3, Ar-H), 3.56-3.48 ( $2 \mathrm{H}, \mathrm{m}, 2-\mathrm{and} 5-\mathrm{H}$ ), 1.88-1.09 ( 18 H , $\mathrm{m}, 1^{\prime}-, 1^{\prime \prime}-, 2^{\prime}-, 2^{\prime \prime}-3-33^{\prime}-, 3^{\prime \prime}-, 4-\mathrm{and} 4^{\prime}-\mathrm{CH}_{2}$ ), $0.94(3 \mathrm{H}, \mathrm{t}$, $J 7.2, \mathrm{Me})$ and $0.87(3 \mathrm{H}, \mathrm{t}, J 6.7, \mathrm{Me}) ; \delta_{\mathrm{C}}(100) 142.8,135.2$ (both Ar-C), 129.5, 127.6 (both Ar-CH), 61.7 (2- and 5-CH), 37.2, 36.9, 31.7, 29.8, 29.6, 28.5, 25.9, 22.6, $22.6\left(\right.$ all CH$\left._{2}\right), 21.5$ (Ar-Me) and 14.1 ( $4^{\prime \prime}-$ and $\left.5^{\prime}-\mathrm{Me}\right) ; m / z$ (ES) $352\left(\mathrm{M}^{+}+1\right.$, $100 \%$ ), 291 (17), 235 (20), 102 (14) and 83 (15). These data are consistent with those in the literature. ${ }^{41}$

## (2SR,5SR)-5-Ethyl-2-phenyl-1-( $p$-tolylsulfonyl)pyrrolidine

50e. Iodopyrrolidine $37 \mathbf{j}(0.10 \mathrm{~g}, 0.30 \mathrm{mmol})$ was deiodinated as described above to yield the title compound $\mathbf{5 0 e}(0.08 \mathrm{~g}, 85 \%$ ) as a colourless, viscous oil: $R_{\mathrm{f}} 0.36$ (dichloromethane); $v_{\max } / \mathrm{cm}^{-1}$ 2960 (s), 2942 (s), 2875 (w), 1461 (w) and 1349 (s); $\delta_{\mathrm{H}}$ (250) 7.72 ( $2 \mathrm{H}, \mathrm{d}, J 8.3, \mathrm{Ar}-\mathrm{H}$ ), $7.41-7.21$ ( $7 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ), 5.98 ( 1 H , dd, $J 6.9$ and $J 6.9,2-\mathrm{H}), 3.72(1 \mathrm{H}$, dddd, $J 11.4,7.4,6.9$ and 4.7, $5-\mathrm{H}), 2.44(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}), 2.09(1 \mathrm{H}, \mathrm{dqd}, J$ 13.1, 7.3 and 4.7, $\left.1^{\prime}-\mathrm{H}_{\omega}\right), 1.97-1.82\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{CH}_{2}\right), 1.66-1.53\left(3 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{H}_{\beta}\right.$ and $4-\mathrm{CH}_{2}$ ) and $0.99\left(3 \mathrm{H}, \mathrm{t}, J 7.4,2^{\prime}-\mathrm{Me}\right)$; $\delta_{\mathrm{C}}$ (100) 143.2, 142.8, 135.1 (all Ar-C), 129.5, 128.2, 127.6 (all Ar-CH), 126.9 (Ar-CH), 128.2 (Ar-CH), 64.6 (2-CH), $63.8(5-\mathrm{CH}), 34.2$ $\left(1^{\prime}-\mathrm{CH}_{2}\right), 29.5\left(3-\mathrm{CH}_{2}\right), 29.3\left(4-\mathrm{CH}_{2}\right), 21.4(\mathrm{Ar}-\mathrm{Me})$ and 10.9 (2'-Me); $m / z$ (ES) 330 ( $\mathrm{M}^{+}+1,100 \%$ ), 176 (23), 174 (63), 83 (7) and 60 (17) [Found: $\mathrm{M}^{+}+1,330.1528 . \mathrm{C}_{19} \mathrm{H}_{24} \mathrm{NO}_{2} \mathrm{~S}$ requires 330.1532].

## Elimination of hydrogen iodide from iodopyrrolidines: general procedure

A solution of the 3-iodopyrrolidine ( $1.0 \mathrm{mmol}, 1 \mathrm{eq}$.) and DBU $(0.18 \mathrm{~g}, 1.2 \mathrm{mmol}, 1.2 \mathrm{eq}$.) in toluene ( 10 ml ) was heated at reflux until the reaction was complete according to the TLC analysis (typically $2-5 \mathrm{~h}$ ). The reaction mixture was cooled, quenched with aqueous 2 M hydrochloric acid ( 5 ml ) and the resulting two layers separated. The aqueous layer was extracted with ether $(3 \times 10 \mathrm{ml})$ and the combined organic solutions dried and the solvent evaporated to yield the product.
(2RS,5SR)-5-Ethyl-2-phenyl-1-(p-tolylsulfonyl)-3-pyrroline
51. Iodopyrrolidine $36 \mathbf{j}(0.05 \mathrm{~g}, 0.11 \mathrm{mmol})$ was reacted as described above to give the 3-pyrroline $51(0.03 \mathrm{~g}, 78 \%)$ as a viscous, colourless oil: $R_{\mathrm{f}} 0.48$ (dichloromethane); $\delta_{\mathrm{H}}$ (270) 7.19-6.83 (9H, m, Ar-H), $5.69(1 \mathrm{H}$, ddd, J 6.3, 1.9 and 1.6, $3-\mathrm{H}), 5.58(1 \mathrm{H}$, ddd, $J 6.3,2.0$ and $1.8,4-\mathrm{H}), 5.50(1 \mathrm{H}$, ddd, $J 5.3,2.0$ and $1.9,2-\mathrm{H}), 4.69-4.57(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.23(3 \mathrm{H}, \mathrm{s}$, Ar-Me), 2.14-1.85 ( $2 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{CH}_{2}$ ) and $0.88(3 \mathrm{H}, \mathrm{t}, J 7.2$, $\left.2^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{C}}$ (67.8) 141.8, 138.1, 137.7 (all Ar-C), 129.9 (Ar-CH), 129.0 (3-CH), 128.9 (4-CH), 128.7, 127.9, 127.8, 126.4 (all $\mathrm{Ar}-\mathrm{CH}), 71.6(2-\mathrm{CH}), 67.7(5-\mathrm{CH}), 28.4\left(1^{\prime}-\mathrm{CH}_{2}\right), 21.2(\mathrm{Ar}-\mathrm{Me})$ and $8.0\left(2^{\prime}-\mathrm{Me}\right) ; m / z(\mathrm{ES}) 328\left(\mathrm{M}^{+}+1,97 \%\right), 245(7), 157$ (8) and 102 (8) [Found: $\mathrm{M}^{+}+1,328.1371 . \mathrm{C}_{19} \mathrm{H}_{22} \mathrm{NO}_{2} \mathrm{~S}$ requires M, 328.1371].
(2SR,5SR)-5-Ethyl-2-phenyl-1-( $p$-tolylsulfonyl)-3-pyrroline
52. Iodopyrrolidine $37 \mathbf{j}(0.05 \mathrm{~g}, 0.11 \mathrm{mmol})$ was reacted as described in the general procedure above to yield the 3-pyrroline 52 ( $0.03 \mathrm{~g}, 72 \%$ ) as a viscous, colourless oil: $R_{\mathrm{f}} 0.48$ (dichloromethane); $\delta_{\mathrm{H}}(270) 7.56$ ( $2 \mathrm{H}, \mathrm{d}, J 8.2$, Ar-H), $7.32-7.15$ ( 7 H , m, Ar-H), $5.68(1 \mathrm{H}$, ddd, $J 6.3,2.0$ and $1.9,3-\mathrm{H}), 5.56(1 \mathrm{H}$, ddd, $J 6.3,2.0$ and $1.9,4-\mathrm{H}), 5.40(1 \mathrm{H}$, ddd, $J 2.0,2.0$ and $2.0,2-\mathrm{H}$ ), 4.46-4.38 ( $1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}$ ), 2.33 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{Me}$ ), 1.99-1.53 ( $2 \mathrm{H}, \mathrm{m}$, $1^{\prime}-\mathrm{CH}_{2}$ ) and $0.90\left(3 \mathrm{H}, \mathrm{t}, J 7.6,2^{\prime}-\mathrm{Me}\right) ; \delta_{\mathrm{C}}(270) 143.8,141.3$, 135.7 (all Ar-C), 130.0, 129.7, 129.1, 128.8, 128.3 (all Ar-CH), 128.0, 127.7 (both CH), 71.1 (2-CH), $69.8(5-\mathrm{CH}), 30.8$ $\left(1^{\prime}-\mathrm{CH}_{2}\right), 21.9$ (Ar-Me), and $10.6\left(2^{\prime}-\mathrm{Me}\right) ; \mathrm{m} / \mathrm{z}$ (ES) 328 $\left(\mathrm{M}^{+}+1,94 \%\right), 245$ (5), 157 (7) and 102 (9) [Found: $\mathrm{M}^{+}+1$, 328.1371].
(2R)-(E)-2-( $\boldsymbol{p}$-Tolylsulfonylamino)non-4-ene $\quad \mathbf{2 4 e} . \quad(2 S, 3 R$, $5 R$ )-Iodopyrrolidine $37 \mathrm{e}(25 \mathrm{mg}, 0.07 \mathrm{mmol})$ in a mixture of ethanol $(2.0 \mathrm{ml})$ and acetic acid ( 0.01 ml ) was treated with zinc dust ( $0.325 \mathrm{mesh}, 10 \mathrm{mg}, 0.07 \mathrm{mmol}$ ). The resulting reaction mixture was stirred at ambient temperature for 6 h then filtered over Kieselguhr and the solid washed with dichloromethane ( $3 \times 2 \mathrm{ml}$ ). The combined organic solutions were washed with 2 M aqueous sodium hydrogen carbonate $(2 \times 5 \mathrm{ml})$, then dried and the solvent evaporated to afford the sulfonamide 24e (18.6 $\mathrm{mg}, 90 \%)$ as a viscous, yellow oil: $[a]_{\mathrm{D}}=+32.1\left(c 0.008, \mathrm{CHCl}_{3}\right)$; other spectral data were identical to those for the racemic sulfonamide 24d (see above).

## X-Ray crystallography for $\mathbf{3 7 f}, \mathbf{3 6 j}$ and $\mathbf{3 7 k} \dagger$

Single crystals were obtained by vapour diffusion as described in the text; X-ray data were collected on a CAD4 diffractometer using monochromatized Mo-K $\alpha$ radiation ( $\lambda=0.71073 \AA$ ).
(2RS,3SR,5SR)-2-Butyl-5-ethyl-3-iodo-1-p-tolylsulfonylpyrrolidine 37f. $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{INO}_{2} \mathrm{~S}=435.35$. Orthorhombic, $P b a 2_{1}$ (No. 29), $\quad a=15.481(2), \quad b=14.469(3), \quad c=16.866(2) ~ \AA$, $V=3778(2) \AA^{3}, Z=8.00, D_{\mathrm{c}}=1.531 \mathrm{~g} \mathrm{~cm}^{-3}, F(000)=1760$. Crystal size $0.26 \times 0.18 \times 0.15 \mathrm{~mm}$. 15269 reflections were measured over the range $1.93<\theta<24.99^{\circ}$ with $-1 \leqslant h \leqslant 18$, $-15 \leqslant k \leqslant 16,-18 \leqslant l \leqslant 15$ giving 5378 independent reflections $[R($ int $)=0.0761]$.

Full matrix least-squares refinement with all non-hydrogen atoms anisotropic and hydrogens in calculated positions. The maximum peak in the final $\Delta F$ map was $1.834 \AA^{-3}$ and the minimum was $-0.801 \AA^{-3}$. Final cycles of refinement gave $R=0.0588$ and $R_{\mathrm{w}}=0.1252$.

## (2RS,3SR,5RS)-5-Ethyl-3-iodo-2-phenyl-1-p-tolylsulfonyl-

 pyrrolidine $\mathbf{3 6 j} . \mathrm{C}_{19} \mathrm{H}_{22} \mathrm{INO} \mathrm{N}_{2} \mathrm{~S}=455.34$. Monoclinic, $P 2_{1} / n$, $a=7.648, b=13.629, c=18.213 \AA, \beta=93.73(2)^{\circ}, \quad V=1894.5$ $\AA^{3}, Z=4.00, D_{\mathrm{c}}=1.596 \mathrm{~g} \mathrm{~cm}^{-3}, F(000)=912$. Crystal size $0.24 \times 0.22 \times 0.18 \mathrm{~mm} .7350$ reflections were measured over the range $1.87<\theta<25.08^{\circ}$ with $-8 \leqslant h \leqslant 8,-14 \leqslant k \leqslant 15$, $-21 \leqslant l \leqslant 21$ giving 2501 independent reflections $[R(\mathrm{int})=$ 0.0507].Full matrix least-squares refinement with all non-hydrogen atoms anisotropic and hydrogens in calculated positions. The maximum peak in the final $\Delta F$ map was $2.851 \AA^{-3}$. Final cycles of refinement gave $R=0.0458$ and $R_{\mathrm{w}}=0.1082$.
(2RS,3SR,5SR)-5-Ethyl-3-iodo-2-phenyl-1-phenylsulfonylpyrrolidine $37 \mathrm{k}(\mathbf{i}) . \mathrm{C}_{18} \mathrm{H}_{20} \mathrm{INO}_{2} \mathrm{~S}=441.31$. Monoclinic, $P 2_{1} / c$, $a=12.577(4), b=12.900(4), c=13.006(6) \AA, \beta=119.06(2)^{\circ}$, $V=1844.5(12) \AA^{3}, Z=4.00, D_{\mathrm{c}}=1.589 \mathrm{~g} \mathrm{~cm}^{-3}, F(000)=880$. Crystal size $0.50 \times 0.50 \times 0.20 \mathrm{~mm} .3696$ reflections were measured over the range $1.85<\theta<24.95$ with $0 \leqslant h \leqslant 14$, $0 \leqslant k \leqslant 15,-15 \leqslant l \leqslant 13$. A semi empirical absorption was applied (transmission factors 0.5123-0.7075) giving 3230 independent reflections $[R(\mathrm{int})=0.0643]$.
Full matrix least-squares refinement with all non-hydrogen atoms anisotropic and hydrogens in calculated positions. The maximum peak in the final $\Delta \mathrm{F}$ map was $0.570 \AA^{-3}$ and the minimum was $-0.565 \AA^{-3}$. Final cycles of refinement gave $R=0.0377$ and $R_{\mathrm{w}}=0.0992$.
(2RS,3SR,5SR)-5-Ethyl-3-iodo-2-phenyl-1-phenylsulfonylpyrrolidine $37 \mathrm{k}(\mathrm{ii}) . \mathrm{C}_{18} \mathrm{H}_{20} \mathrm{INO}_{2} \mathrm{~S}=441.31$. Monoclinic, $P 2_{1} / n$, $a=14.617$ (3), $b=14.417(3), c=8.616(2) \AA, \beta=94.63(3)^{\circ}, V=$ 1809.8(6) $\AA^{3}, Z=4.00, D_{\mathrm{c}}=1.620 \mathrm{~g} \mathrm{~cm}^{-3}, F(000)=880$. Crystal size $0.60 \times 0.40 \times 0.20 \mathrm{~mm} .2851$ reflections were measured over the range $1.99<\theta<22.95^{\circ}$ with $0 \leqslant h \leqslant 16,0 \leqslant k \leqslant 15$, $-9 \leqslant l \leqslant 9$. A semi-empirical absorption was applied (transmission factors 0.6665-0.9724) giving 2501 independent reflections $[R(\mathrm{int})=0.0507]$.
Full matrix least-squares refinement with all non-hydrogen atoms anisotropic and hydrogens in calculated positions. The maximum peak in the final $\Delta F$ map was $0.767 \AA^{-3}$ and the minimum was $-0.722 \AA^{-3}$. Final cycles of refinement gave $R=0.0410$ and $R_{\mathrm{w}}=0.1041$.

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